X¹MT_EX (Versions 5.00) for Typesetting Chemical Structural Formulas: I. Portable-Document-Format-(PDF-)Compatible Mode Supported by the xymtx-pdf Package and the chmst-pdf Package, as Well as II. Coloring Bonds Supported by the bondcolor.

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This document has been typeset by the PDF-compatible mode and the resulting dvi file has been converted into a PDF file by using the dvipdfmx converter.

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Introduction

1.1 History of the XIMT_EX System

The history of the $\hat{X}^{1}MT_{E}X$ system is summarized in Table 1.1. The boldfaced versions have provided substantial improvements for drawing chemical structural formulas. The present version (version 5.00) contains two main improvements, i.e., the development of the PDF mode and the development of bond coloring.

Recent books on $\mathbb{IAT}_{EX} 2_{\varepsilon}$ have referred to the $\hat{X}^{2}MT_{EX}$ system, e.g., pages 520–540 of [14] and pages 551–598 of Vol. II of [15].

1.2 Backgrounds and Motivations of $\hat{X}MT_EX$ Version 5.00

1.2.1 PDF Printing

The previous version (version 4.06) of the $\hat{X}^{1}MT_{E}X$ system has supported the $T_{E}X/I^{4}T_{E}X$ mode based on the IATEX picture environment as well as the PostScript mode based on the PSTricks package linked with the IATEX picture environment. The latter PostScript mode requires after-processing by the dvips converter (dvi files \rightarrow ps files), where resulting PostScript files can be browsed by using the Ghostscript system (coupled with Ghostview). The PostScript files are further transformed into PDF files by using an appropriate converter (e.g., Adobe Distiller). It follows that the PDF printing of $\hat{X}^{1}MT_{E}X$ structural formulas is available via such a route as

 $< \mathrm{tex} \ \mathrm{with} \ \hat{X^{1}}\!\!MT_{E}\!X \ \mathrm{codes} > \xrightarrow{T_{E}X/L^{A}T_{E}X} < \mathrm{dvi} > \xrightarrow{\mathrm{dvips}} < \mathrm{ps} > \xrightarrow{\mathrm{Distiller}} < \mathrm{PDF} >.$

On the other hand, structural formulas drawn by the $\hat{X}^{1}MT_{E}X$ system can also be transformed into EPS (encapsulated PostScript) files by using the Ghostscript utilities so as to be incorporated into PDF files. This means that we are alternatively able to use the dvipdfm(x) converter in order to convert $I^{A}T_{E}X$ document files with $\hat{X}^{1}MT_{E}X$ structural formulas (EPS files) into PDF files, i.e.,

 $< \text{tex with XIMTEX EPS files} > \stackrel{\text{TEX/IATEX}}{\longrightarrow} < \text{dvi} > \stackrel{\text{dvipdfm}(x)}{\longrightarrow} < \text{PDF} >.$

The PDF printing of \hat{X}^{1} MT_EX structural formulas has been discussed under the title "Articles, Books, and Internet Documents with Structural Formulas Drawn by \hat{X}^{1} MT_EX — Writing, Submission, Publication, and Internet Communication in Chemistry." [16], where such state-of-the-art routes as described above have been compared in preparing PDF documents with \hat{X}^{1} MT_EX structural formulas.

Although the above-mentioned methods are capable of providing PDF documents of high quality, they are incapable of converting dvi files into PDF files, when the dvi files have been prepared from $L^{2}T_{E}X$ document files containing $X^{2}MT_{E}X$ codes (not EPS files). In other words, the dvi files cannot be treated directly by the dvipdfm(x) converter. Because PDF files become more and more popular in

version	package files and comments
1.00 (1993)	(for LATEX2.09) See Ref. [1, 2]. aliphat.sty, carom.sty, lowcycle.sty, hetarom.sty, hetarom.sty, hetarom.sty, chemstr.sty, locant.sty, xymtex.sty
$1.01\ (1996)$	(for $\operatorname{IAT}_{E} X 2_{\mathcal{E}}$) See Ref. [3]. ccycle.sty, polymers.sty, chemist.sty
1.02(1998)	(not released) Nested substitution by 'yl'-function.
2.00 (1998)	Enhanced version based on the $\hat{X^{1}\!M}$ Notation. See Ref. [4, 5, 6]. fusering.sty, methylen.sty
$2.01\ (2001)$	(not released) Size reduction, sizeredc.sty (version 1.00)
3.00 (2002)	Size reduction (sizeredc.sty, version 1.01), and reconstruction of the command system. See Ref. [7]
4.00 (2002)	(not released) PostScript printing (xymtx-ps.sty, version 1.00 and chmst-ps.sty, version 1.00)
4.01 (2004)	The xymtx-ps package for PostScript printing and length-variable central atoms [8]
4.02(2004)	PostScript printing and wedges bonds for stereochemistry
4.03(2005)	PostScript printing and wavy bonds for stereochemistry. See Ref. [9]
4.04(2009)	Macros for drawing steroids (steroid.sty, ver 1.00). See Ref. [10]
4.05 (2009)	Macros for drawing Lewis structures of the lewissturc package (lewisstruc.sty, version 1.00), revised and improved macros added to the chemist package (ver 4.05) [and the chemst-ps package (ver 1.02)], and the first release of the chemtimes package (ver 1.00)
4.06 (2009)	The chmst-ps package (ver 1.03) for supporting bent (curved) harpoons. See Refs. [11, 12, 13]
5.00 (2010)	The present version: the xymtx-pdf package (ver 5.00) for supporting PDF printing, the bondcolor package (ver 5.00) for coloring double bonds and skeletal bonds as well as the assurelatexmode package for assuring compatibility of the three modes. This version also contains the chmst-pdf package (ver 5.00) for extending the chemist package to support PDF printing and the assurechemist package for assuring compatibility of the three modes.

Table 1.1: Versions of $\hat{X}^{1}MT_{E}X$

writing, publication and internet documentation, and because the dvipdfm(x) converter becomes the de facto standard for preparing PDF files, it is highly desirable to develop a direct route for processing dvi files prepared from T_EX/IAT_EX documents with X^{AMT_EX} codes, i.e.,

Such a direct route is now available by the present $\hat{X}^{T}MT_{E}X$ system (version 5.00), where the xymtx-pdf package added as part of the $\hat{X}^{T}MT_{E}X$ 5.00 supports the PDF-compatible mode for using the dvipdfm(x) converter or equivalent.

Because the drawing mechanism of structural formulas in the X²MT_EX 5.00 relies on graphic utilities of the pgf package developed by T. Tantau [18], you should download the the pgf package from his site: http://sourceforge.net/projects/pgf.

1.2.2 Bond Coloring

Even within the scope of the previous version (4.06), the code:

{\red \bzdrv{1=={\blue OH};4=={\green NO\$_{2}\$}}

produces the following colored structure:



On a similer line, electron shifts in a resonance structure of benzene can be colored as follows:



by writing the following code:

```
\sixheterov[bdf]
{2s=={\red\electronrshiftarrow(-40,-90)(-90,-240)};%
4s=={\red\electronlshiftarrow(-160,200)(-70,100)};%
6s=={\red\electronrshiftarrow[1](100,20)(260,20)}}{}
```

However, there has been no systematic way of coloring a specific bond to be emphasized. For example, the substitution bond between the OH group and the aromatic ring in the first example cannot be colored differently from other portions of the molecule. The three double bonds of the benzene ring in the second example cannot be colored if we rely on the previous techniques of \hat{X}^2MTEX version 4.06.

Hence, it is highly desirable to differentiate a specific bond by color if we pursue intuitively- and visually-understandable documentation. The **bondcolor** package newly added in the $\hat{X}^{2}MT_{E}X$ version 5.00 supports commands for coloring double bonds, skeletal bonds and others.

1.3 About On-Line Manuals for X²MT_EX Version 5.00

Two on-line manuals (xymtx500.pdf and xymtx500PS.pdf) are available for $\hat{X}^{1}MT_{E}X$ Version 5.00, where their contents are substantially the same. The manual xymtx500.pdf has been processed by the PDF-compatible mode, i.e.,

<tex with XÎMT_FX codes $> \stackrel{T_{E}X/IAT_{E}X}{\longrightarrow} < dvi > \stackrel{dvipdfm(x)}{\longrightarrow} < PDF >$

where the preamble of the corresponding tex file has been described as follows:

```
%xymtx500.tex
\documentclass{book}
%%\usepackage[dvips]{xymtexpdf}%draft making
\usepackage{xymtexpdf}
\usepackage{chmst-pdf}
\begin{document}
<text with XyMTeX codes>
\end{document}
```

The resulting dvi file has been converted into the PDF file (xymtx500.pdf) by means of the dvipdfmx converter (cf. Section 2.2). Note that \usepackage[dvips]{xymtexpdf} has been used during draft making, where the resulting PostScript file has been browsed by GSview coupled with Ghostscript (cf. Section 2.3).

On the other hand, the manual xymtx500PS.pdf has been processed by the PostScript-compatible mode, i.e.,

 $< \text{tex with XIMT}_{EX} \text{ codes} > \xrightarrow{T_{EX}/IAT}_{EX} < \text{dvi} > \xrightarrow{\text{dvips}} < \text{ps} > \xrightarrow{\text{Distiller}} < \text{PDF} >$

where the preamble of the corresponding tex file has been described as follows:

```
%xymtx500PS.tex
\documentclass{book}
\usepackage{xymtexps}
\usepackage{chmst-ps}
\begin{document}
<text with XyMTeX codes>
\end{document}
```

The text <text with XyMTeX codes> of the PostScript-compatible mode is essentially the same as the counterpart of the PDF-compatible mode described above. The resulting dvi file has been converted into the PDF file (xymtx500PS.pdf) via a PostScript file (dvips and Adobe Distiller).

Although these two manuals have essentially the same context, they are different in their document sizes (xymtx500.pdf ca. 3.5 Mbyte vs. xymtx500PS.pdf ca. 1 Mbyte).

XIMT_EX Version 5.00

2.1 Package Files of X²MT_EX Version 5.00

The X²MT_EX system (version 5.00) consists of the package files listed in Table 2.1, where the package xymtx-pdf.sty has been developed to realize the PDF-compatible mode.

The XIMT_EX system (version 5.00) supports three modes of structural drawing.

- 1. T_EX/IAT_EX Mode: The declaration of \usepackage{xymtex} in the preamble of a tex file results in the reading of all the package files listed in the X¹MT_EX Structural Files of Figure 2.1, which permits X¹MT_EX drawing according to the IAT_EX picture environment and the epic package (modified slightly). Please consult the on-line manuals of X¹MT_EX version 4.06 and before.
- 2. PostScript-Compatible Mode: The declaration of \usepackage{xymtexps} in the preamble of a tex file results in the reading of the xymtx-ps package as well as all the package files listed in the X⁴MT_EX Structural Files of Figure 2.1. The drawing of structural formulas is based on the L⁴T_EX picture environment and the pstricks package [17]. Please consult the on-line manuals of X⁴MT_EX version 4.06 and before. A dvi file produced by T_EX/L⁴T_EX processing should be further converted into a ps file by dvips (or equivalents) in order to browse X⁶MT_EX structural formulas:

 $< \text{tex with X^{1}MT_{FX} codes} > \xrightarrow{T_{EX}/L^{A}T_{EX}} < \text{dvi} > \xrightarrow{\text{dvips}} < \text{ps} > \xrightarrow{\text{Distiller}} < \text{PDF} >.$

Of course, an alternative process using a tex file with X²MT_EX EPS files:

<tex with XÎMT_FX EPS files $> \xrightarrow{T_EX/LAT_EX} < dvi > \xrightarrow{dvips} < ps > \xrightarrow{Distiller} < PDF >$

is also effective in this mode. As a result, a tex file for drawing structural formulas is permitted to contain both $\hat{X}^{2}MT_{F}X$ codes and $\hat{X}^{2}MT_{F}X$ EPS files.

3. **PDF-Compatible Mode**: The declaration of \usepackage{xymtexpdf} in the preamble of a tex file results in the reading of the xymtx-pdf package as well as all the package files listed in the X¹MT_EX Structural Files of Figure 2.1. The drawing of structural formulas is based on the L^AT_EX picture environment and the pgf package [18]. A dvi file produced by T_EX/L^AT_EX processing should be further converted into a PDF file by dvipdfm(x) in order to browse X¹MT_EX structural formulas by means of Adobe Reader. The direct process described on Page 8:

 $< \mathrm{tex \ with \ } X^{\mathrm{T}}_{\mathrm{F}} X \ \mathrm{codes} > \overset{\mathrm{T}_{\mathrm{E}} X/\mathrm{L}^{\mathrm{A}}\mathrm{T}_{\mathrm{E}} X}{\longrightarrow} < \mathrm{dvi} > \overset{\mathrm{dvipdfm}(x)}{\longrightarrow} < \mathrm{PDF} >.$

is now realized by the PDF-compatible mode.

Of course, the previous process using a tex file with X¹MT_FX EPS files:

Table 2.1: Package Files of $\hat{X}^{2}MT_{E}X$ and Related Files

package name	included functions
XIMT _E X Struc	tural Files
aliphat.sty	macros for drawing aliphatic compounds
carom.sty	macros for drawing vertical and horizontal types of carbocyclic compounds
lowcycle.sty	macros for drawing five-or-less-membered carbocycles.
ccycle.sty	macros for drawing bicyclic compounds etc.
hetarom.sty	macros for drawing vertical types of heterocyclic compounds
hetaromh.sty	macros for drawing horizontal types of heterocyclic compounds
hcycle.sty	macros for drawing pyranose and furanose derivatives
chemstr.sty	basic macros for atom- and bond-typesetting
locant.sty	macros for printing locant numbers
polymers.sty	macros for drawing polymers
fusering.sty	macros for drawing units for ring fusion
methylen.sty	macros for drawing zigzag polymethylene chains
sizeredc.sty	macros for size reduction
steroid.sty	macros for drawing steroid derivatives contained in the steroid package
lewissturc	macros for drawing Lewis structures
bondcolor	macros for coloring double bonds and skeletal bonds ($\hat{X}MTEX$ Version 5.00)
assurelatexmode	dummy declaration for compatibility of the three modes ($\hat{X}MT_{E}X$ Version 5.0
Packages for P	ostScript- and PDF-Compatible Modes
xymtx-ps.sty	macros for PostScript printing ($\hat{X}^{f}MT_{E}X$ Version 4.02). These macros are su
	stituted for several macros contained in the chemstr package.
xymtx-pdf.sty	macros for PDF printing ($\hat{X}^{2}MT_{E}X$ Version 5.00). These macros are substitute
	for several macros contained in the chemstr package.
XIMTEX Utilit	ies
xymtex.sty	a package for calling all package files (listed in XIMTEX Structural Files)
	without loading xymtx-ps.sty and xymtx-pdf.sty (for the T _E X/IAT _E X mode)
xymtexps.sty	a package for calling all package files (listed in $\hat{X}MT_{E}X$ Structural Files)
	and xymtx-ps.sty (for the PostScript-compatible mode). Not with xymt
	pdf.sty
xymtexpdf.sty	a package for calling all package files (listed in $\hat{X}MT_{E}X$ Structural Files)
	and xymtx-pdf.sty (for the PDF-compatible mode). Not with xymtx-ps.sty
Related Files	
chemist.sty	commands for using 'chem' version and chemical environments
assurechemist.sty	dummy commands for compatibility of the three modes (Version 5.00)
chmst-ps.sty	macros for PostScript printing. These macros are substituted for several macro
	contained in chemist package.
مام معمد مع مالا معد م	macros for PDF printing. These macros are substituted for several macro
chmst-pdf.sty	macros for i Di printing. These macros are substituted for several macro

< tex with XIMT_FX EPS files $> \xrightarrow{T_EX/LAT_EX} < dvi > \xrightarrow{dvipdfm(x)} < PDF >$

is also effective in this mode. It follows that a tex file is permitted to contain both $\hat{X}^{2}MT_{E}X$ codes and $\hat{X}^{2}MT_{E}X$ EPS files for drawing structural formulas.

The T_EX/IAT_EX mode and the PostScript-compatible mode can be switched in a tex file. On a similar line, the T_EX/IAT_EX mode and the PDF-compatible mode can be switched in a tex file.

It should be noted, however, that the PostScript-compatible mode and the PDF-compatible mode cannot coexist in a tex file. More precisely speaking, X²MT_EX Version 5.00 does not support simultaneous loading of the xymtexps package and the xymtexpdf package or, in more basic levels, simultaneous loading of the pstricks package and the pgf package. Hence, raw commands of the pstricks package (e.g., \psline) cannot be used in the PDF-compatible mode, while raw commands of the pgf package (e.g., \draw) cannot be used in the PostScript-compatible mode.

The exclusive loading between the PostScript-compatible mode and the PDF-compatible mode, however, is not an essential drawback in practical use of the $\hat{X}^{1}MT_{E}X$ system. Because the commands defined in the package files of the $\hat{X}^{1}MT_{E}X$ Structural Files (Figure 2.1) are common in the three modes, they are unnecessary to be rewritten even if one selected mode is changed into another. It follows that the PostScript-compatible mode and the PDF-compatible mode can be switched only by exchanging the declarations: $\selected = \selected =$

2.2 PDF-Compatible Mode of XIMT_EX

The present on-line manual mainly describes the PDF-compatible mode of the XIMTEX system.

2.2.1 Templates for the PDF-Compatible Mode

The macro codes for the PDF-compatible mode of X¹MT_EX are defined in xymtx-pdf.sty, which is incompatible with xymtx-ps.sty for the PostScript-compatible mode. The following template (named test1.tex) indicates a typical format for loading xymtexpdf.sty for the PDF-compatible mode of the X¹MT_EX system.

```
%test1.tex
\documentclass{article}
\usepackage{xymtexpdf}
\begin{document}
(\XyMTeX{} macros)
\end{document}
```

When you load the utility package xymtexpdf.sty by using \spackage , all the package files listed in $\hat{X}^{1}MT_{E}X$ Structural Files (Table 2.1) as well as xymtx-pdf.sty for the PDF-compatible mode of the $\hat{X}^{1}MT_{E}X$ system are loaded to draw structural formulas. Note that the utility package xymtexps.sty, on the other hand, loads xymtx-ps.sty for the PostScript-compatible mode as well as all the package files listed in $\hat{X}^{1}MT_{E}X$ Structural Files (Table 2.1).

2.2.2 LATEX Processing

For the purpose of LATEX processing, you should type the following command in the command prompt:

```
elatex test1
```

where the extended eT_EX version of $L^{A}T_EX$ is used. In Japanese environment of such an extended eT_EX version, type in the command prompt as follows:

eplatex test1

Thereby, the corresponding dvi file (test1.dvi) is produced.

2.2.3 Conversions by dvipdfm(x)

The dvi file is ready to be processed by the dvipdfm(x) converter by typing the following command in the prompt:

dvipdfmx test1

which produces a PDF file (named test1.pdf) which contains structural formulas drawn by the $\hat{X}^{1}MT_{E}X$ system.

2.3 Option "dvips"

2.3.1 LATEX Processing With "dvips" Option

The standard PDF-compatible mode (Subsections 2.2.1 and 2.2.2) produces a dvi file suitable to the subsequent processing by the dvipdfm(x) converter, where the dvi file is converted into a PDF file. The PDF-compatible mode can alternatively produce a dvi file suitable to the dvips converter, which generates a PostScript file.

For the latter purpose, you should use a tex file in accord to the following template (named test2.tex), where xymtexpdf.sty is loaded with the option "dvips":

```
%test2.tex
\documentclass{article}
\usepackage[dvips]{xymtexpdf}
\begin{document}
(\XyMTeX{} macros)
\end{document}
```

For the purpose of LATEX processing, you should type the following command in the command prompt:

elatex test2

where the extended eT_EX version of L^AT_EX is used. In Japanese environment of such an extended eT_EX version, type in the command prompt as follows:

eplatex test2

Thereby, the corresponding dvi file (test2.dvi) is produced. The dvi file is ready to be processed by the dvips converter by typing the following command in the prompt:

dvips test2

The resulting PostScript file (named test2.ps) contains structural formulas drawn by the XIMTFX system.

Representative Examples

The scope of organic structural formulas to be drawn is common to the three modes of the $\hat{X}^{T}MT_{E}X$ system. This chapter is devoted to show that the present PDF mode is also effective to draw representative structural formulas which have been described in the previous documents for the other two modes (the $T_{E}X/I^{A}T_{E}X$ mode and the PostScript-compatible mode).

3.1 Acyclic Compounds

To draw acyclic compounds, the codes:

\Rtrigonal{0==C;1D==0;2==C1;3==F}\qquad \Ltrigonal{0==C;1D==0;2==C1;3==F}

produce the following structures:



Ethylene derivatives are drawn by the codes:

 $\label{l=C;2=C}{1==F;2==C1;3==H;4==Br}\quad\\ Ethyleneh{1==C;2==C}{1==CH$_{3}$;2==H;3==CH$_{2}$0H;4==H}\par\\ Ethyleneh{1==C;2==N}{1==Ph;2==Ph;3==OH}\quad\\ Ethyleneh[t{2+}]{1==C;2==N}{1==CH$_{3}$;2==CH$_{3}$;3==H}$

which produce the following structures:



3.2 Carbocyclic Compounds

As examples of carbocyclic compounds, decaline skeletons of various directions are drawn by the following codes. Examples:

```
\decalinevb{1D==0;0FB==H;0GA==H} \qquad
\decalinevb{1B==CH$_{2}$0SiR$_{3}$;3D==0;4A==CH$_{3}$0C0;%
0FB==CH$_{3}$;0GA==H}
```

These commands produce:



3.3 Aromatic Compounds

As examples of aromatic compounds, benzene derivatives are drawn by the following codes:

\bzdrv{1==0H}
\bzdrv{2==0H}
\bzdrv[r]{1==0H}
\bzdrv[1]{1==0H}
\bzdrv[A]{1==0H}

These commands produce:



Naphthalene derivatives are drawn by the following codes:

\naphdrv{1==0H}
\naphdrv{2==0H}
\naphdrv[A]{1==0H}

These commands produce:



3.4 Steroid Derivatives

The following example shows a steroid derivative having a side chain drawn by the Newman projection.



Estradiol, which is a much more potent estrogen than estrone, is drawn by using the command **\steroid** as follows:



The formula of mifepristone (RU-486), which is a synthetic steroid used as a abortifacient in the first two month of pregnancy, is drawn by using the command **\stereoid**.



3.5 Hetrocyclic Derivatives

Penicillin V can be drawn as follows:

\begin{XyMcompd}(2100,600)(-800,100){}{}
\fourhetero[{b\fivefusevi{1==S;4==\null}{2Sa==CH\$_{3}\$;2Sb==CH\$_{3}\$;3A==COOH}{d}]%
{2==N}{1D==0;3FA==H;4GA==H;4Su==\lyl(4==OCH\$_{2}\$CONH){4==\bzdrh{4==(yl)}}}
\end{XyMcompd}



A bold dash bond may be used instead of a bold wedged bond according to IUPAC Recommendations 1996 [19]. By declaring the switching command \dashhasheddash, one can draw structural formulas by using the combination of bold dash bonds and hashed dash bonds. For example, penicillin V can be drawn as follows:

\begin{XyMcompd}(2100,600)(-800,100){}{}
\fourhetero[{b\fivefusevi{1==S;4==\null}{2Sa==CH\$_{3}\$;2Sb==CH\$_{3}\$;3A==COOH}{d}]%
{2==N}{1D==0;3FA==H;4GA==H;4Su==\lyl(4==OCH\$_{2}\$CONH){4==\bzdrh{4==(yl)}}}
\end{XyMcompd}



Stereochemistry

In the present PDF-compatible mode (due to \usepackage{xymtexpdf}), wedged bonds for stereochemistry are drawn on the basis of the graphics mechanism supported by the pgf package. On the other hand, the PostScript-compatible mode (due to \usepackage{xymtexps}) has relied on the pstricks package. However, each drawing unit in either case is placed on the basis of the LATEX picture environment, so that appearances of generated structures are not different so much, even if either of the two modes is selected. The following examples show how the PDF-compatible mode produces structural formulas exhibiting similar appearances to the PostScript-compatible mode.

4.1 Wedged Bonds

According to IUPAC Recommendations 1996 [19], the default mode of the $\hat{X}^{2}MT_{E}X$ system adopts wedge bonds for representing β bonds and hashed dash bonds for representing α bonds. For example, illudin S, an anti-tumor antibiotic substance, is drawn in two ways in which the directions of wedges are altered:

```
\nonaheterovi[di]{5s=\cyclopropanev{2==(y1)}}%
{2SB==CH$_{3}$;2SA==CH$_{2}$0H;3B==0H;4==CH$_{3}$;6SB==CH$_{3}$;6SA==H0;7D==0}
\hskip1cm
\nonaheterovi[di]{5s==\cyclopropanev{2==(y1)}}%
{2FB==CH$_{3}$;2GA==CH$_{2}$0H;3B==0H;4==CH$_{3}$;%
6GB==\lmoiety{H$_{3}$C};6FA==H0;7D==0}
```



In some cases, hashed wedge bonds are used to designate α -bonds. The X¹MT_EX system supports this type of representations, where the \wedgehashedwedge command changes dashed bonds into such hashed wedge bonds.

```
\wedgehashedwedge
\nonaheterovi[di]{5s==\cyclopropanev{2==(y1)}}%
{2SB==CH$_{3}$;2SA==CH$_{2}$0H;3B==OH;4==CH$_{3}$;6SB==CH$_{3}$;6SA==H0;7D==0}
\hskip1cm
```

\nonaheterovi[di]{5s==\cyclopropanev{2==(y1)}}%
{2FB==CH\$_{3}\$;2GA==CH\$_{2}\$0H;3B==0H;4==CH\$_{3}\$;%
6GB==\lmoiety{H\$_{3}\$C};6FA==H0;7D==0}



If the switch \wedgehasheddash is declared, the drawing mode is returned to the default mode, as shown in the following examples:

\wedgehashedwedge

\cyclohexaneh{2D==0;4GA==H;4==\cyclohexaneh{1==(yl);3D==0;1GA==H}}\hskip1cm \wedgehasheddash%return to the default mode

\cyclohexaneh{2D==0;4GB==H;4==\cyclohexaneh{1==(yl);3D==0;1GA==H}}\hskip1cm \wedgehashedwedge

\cyclohexaneh{2D==0;4GA==H;4==\cyclohexaneh{1==(yl);3D==0;1GB==H}}



Thus, the central structure is drawn in the default mode.

4.2 Wavy Bonds

Wavy bonds drawn by the PDF mode of the X²MT_EX system are shown in the following examples:

```
\sixheterov[e]{}{1U==OAc; 3A==; 4B==COOEt; 5==}
\hskip2cm
\sixheterov[{e\fourfuse{}{4Sa==; 4Sb==}{b}}]{}
{4D==0; 3SU==Me; 3SV==COOMe; 5FB==H; 6GB==H}
OAc
OAc
H
COOMe
H
COOMe
```

COOEt

where a bond specifier U in such designations as 1U=0Ac is used to draw a wavy line. Note that a ring fusion is realized by writing \fourfuse in the BONDLIST of \sixheterov.

Ο

An additional example is shown as follows:

```
\fiveheterovi{1s==\sixheterov[f]{}{4==(y1);2D==0;1==}}%
{2U==CN;5Su==\LtetrahedralS{1==(y1);2==;3==}}
```



where a bond specifier U in such designations as 1U==CN is used to draw a wavy line. Note that a spiroring is generated by writing \sixheterov in the ATOMLIST of \fiveheterovi in a nested fashion.

Optional Bonds

The graphics mechanism of the pgf package can be directly applied to draw extra bonds which cannot be drawn by the standard recipes of the $\hat{X}^{2}MT_{E}X$ system. However, direct applications of raw commands of the pgf package spoil the compatibility of a tex file containing $\hat{X}^{2}MT_{E}X$ codes. For the sake of assuring compatibility, such macros as WedgeAsSubst, PutBondLine and PutDashedBond have been defined to pursue the same purpose conveniently.¹

5.1 Bold Bonds of Cyclic Skeletons

5.1.1 Furanoses

This subsection is a rewritten version of Section 3.2 in the on-line manual of $\hat{X}^{2}MT_{E}X$ versions 4.02 and 4.03 (xymtx402403.pdf), where the use of the raw command \psline of the pstricks package is ceased for the same of compatibility to the PDF mode.

The default expression of a furanose skeleton has three front skeletal bonds of bold dashes. For example, the structural formula of ribavirin is drawn by the code:

\furanose{1Sa==H;2Sb==H;2Sa==OH;3Sb==H;3Sa==OH;4Sa==H;4Sb==HOC\rlap{H\$_{2}\$; 1Sb==\fiveheterov[bd]{1==N;2==N;4==N}{1==(y1);3==CONH\$_{2}\$}



Although the quality of the resulting diagram is sufficient to be printed, one may require a more sophisticated format in which the three front bonds are expressed by the combination of wedge-dash-wedge. This type of formats can be drawn by using the command \WedgeAsSubst and the \PutBondLine command both of which are defined in the xymtx-pdf package:

\WedgeAsSubst(Starting Point)(Slope){LineLength}
\PutBondLine(Starting Point)(End Point){LineWidth}

¹The PutBondLine command may be replaced by PutPDFLine in the PDF-compatible mode or by PutPSLine in the PostScript-compatible mode. However, such replacement is not recommended.

Thereby, the code:

```
\begin{XyMcompd}(1200,1000)(100,0){}{}
\fivesugarh{5==0;1s==\WedgeAsSubst(0,0)(-3,-5){120};%
4s==\WedgeAsSubst(0,0)(3,-5){120};%
3s==\PutBondLine(-15,0)(305,0){3.2pt}%
}{1Sa==H;2Sb==H;2Sa==0H;3Sb==H;3Sa==0H;4Sa==H;4Sb==H0C\rlap{H$_{2}$};
1Sb==\fiveheterov[bd]{1==N;2==N;4==N}{1==(y1);3==C0NH$_{2}$}%
}[abc]
\end{XyMcompd}
```

generates the following formula:



This input code can be simplified by making a tentative macro named \myfuranose as follows:

```
\makeatletter
\def\myfuranose{\@ifnextchar[{\@myfuranose}{\@myfuranose[]}}
\def\@myfuranose[#1]#2{%
\fivesugarh[#1]{5==0;1s==\WedgeAsSubst(0,0)(-3,-5){120};%
4s==\WedgeAsSubst(0,0)(3,-5){120};%
3s==\PutBondLine(-15,0)(305,0){3.2pt}}{#2}[abc]}
\makeatother
```

Thereby, the same formula can be typeset by writing a more simplified code:

\myfuranose{1Sa==H;2Sb==H;2Sa==OH;3Sb==H;3Sa==OH;4Sa==H;4Sb==HOC\rlap{H\$_{2}\$; 1Sb==\fiveheterov[bd]{1==N;2==N;4==N}{1==(y1);3==CONH\$_{2}}



The tentative macro $\mbox{myfuranose}$ is convenient to draw various furanoses, e.g., α -D-ribofuranose and its 5-phosphoric acid:

```
\myfuranose{1Sa==OH;1Sb==H;2Sb==H;2Sa==OH;3Sb==H;3Sa==OH;4Sa==H;%
4Sb==HOC\rlap{H$_{2}$}
\myfuranose{1Sa==OH;1Sb==H;2Sb==H;2Sa==OH;3Sb==H;3Sa==OH;4Sa==H;%
4Sb==H$_{2}$0$_{3}$POC\rlap{H$_{2}$}
```

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Simply by converting myfuranose into furanose, the corresponding default expressions can be obtained. Thus, the default structural formulas of α -D-ribofuranose and its 5-phosphoric acid are obtained by inputting the following codes:

\furanose{1Sa==OH;1Sb==H;2Sb==H;2Sa==OH;3Sb==H;3Sa==OH;4Sa==H;%
4Sb==HOC\rlap{H\$_{2}\$}
\furanose{1Sa==OH;1Sb==H;2Sb==H;2Sa==OH;3Sb==H;3Sa==OH;4Sa==H;%

\Iuranose{15a==0H;15b==H;25b==H;25a==0H;35b==H;35a==0H;45a==H 4Sb==H\$_{2}\$0\$_{3}\$POC\rlap{H\$_{2}}}



5.1.2 Optional Arguments for Drawing Skeletal Bonds

As found in the on-line manual of $\hat{X}^{2}MT_{E}X$ version 2.00 [6], the second argument of the \ryl command can accommodate substituents other than a substituent generated by the "yl" function. For example, the inner code $ryl{OA==Me;...}$ in the code,

```
\sixheterov({bB}{eA}){3==0;5==0}{1A==Me;4Sa==\null;4Sb==\null;%
6==\pentamethylenei[a]{}{4B==0H;5B==Me;5==(yl)};%
2==\ryl{0A==Me;5==\sixheterov({eA}){3==0;5==0}{6==(yl);1B==Me;%
4Sa==\null;4Sb==\null}}
```

represents a methyl group on a vertex due to the command \ryl. Thereby, we have



The optional argument ({bB}{eA}) of \sixheterov indicates that bond b is a β -bond (a bold bond) and bond e is an α -bond (a dashed bond) in the left six-membered ketal.

Because three bonds (ring skeletal bonds) in the above structure are shorter than the other polymethylene bonds of standard bond length, they should be lengthened, rigorously speaking. For this purpose, the same compound can be alternatively drawn by a spiro technique, where, for example, the code 6s==\hexamethylenei... is designated in the atom list of \shixheterov. \begin{XyMcompd}(1800,800)(-600,100){}{}
\sixheterov({bB}{eA}){3==0;5==0;%
6s==\hexamethylenei[a]{}{4B==0H;5B==Me;6==(y1)};%
2s==\trimethylene{%
3s==\sixheterov({eA}){3==0;5==0}{6==(y1);1B==Me;%
4Sa==\null;4Sb==\null}
}{1==(y1);2A==Me}{1A==Me;4Sa==\null;4Sb==\null}
\end{XyMcompd}



The skeletal bold bond in the left six-membered ketal can be changed into a wedge bond by using the **\WedgeAsSubst** command as follows:

```
\begin{XyMcompd}(1800,800)(-600,100){}{}
\sixheterov({eA}){3==0;5==0;%
6s==\hexamethylenei[a]{}{4B==0H;5B==Me;6==(y1)};%
2s==\trimethylene{%
3s==\sixheterov({eA}){3==0;5==0}{6==(y1);1B==Me;%
4Sa==\null;4Sb==\null}
}{1==(y1);2A==Me};%
2s==\WedgeAsSubst(0,0)(0,-1){150}%
}{1A==Me;4Sa==\null;4Sb==\null}[b]
\end{XyMcompd}
```



5.2 Linkage Between Non-Adjacent Atoms

This subsection is a partially rewritten version of Subsection 2.4.2 in the on-line manual of $\hat{X}^{1}MT_{E}X$ versions 4.04 (xymtx404.pdf), where the use of the raw command \psline of the pstricks package is ceased for the same of compatibility to the PDF mode.

The linkage between the C-3 and the C-5 results in the formation of a cyclopropane ring as well as a cyclopentane ring. The linking bond is drawn by using **\PutDashedBond**, where the format of arguments is defined as follows:

\PutDashedBond(Starting Point)(End Point){LineWidth}

The dimension register \thickLineWidth is equal to 1.6pt as a default setting.



The linkage between the C-5 and the C-7 results in the formation of a cyclopropane ring as well as a cyclopentane ring. The linking bond is also drawn by using \PutDashedBond, as shown in the following example.



A peroxide bridge between non-adjacent positions of a steroid skeleton cannot be drawn by standard techniques supported by the $\hat{X}^{2}MT_{E}X$ system. But optional commands of the PDF mode of the $\hat{X}^{2}MT_{E}X$ system can be used in the arguments of $\hat{X}^{2}MT_{E}X$ commands. For example, a peroxide bridge can be drawn by means of $\forall PutBondLine$, where the $\forall thinLineWidth (= 0.4pt)$ specifies the width of a line.



Drawing a peroxide bridge between the C-3 and the C-9 requires a more complicated set of commands for the PDF mode, i.e., \PutBondLine and \PutDashedBond, where \thinLineWidth is equal to 0.4pt, while \thickLineWidth is equal to 1.6pt. Although the \PutBondLine and the \PutDashedBond are used in the argument of \put in the following program, the \put can be omitted after appropriate adjustment of output positions. 3α ,9-epidioxy- 5α -androstan-17-one



Commands \PutPDFLine and \PutPDFdashed for the PDF mode (or \PutPSLine and \PutPSdashed for the PostScript mode) would be used in place of the commands \PutBondLine and \PutDashedBond. Because they are as local as defined in the xymtx-pdf (or xymtx-ps) package only, as found by the fact that their names contain "PDF" (or "PS"), you should use \PutBondLine and \PutDashedBond in order to pursue the compatibility to the PDF and the PostScript modes.

5.3 Variable Bond Lengths

 CH_3

The function of variable bond lengths supported by $\hat{X}^{1}MT_{E}X$ (Version 4.01 and later) provides us with an elegant solution to draw structures which are crowded around a tetrahedral carbon. Such a structure can be drawn by using the redefined command \tetrahedral. Thus, the codes:

```
\bscolorswOFF%no bond coloring
%\fbox{%
\begin{XyMcompd}(3800,1500)(-100,-700){cpd:06}{}
\tetrahedral{0==C\rlap{$^{*}};1==CH$_{3};2==CH$_{3};3==CH$_{3};%
4==\tetrahedral{0==C0;2==(y1);%
4==\tetrahedral{0==CH;2==(y1);%
3==\fiveheterovi{1==N;3==N}{1==(y1);2D==0;5D==0;4==C$_{2}$H$_{5}$0;%
3==\ry1(3==CH$_{2}$){4==\bzdrh{1==(y1)}};%
4==\tetrahedral{0==CONH;2==(y1);%
4==\bzdrh{1==(yl);2==Cl;%
5==\tetrahedral{0==NHCO;2==(y1);%
4==\tetrahedral{0==CH;2==(y1);3==C$_{2}$H$_{5};%
4==\ryl(4==0){4==\bzdrh{1==(y1);2==C$_{5}$H$_{11}$-$t$;%
4==C$_{5}$H$_{11}$-$t$}%
}}<,,,50>}<,,,50>}}}<,,250,>}}%
\end{XyMcompd}%
%}
```

typeset the following structure:



It should be noted that this drawing is started at the asterisked carbon by using the redefined command **\tetrahedral**. The command **\bscolorswOFF** is a switch for stopping bond coloring. See Chapter 11.

Size Reduction and Enlargement

For size reduction and enlargement, see the on-line manual of $\hat{X}^{2}MT_{E}X$ version 3.00 (xymtx300.pdf). This chapter is devoted to confirm the fact that the mechanism of size reduction and enlargement is effective to the PDF mode of $\hat{X}^{2}MT_{E}X$ version 5.00

6.1 Utility of XIMT_EX

The $\hat{X}^{1}MT_{E}X$ system supports the \changeunitlength command for changing sizes of structural formulas, where the unit length defined by \unitlegth is changed in $\hat{X}^{1}MT_{E}X$ drawing. The following examples show the use of the \changeunitlength command in size reductions.

```
\steroid{3B==H0;5A==OH}%default unit length = 0.1pt
{\changeunitlength{0.08pt} \steroid{3B==H0;5A==OH}}
{\changeunitlength{0.05pt} \steroid{3B==H0;5A==OH}}
```



Note that font sizes are also changed during the size reduction. In contrast, the direct change of \unitlength causes no change of font sizes. The following examples shows the use of \substfontsize for changing font sizes:

```
{\unitlength=0.08pt \steroid{3B==H0;5A==OH}}
{\unitlength=0.08pt \let\substfontsize=\small \steroid{3B==H0;5A==OH}}
{\unitlength=0.08pt \let\substfontsize=\footnotesize \steroid{3B==H0;5A==OH}}
```



It should be added that the line width of each bond is not changed during size reduction and enlargement. This behavior is amplified by changing a standard line into a thicker one. Because the standard line width is stored in \thinLineWidth (0.4pt) as a letter string, it can be changed by its redefinition as follows:

```
\def\thinLineWidth{0.8pt}
\steroid{3B==H0;5A==OH}%default unit length = 0.1pt
{\changeunitlength{0.08pt} \steroid{3B==H0;5A==OH}}
{\changeunitlength{0.05pt} \steroid{3B==H0;5A==OH}}
```



Undoubtedly, the line width of 0.8pt is unchanged in these structural formulas.

6.2 Utilities Supported by the graphicx Package

The \scalebox command of the graphicx package can also be used to change the sizes of structural formulas. Thus, essentially equivalent results to those of the preceding subsection can be obtained as follows:

```
\steroid{3B==H0;5A==OH}
\scalebox{0.8}{\steroid{3B==H0;5A==OH}}
\scalebox{0.5}{\steroid{3B==H0;5A==OH}}
```



Because the \scalebox command causes total size reduction (or enlargement) of a structural formula as an object, bond widths and font sizes are changed simultaneously. This behavior is amplified by changing a standard line into a thicker one, as follows:

```
\def\thinLineWidth{0.8pt}
\steroid{3B==H0;5A==OH}
\scalebox{0.8}{\steroid{3B==H0;5A==OH}}
\scalebox{0.5}{\steroid{3B==H0;5A==OH}}
```



where the initial width (0.8pt) is reduced in accord with a scale factor selected as equal to 0.8 and 0.5. The \scalebox command is capable of taking an optional argument, by which the aspect ratio of a structural formula can be changed, as shown in the following examples:

\scalebox{0.8}[0.5]{\steroid{3B==H0;5A==OH}} \scalebox{0.5}[0.8]{\steroid{3B==H0;5A==OH}} \scalebox{0.8}[1.5]{\steroid{3B==H0;5A==OH}}



The chmst-pdf Package

The chmst-pdf package based on the LATEX picture environment and the pgf package is an enhanced version of the chemist package, which provides us with a useful set of utilities for cooperating with the \hat{X}^2MT_EX system. The chmst-pdf package has been designed to be used under the PDF-compatible mode of the \hat{X}^2MT_EX system, which works after reading the xymtexpdf package. Such a set of utilities as supported by the chmst-pdf package is essentially equivalent to that of the chmst-ps based on the LATEX picture environment and the pstricks package (the PostScript-compatible mode of the \hat{X}^2MT_EX system, which works after reading the xymtexpdf package.

7.1 Use of the chmst-pdf Package

To use the chmst-pdf package, the file chmst-pdf.sty should be loaded by using the \usepackage command after reading xymtexpdf.sty, e.g.,

```
\documentclass{article}
\usepackage{xymtexpdf}%PDF-compatible mode
\usepackage{chmst-pdf}
\begin{document}
(Contents)
\end{document}
```

Subsequent $L^{AT}EX$ processing (Subsection 2.2.2)² and dvipdfm(x) conversion (Subsection 2.2.3) produce a PDF file which contains objects due to chmst-pdf package as well as $\hat{X}^{A}MTEX$ structural formulas.

7.2 Arrows of Fixed Lengths

Because longer arrows of fixed lengths are frequently used in chemical equations, they are supported by the chemist (chmst-pdf) package.³ Such commands for drawing longer arrows should be used in an in-text or display math mode (e.g., \$\llongrightarrow\$ and \ChemForm{\llongrightarrow}), because they are defined as relational operators. The appearances of arrows produced by a command of the same name in chemist and chmst-pdf are different, as summarized in Table 7.1.

¹Because the chmst-ps package and the present chmst-pdf package have the same set of utilities, the on-line manual of the chmst-ps package (xymtx405406B.pdf) is effective to the present chmst-pdf package.

²It should be emphasized that the processing requires $eIAT_{EX} 2_{\varepsilon}$ (an eT_{EX} extended version of $IAT_{EX} 2_{\varepsilon}$). In particular, arrows of the pgf package would disappear if a usual $IAT_{EX} 2_{\varepsilon}$ is used.

³Arrows and harpoons produced by chemist and chmst-ps packages have been discussed in Sections 2 and 5 of the on-line manual of versions 4.05 and 4.06 (xymtx405406.pdf). If the combination of xymtexps, chemist, and chmst-ps (the PostScript-compatible mode) is replaced by the combination of xymtexpdf, chemist, chmst-pdf (the PDF-compatible mode), the descriptions of the on-line manual for the former combination are applicable to the latter combination.

d	chemist	abmat ndf	comment
command	cnemist	chmst-pdf	comment
\llongrightarrow	\longrightarrow	\longrightarrow	
\llongleftarrow	←	←──	
$\line \$	\longleftrightarrow	\longleftrightarrow	
\Llongrightarrow	\implies	\rightarrow	
\Llongleftarrow	<	\leftarrow	
Llongleftrightarrow	\iff	\Leftrightarrow	
$\line \$	<u></u>	<u> </u>	
\lines		`	
\llongleftharpoonup	(←−−−−)	/	not supported by chemist
\llongrightharpoondown	()		not supported by chemist
\equilibarrow		<u> </u>	
\Equilibarrow			
\lllongrightarrow		>	
\lllongleftarrow	←−−−−	←───	
\lllongleftrightarrow	\longleftrightarrow	\longleftrightarrow	
\Lllongrightarrow	\implies	\implies	
\Lllongleftarrow	<		
\Lllongleftrightarrow	${\longleftrightarrow}$	\sum	
\llongleftharpoondown	<u> </u>		
\llongrightharpoonup	``		
\lllongleftharpoonup	()	<u> </u>	not supported by chemist
\lllongrightharpoondown	(\longrightarrow)		not supported by chemist
\equiliblongarrow	(`	not supported by chemist
\Equiliblongarrow	<u> </u>		
/EduitibionBarrow	<u> </u>	~	

Table 7.1: Arrows of Fixed Lengths Supported by chemist and chmst-pdf

7.3 Arrows for Organic Chemistry

Arrows due to the chemist package have been introduced in Section 12.2 of the manual of $\hat{X}^{1}MT_{E}X$ version 2.00 (xymtx200PS.pdf, cf. [5]). The list of arrows of the manual is cited for convenience, as shown in Figure 7.1, where the four arrows for representing equilibriums in the fourth row are new matters in the present version of chemist (chmst-pdf) package.⁴ The arrows in the fifth row have been renamed into the present names in order to assign the previous names to the arrows in the fourth row. Note that a combination of left and right arrows is used to represent a forward and reverse reaction, while a combination of left and right harpoons is used to represent an equilibrium.

7.4 T_EX/I^AT_EX Mode and PDF-Compatible Mode

The commands for drawing arrows (Table 7.1) can be used in \ChemForm as part of a chemical equation. The following list shows several examples, where horizontal spaces before and after each arrow (as a relational operator) are placed automatically.

 $^{^{4}}$ The corresponding summary of chemical arrows produced by chemist and chmst-ps packages have been discussed in Sections 2 and 5 of the on-line manual of versions 4.05 and 4.06 (xymtx405406.pdf).


Figure 7.1: Reaction arrows of various types

stoichiometric relationship	
$ChemForm{H_{2}+Br_{2} = 2HBr}$	$\mathrm{H}_2 + \mathrm{Br}_2 = 2\mathrm{HBr}$
forward reaction	
\ChemForm{H_{2}+Br_{2} \llongrightarrow 2HBr}	$H_2 + Br_2 \longrightarrow 2HBr$
reverse reaction	
\ChemForm{H_{2}+Br_{2} \llongleftarrow 2HBr}	$H_2 + Br_2 \longleftarrow 2HBr$
equilibrium	
\ChemForm{H_{2}+Br_{2} \equilibarrow 2HBr}	$H_2 + Br_2 \Longrightarrow 2HBr$
forward and reverse	2 2
\ChemForm{H_{2}+Br_{2} \Equilibarrow 2HBr}	$H_2 + Br_2 \rightleftharpoons 2HBr$
resonance	2 2
<pre>\ChemForm{H\sbond Br \llongleftrightarrow H^{+} Br^{-}}</pre>	$\operatorname{H}\operatorname{\!\!-\!\!-\!\!-}\operatorname{Br} \longleftrightarrow \operatorname{H}^+ \operatorname{Br}^-$

Because the present document is typeset under the PDF-compatible mode (i.e., the use of the chmst-pdf package), such newly-defined arrows as shown above are drawn by using PDF utilities. If you want to print such arrows according to the embodiment of $T_EX/I_eT_EX 2_{\varepsilon}$, you should declare the switching command \chemistsw as follows:

\chemistsw stoichiometric relationship $\Gamma_{H_{2}+Br_{2} = 2HBr}$ $\mathrm{H}_2 + \mathrm{Br}_2 = 2\mathrm{HBr}$ forward reaction $H_2 + Br_2 \longrightarrow 2HBr$ \ChemForm{H_{2}+Br_{2} \llongrightarrow 2HBr} reverse reaction $H_2 + Br_2 \longleftarrow 2HBr$ \ChemForm{H_{2}+Br_{2} \llongleftarrow 2HBr} equilibrium $H_2 + Br_2 \rightleftharpoons 2HBr$ \ChemForm{H_{2}+Br_{2} \equilibarrow 2HBr} forward and reverse $H_2 + Br_2 \rightleftharpoons 2HBr$ \ChemForm{H_{2}+Br_{2} \Equilibarrow 2HBr} resonance $H \longrightarrow Br \longleftrightarrow H^+Br^-$ \ChemForm{H\sbond Br \llongleftrightarrow H^{+} Br^{-}}

Attention should be focused on arrowheads; these appearances of arrows are inherent to the chemist package (without loading the chmst-pdf package). To return to the PDF mode, the switching command should be declared.

 $\label{eq:hermitian} $$ to characterize the state of th$

7.5 ChemEquation Environment

A ChemEquation environment is used to describe a chemical equation, where chemical compounds are printed in upright fonts. The following code is a typical example containing a chemical compound and ionic species. Thus, solid limestone $(CaCO_3)$ is almost water insoluble, but a very small quantity dissolves in water according to the following process:

 $\label{eq:caco_data} $$ CaCO_{3(s)} \longrightarrow Ca_{(aq)}^{2+} + CO_{3(aq)}^{2-} \end{ChemEquation} $$$

$$\operatorname{CaCO}_{3(\mathrm{s})} \longrightarrow \operatorname{Ca}_{(\mathrm{ag})}^{2+} + \operatorname{CO}_{3(\mathrm{ag})}^{2-}$$

$$\tag{7.1}$$

The molecular formulas are printed in upright fonts, although they are written directly without using the mathrm command. Compare this output with the following one due to an equation environment of LATEX 2_{ε} :

\begin{equation}
CaCO_{3(s)} \llongrightarrow Ca_{(aq)}^{2+} + CO_{3(aq)}^{2-}
\end{equation}

$$CaCO_{3(s)} \longrightarrow Ca^{2+}_{(aq)} + CO^{2-}_{3(aq)}$$

$$\tag{7.2}$$

where each molecular formula written without using the \mathrm command is printed in italic fonts.

Arrows for organic chemistry (Section 7.3) can also be used for outputting objects over or below arrows in inorganic chemical equations. An equivalent result is obtained by using \reactrarrow, where \scriptsize is declared to adjust the sizes of objects over and below an arrow:

\begin{ChemEquation}
Na + Al + 2H_{2}
\reactrarrow{0pt}{3cm}{\scriptsize \ChemForm{THF/140^{\circ}/3\: h}}
{\scriptsize 350^atom}
NaAlH_{4}\quad (99\%~yield)
\end{ChemEquation}
THF/140°/3 h

$$Na + Al + 2H_2 \xrightarrow{\text{THF/140°/3 h}} NaAlH_4 \quad (99\% \text{ yield}) \tag{7.3}$$

On similar lines, the following set of arrows for organic chemistry (Section 7.3) can be used to draw reaction equations for inorganic chemistry.

<pre>\reactrarrow{0pt}{1cm}{A}{B}</pre>	A
	В
<pre>\reactlarrow{0pt}{1cm}{A}{B}</pre>	▲ —
	В
<pre>\reactlrarrow{0pt}{1cm}{A}{B}</pre>	$\stackrel{A}{\longleftarrow}$
	В
<pre>\reacteqarrow{0pt}{1cm}{A}{B}</pre>	A
	В

7.6 Round Arrows for Representing Electron Migrations

7.6.1 Round Arrows or Harpoons with Autocalculated Control Points

In order to confirm the compatibility between the PostScript mode and the PDF mode, this subsection is provided as a reproduced version of Section 4.2 of the on-line manual of $\hat{X}^{2}MT_{E}X$ version 4.01 (xymtx401.pdf), which has relied on the PostScript mode. The mechanism of drawing round arrows has been improved to be based on Bezier curves, so that it becomes effective to both the PostScript mode and the PDF mode.

To illustrate the mechanisms of organic reactions, round or curved arrows are used to show electron shift or migration. The macros for drawing such round or curved arrows are defined in the chmst-ps package and also in the chmst-pdf package: \electronrshiftarrow and \electronlshiftarrow. Each of these commands takes two pairs of xy-coordinates as arguments (the starting point and the end point), from which two intermediate control points are automatically calculated to draw an arrow.

\electronrshiftarrow[Arc Direction](Starting Point)(End Point) \electronlshiftarrow[Arc Direction](Starting Point)(End Point)

The xy-coordinates of the starting and end points are relative values with respect to the origin (0,0) (fixed) of the resulting arrow. The automatic calculation of control points postulates that the x-coordinate of the stating point is smaller than the x-coordinate of the end point. Note that (Starting Point) and (End Point) are not used in a usual meaning (due to the direction of an arrow); thus, an arrow head drawn by \electronrshiftarrow appears at the end point, while an arrow head drawn by \electronlshiftarrow appears at the starting point. The optional argument [Arc Direction] indicates whether a curve (arc) is upward or downward. Even numbers ([0] (or default with no designation), [2], and [4]; wide to tight) for the optional argument [Arc Direction] represent an upward curve, while odd numbers ([1], [3], and [5]; wide to tight) for the optional argument [Arc Direction] represent a downward curve.

Several upward-curved arrows generated by these commands are shown as examples, where the arrows bend in wide to tight curves as even numbers assigned to the optional argument [Arc Direction] become larger.

[0]

```
\electronrshiftarrow(0,0)(100,100)\qquad
\electronrshiftarrow(0,0)(100,-100)\qquad
\electronrshiftarrow(0,0)(100,0)\qquad\qquad
\electronlshiftarrow(0,0)(100,100)\qquad
\left(0,0\right)(100,-100)
\electronlshiftarrow(0,0)(100,0) \\[10pt]
[2]
\electronrshiftarrow[2](0,0)(100,100)\qquad
\electronrshiftarrow[2](0,0)(100,-100)\qquad
\electronrshiftarrow[2](0,0)(100,0)\qquad\qquad
\electronlshiftarrow[2](0,0)(100,100)\qquad
\electronlshiftarrow[2](0,0)(100,-100)\qquad
\electronlshiftarrow[2](0,0)(100,0) \\[10pt]
[4]
\electronrshiftarrow[4](0,0)(100,100)\qquad
\left(100,-100\right) \right)
\electronrshiftarrow[4](0,0)(100,0)\qquad\qquad
\left(100,100\right)
\electronlshiftarrow[4](0,0)(100,-100)\qquad
\electronlshiftarrow[4](0,0)(100,0)
```



Several downward-curved arrows generated by these commands are shown as examples, where the arrows bend in wide to tight curves as odd numbers become larger.

[1] \electronrshiftarrow[1](0,0)(100,100)\qquad \electronrshiftarrow[1](0,0)(100,-100)\qquad \electronrshiftarrow[1](0,0)(100,0)\qquad\qquad \electronlshiftarrow[1](0,0)(100,100)\qquad \electronlshiftarrow[1](0,0)(100,-100)\qquad \electronlshiftarrow[1](0,0)(100,0) \\[10pt] [3] $\ensuremath{\scale{2}}\$ \electronrshiftarrow[3](0,0)(100,-100)\qquad \electronrshiftarrow[3](0,0)(100,0)\qquad\qquad $\left[3\right](0,0)(100,100)$ $\left[3\right](0,0)(100,-100)$ \electronlshiftarrow[3](0,0)(100,0) \\[10pt] [5] \electronrshiftarrow[5](0,0)(100,100)\qquad \electronrshiftarrow[5](0,0)(100,-100)\qquad \electronrshiftarrow[5](0,0)(100,0)\qquad\qquad \electronlshiftarrow[5](0,0)(100,100)\qquad \electronlshiftarrow[5](0,0)(100,-100)\qquad \electronlshiftarrow[5](0,0)(100,0)

[1]	\smile		N
	\smile		
	\checkmark	J.	

An attack of an amide anion $^{\ominus}NH_2$ on the 2-position of pyridine is illustrated as follows. According to the specification of the $\hat{X}^{1}MT_{E}X$ system, the code 1==N should be placed at the last part of the ATOMLIST (atom list) of \sixheterovi.⁵

```
\sixheterovi[ace]{2==\futuresubst{\kern6pt$^{\ominus}$NH$_{2}$;
2s==\electronlshiftarrow(10,0)(160,0);%
1s==\electronlshiftarrow[1](0,-30)(100,50);1==N}{}
```

The command \futuresubst is defined to show the amide anion $^{\ominus}NH_2$ that will be involved as a future substituent. The command \electronlshiftarrow for drawing a left curly arrow is designated in the atom list of the command \sixheterovi.

⁵Otherwise, the flag for truncating a vertex is deleted so as to print the vertex of the six-membered ring.



Similarly, the command **\electronrshiftarrow** is used to draw a right curly arrow.

```
\sixheterovi[ce]{%
1s==\electronrshiftarrow[1](50,-70)(100,50);%
2s==\electronrshiftarrow(70,20)(200,70);1==\downnobond{N}{$\ominus}}%
}{2SB==NH$_{2}$;2SA==H}
```



Arrows having an arrow head of harpoon-type are also available. An intermediate keyword in the middle of each command name (e.g., Hru in \electronHrushiftarrow) represents the feature of the harpoon, i.e., Hru (harpoon right upward), Hrd (harpoon right downward), Hlu (harpoon left upward), or Hld (harpoon left downward). The formats of these commands are the same as those of \electronrshiftarrow and \electronlshiftarrow described above, i.e.,

```
\electronHrushiftarrow[Arc Direction](Starting Point)(End Point)
\electronHrdshiftarrow[Arc Direction](Starting Point)(End Point)
\electronHlushiftarrow[Arc Direction](Starting Point)(End Point)
\electronHldshiftarrow[Arc Direction](Starting Point)(End Point)
```

The outputs of these commands are illustrated as follows:

```
[0]
\electronHrushiftarrow(0,0)(100,100)\qquad
\left(0,0\right)(100,-100)
\electronHrushiftarrow(0,0)(100,0)\qquad
\electronHlushiftarrow(0,0)(100,100)\qquad
\electronHlushiftarrow(0,0)(100,-100)\qquad
\electronHlushiftarrow(0,0)(100,0) \\[10pt]
[2]
\left[2\right](0,0)(100,100)
\electronHrushiftarrow[2](0,0)(100,-100)\qquad
\electronHrushiftarrow[2](0,0)(100,0)\qquad
\left[2\right](0,0)(100,100)
\electronHlushiftarrow[2](0,0)(100,-100)\qquad
\electronHlushiftarrow[2](0,0)(100,0) \\[10pt]
[4]
\electronHrushiftarrow[4](0,0)(100,100)\qquad
\label{eq:lectronHrushiftarrow[4](0,0)(100,-100)\quad
\electronHrushiftarrow[4](0,0)(100,0)\qquad
```

 $\label{eq:loss} $$ electronHlushiftarrow[4](0,0)(100,100)\quad electronHlushiftarrow[4](0,0)(100,-100)\quad electronHlushiftarrow[4](0,0)(100,0) $$$

$$\begin{bmatrix} 0 \end{bmatrix} \begin{pmatrix} & & & & & & & \\ & & & & & \\ 2 \end{bmatrix} \begin{pmatrix} & & & & & & \\ & & & & & \\ & & & & & \\ \end{bmatrix} \begin{pmatrix} & & & & & \\ & & & & \\ & & & & \\ \end{bmatrix} \begin{pmatrix} & & & & \\ & & & \\ & & & \\ & & & \\ \end{bmatrix} \begin{pmatrix} & & & & \\ & & & \\ & & & \\ & & & \\ \end{pmatrix} \begin{pmatrix} & & & & \\ & & \\ & & & \\ & & & \\ \end{pmatrix}$$

[1]

\electronHrushiftarrow[1](0,0)(100,100)\qquad \electronHrushiftarrow[1](0,0)(100,-100)\qquad \electronHrushiftarrow[1](0,0)(100,0) \qquad \electronHlushiftarrow[1](0,0)(100,100)\qquad \electronHlushiftarrow[1](0,0)(100,-100)\qquad \electronHlushiftarrow[1](0,0)(100,0) \\[10pt] [3] \electronHrushiftarrow[3](0,0)(100,100)\qquad \electronHrushiftarrow[3](0,0)(100,-100)\qquad $\ensuremath{\columnwidth{\columnwidth}\) (0,0) (100,0) \quad$ \electronHlushiftarrow[3](0,0)(100,100)\qquad \electronHlushiftarrow[3](0,0)(100,-100)\qquad \electronHlushiftarrow[3](0,0)(100,0) \\[10pt] [5] \electronHrushiftarrow[5](0,0)(100,100)\qquad \electronHrushiftarrow[5](0,0)(100,-100)\qquad

\electronHrushiftarrow[5](0,0)(100,0) \qquad \electronHlushiftarrow[5](0,0)(100,100)\qquad \electronHlushiftarrow[5](0,0)(100,-100)\qquad \electronHlushiftarrow[5](0,0)(100,0) \\[10pt]

7.6.2 Round Arrows or Harpoons with Given Control Points

For the purpose of representing electron migration, the chmst-pdf (or chmst-ps) package supports another set of commands for drawing round arrows, i.e., the command \electronshiftArrowr for typesetting a right round arrow and the command \electronshiftArrowl for typesetting a left round arrow. These commands take one or two additional pairs of xy-coordinates (Control Points) to set up control points explicitly:

\electronshiftArrowr(Starting Point)(Control Points)(End Point)
\electronshiftArrowl(Starting Point)(Control Points)(End Point)

Because control points are given explicitly, such an optional argument as setting up an upward or downward arc is unnecessary. Their behaviors are illustrated in the following examples:

```
\electronshiftArrowr(0,0)(100,100)(200,100)(300,0)\qquad\qquad
\electronshiftArrowr(0,0)(100,-100)(200,-100)(300,0)\qquad\qquad
\electronshiftArrowl(0,0)(100,100)(200,100)(300,0)\qquad\qquad
\electronshiftArrowl(0,0)(100,-100)(200,-100)(300,0)
```

where three or four pairs of xy-coordinates are placed; the first pair of parentheses represents a starting point, the last pair of parentheses represents an end point, and the intermediate one or two pairs designate control point(s). The above codes typeset the following round arrows:



The following example shows round arrows placed at the reaction sites of a benzene ring:

```
\sixheterov[bdf]{%
1s=={\red \electronshiftArrowl(0,0)(-70,130)(-140,150)(-200,100)};%
1s==\put(-200,100){\makebox(0,0)[rc]{\red Nu:~}};%
6s=={\red \electronshiftArrowl(0,0)(50,-60)(90,-80)(120,20)}%
}{1==Cl;2==N0$_{2}$;4==N0$_{2}$}
```



The pairs of xy-coordinates for the first $\ensuremath{\ensuremath{\mathsf{velectronshiftArrowl}}\xspace$ command are designated inversely, so that the arrow head of the first arrow appears to be irregular. However, the behavior obeys a rule, because the suffix 1 indicates that an arrow head appears at the position of the first pair of xy-coordinates among the pairs of xy-coordinates, e.g., (0,0) among (0,0)(-70,130)(-140,150)(-200,100). On a similar line, the arrow head of the second arrow drawn by \electronshiftArrowl appears at the position (0,0) among (0,0)(50,-60)(90,-80)(120,20).

On the other hand, the suffix r of \electronshiftArrowr indicates that an arrow head appears at the position of the last pair of xy-coordinates among the pairs of xy-coordinates, e.g., (-200,100) among (0,0)(-70,130)(-140,150)(-200,100).

It should be emphasized that the origin of each arrow (having no dimensions) is always located at the position (0,0), so that the origin (0,0) of the arrow is superposed on a substitution position designated

in a structural formula. Note that the origin of an arrow is not always identical with the starting point (or the end point) of the arrow. The following examples illustrate how the origin of an arrow is placed at a substitution position, where such a substitution position is tentatively given as a circle (or as a double circle) by the **\put** command in the LATEX picture environment.

```
\begin{picture}(1000,300)(-200,0)
%origins of arrows (0,0) on (0,0) or (0,150)
\put(0,0){\circle{40}}
\put(0,0){\electronshiftArrowr(0,0)(-70,130)(-140,150)(-200,100)}
\put(0,150){\circle{40}}
\put(0,150){\electronshiftArrowl(0,0)(-70,130)(-140,150)(-200,100)}
%origins of arrows (0,0) on (300,0) or (300,150)
\put(300,0){\circle{40}}\put(300,0){\circle*{20}}
\put(300,0){\electronshiftArrowr(-200,100)(-140,150)(-70,130)(0,0)}
\put(300,150){\circle{40}}\put(300,150){\circle*{20}}
\put(300,150){\electronshiftArrowl(-200,100)(-140,150)(-70,130)(0,0)}
%origins of arrows (0,0) on (800,0) or (800,150)
\put(800,0){\circle{50}}\put(800,0){\circle{30}}
\put(800,0){\electronshiftArrowr(-300,100)(-240,150)(-170,130)(-100,0)}
\put(800,150){\circle{50}}\put(800,150){\circle{30}}
\put(800,150){\electronshiftArrowl(-300,100)(-240,150)(-170,130)(-100,0)}
\end{picture}
```



On the other hand, one-electron migration is represented by a round arrow with a harpoon-type arrow head:

```
\electronshiftHru(Starting Point)(Control Points)(End Point)
\electronshiftHrd(Starting Point)(Control Points)(End Point)
\electronshiftHlu(Starting Point)(Control Points)(End Point)
\electronshiftHld(Starting Point)(Control Points)(End Point)
```

where the formats of arguments are designed to be the same as those of **\electronshiftArrowr** and **\electronshiftArrowl** described above.

The following examples shows harpoons drawn by these commands:

\electronshiftHru(0,0)(100,100)(200,100)(300,0) \qquad\qquad \electronshiftHrd(0,0)(100,-100)(200,-100)(300,0) \qquad\qquad \electronshiftHlu(0,0)(100,100)(200,100)(300,0) \qquad\qquad \electronshiftHld(0,0)(100,-100)(200,-100)(300,0)



7.6.3 Notes on Control Points

The autocalculation of control points for **\electronrshiftarrow** etc. (Subsection 7.6.1) is based on a simple model as follows:



where two circles represent control points. For example, when the starting point (0,0) and the end point (400,0) are given for an arrow, two control points are calculated to be (100,200) and (300,200). Hence, \electronrshiftarrow (Subsection 7.6.1) and \electronshiftArrowr (Subsection 7.6.2) give equivalent results, when the following codes are written:

\electronrshiftarrow(0,0)(400,0)
\hskip50pt
\electronshiftArrowr(0,0)(100,200)(300,200)(400,0)

Thereby, we obtain the following arrows:

Chapter 8

EPS Files Containing X⁴MT_EX Formulas

8.1 LATEX Documents with XIMTEX Codes

As shown in Section 2.2, a $\square T_E X$ document with $\hat{X}^{1}MT_E X$ codes can be directly processed by the PDF-compatible mode. For example, the $\hat{X}^{1}MT_E X$ code placed in the present $\square T_E X$ document (named xymtx500.tex as a tex file):

```
\fbox{%
\begin{XyMcompd}(300,900)(250,0){}{}
\bzdrv{1==0H;4==0H}
\end{XyMcompd}
}
```

produces the following structural formula:



where the bounding box of the formula is specified by using the XyMcompd environment defined in the chemist package.

Another route for incorporating $\hat{X}^{T}MT_{E}X$ formulas is to use $E^{T}EX$ documents with EPS files of $\hat{X}^{T}MT_{E}X$ structures produced by separate procedures. The present chapter is devoted to describe know-how of making and manipulating EPS files.

8.2 Incorporation of X²MT_EX Formulas as EPS Files

8.2.1 Making a Single-Page PostScriptFile with X¹MT_EX Formulas

As shown in Section 2.3, the option "dvips" of the xymtexpdf package (loaded by means of the declaration: \usepackage[dvips]{xymtexpdf}) results in the generation of a PostScript file, which can be converted

into the corresponding EPS file. For the purpose of accomplishing smooth conversion, a PostScript file should be of a single page, which contains a set of $\hat{X}^{2}MT_{E}X$ structures to be bundled. For example, the following tex file (named eps-test1.tex) shows a typical format for producing a single-page PostScript file.

```
%eps-test1.tex
\documentclass{article}
\usepackage[dvips]{xymtexpdf}
\usepackage{chmst-pdf}
\pagestyle{empty}
\begin{document}
\begin{XyMcompd}(300,900)(250,0){}{}
\bzdrv{1==0H;4==0H}
\end{XyMcompd}
\end{document}
```

Note that the declaration \pagestyle{empty} is important to obtain a correct bounding box. In addition, the declaration \usepackage[dvips]{xymtexpdf} contains an optional argument [dvips].

According to the procedures of Section 2.3, the elaTEX processing of the tex file (eps-test1.tex):

elatex eps-test1

and the subsequent processing by dvips:

dvips eps-test1

produce the corresponding PostScript file (eps-test1.ps), which can be browsed by GSview coupled with Ghostscript.

8.2.2 EPS Files with Correct Bounding Boxes

Conversion of PostScript Files to EPS Files

In order to convert PostScript file (eps-test1.ps) into an EPS file, the following procedure is typical:

- 1. The PostScript file (eps-test1.ps) is browsed by GSview (coupled with Ghostscript), where the button "File" is clicked to show its menu (Figure 8.1).
- 2. The button "PS to EPS" in the menu (Figure 8.1) is selected to result in the appearance of a confirmation box (Figure 8.2), in which the button "yes" is clicked after checking "automatically calculated Bounding Box".
- 3. Thereby, there appears an input box (Figure 8.3), from which the resulting eps file is stored after inputting an appropriate name (here "eps-test1Figr.eps"). Note that the "保存" button in Japanese means "store".
- 4. The resulting file (eps-test1Figr.eps) can be browsed by GSview as shown in Figure 8.4, where the $\hat{X}^{1}MT_{E}X$ structure is surrounded by a dotted frame showing the corresponding bounding box.

Conversion of DVI Files to EPS Files

The dvips converter can produce EPS files directly from dvi files when the option -E is added to the command line, e.g.,

dvipsk -E -D2400 -Pdl -p1 -n1 eps-test1.dvi -o eps-test1FigA.eps



Figure 8.1: Making EPS files of structural formulas. "EP to EPS" button



Figure 8.2: Making EPS files of structural formulas. Confirmation box

	²⁰ MIFES for Windows Version 6.0 - [C:¥fujita0¥xymtex-pdf¥xy ■ □ ▼ □ アイル(D) 編集(D) 検索・置換・ジャンプ(S) ウィンドウ(M) マクロ(M) ② eps-test1.ps - GS ハレブ(M)	
OpenOffice.org 8.0	ウイントウ eps-test1.tex 選択 正 Elle Edit Options Vir こ こ こ こ こ 二 こ	ew Orientation mount Holp
名前を付けて保 保存する場所) 📄 epstest 💽 🔶 🖆 📰 -	N OH
最近使ったファイ、 ほ近使ったファイ、	∭eps-test1Fig.eps	
الرلاية الرلاية الرلاية		ОН
マイ コンピュー5	ファイル名(1): eps-test1Figr.eps (保存(5)) ファイルの種類(1): EPS Files (*.eps) キャンセル	79,744pt Page:"1" 1 off
)) (./eps-test1.aux) [1] (./eps-test1.aux)) Output written on eps-test1.dvi (1 page, 8700 bytes Transcript written on eps-test1.log.	J dchem' on input line 959.

Figure 8.3: Making EPS files of structural formulas. Naming and storing an EPS file



Figure 8.4: Making EPS files of structural formulas. EPS file with a bounding box

However, the resulting EPS files sometimes exhibit incorrect bounding boxes.

To calculate correct bounding boxes, the resulting EPS files (e.g., eps-test1FigA.eps) are opened by GSview and processed according to Figures 8.1 to 8.4. Thereby, new EPS files with a recalculated bounding box are generated.

8.2.3 Incorporation of EPS Files in LaTEX Documents

The following tex file (eps-test2.tex) is prepared to incorporate the file (eps-test1Figr.eps) described above. Because the command \usepackage{xymtexpdf} has no optional argument, the resulting dvi file is processed by the dvipdfm(x) converter so as to generate the corresponding PDF file (eps-test2.pdf) directly. The command \includegraphics is supported by the graphicx package. Note that the xymtexpdf package automatically loads the pgf package, which in turn loads the graphicx package automatically.

```
%eps-test2.tex
\documentclass{article}
\usepackage{xymtexpdf}
\usepackage{chmst-pdf}
%\pagestyle{empty}
\begin{document}

\fbox{%
\includegraphics{eps-test1Figr.eps}
```

```
\end{document}
```

}

To simulate the processing of eps-test2.tex, the same code for the $\hat{X}^{2}MT_{E}X$ structure is written here in the present document:

```
\fbox{%
\includegraphics{eps-test1Figr.eps}
}
```

which generates the following structural formula:



The bounding box of the EPS file (eps-test1Figr.eps) can be checked by an appropriate editor, where it contains the following line:

%%BoundingBox: 151 627 188 715

where the set of four integers represents a boundary box as a rectangle. Thus, (151 627) represents the xy-coordinates of the left bottom point of the bounding box, while (188 715) represents the xy-coordinates of the right upper point.

The bounding box can be changed directly by replacing these four integers by others, because the EPS file can be edited freely by the editor. The bounding box can alternatively be changed by setting up an optional argument [bb=...] of \includegraphics as follows:

```
\fbox{%
\includegraphics[bb=140 627 200 715]{eps-test1Figr.eps}
}
```

This change gives a figure with a slightly wider bonding box.



Chapter 9

Mathversion Utilities

All of the functions of the chmst-ps package (on-line manual: xymtx405406B.pdf) are also supported by the chmst-pdf package. A part of the chmst-pdf package has been already discussed in Chapter 7. This section is a rewritten version of mathversion utilities described in Section 1.2 of the on-line manual ("Use of chemist and chmst-ps Packages" in xymtx405406B.pdf), where the previous chmst-ps package is replaced by the present chmst-pdf package.

9.1 Mathversions of the chmst-pdf Packages

9.1.1 chemist, chmst-ps, and chmst-pdf

According to the three modes of the $\hat{X}^{2}MT_{E}X$ system, there are three sets of mathversion utilities. The following templates are ready to switch the three sets.

1. **T_EX/IAT_EX mode**: The chemist package is read by a command \usepackage declared in the preamble of a tex file, as shown in the following template:

```
\documentclass{article}
\usepackage{xymtex}
\usepackage{chemist}
\begin{document}
(text)%default (normal)
\mathversion{bold}
(text)
\mathversion{chem}
(text)
\mathversion{boldchem}
(text)
\mathversion{normal}
```

(text)%return to the default \end{document}

2. PostScript-compatible mode: Alternatively, the chmst-ps package is read to meet PostScript requirements, where the chemist package is automatically loaded. See the on-line manual of

 $\hat{X}MT_EX$ versions 4.05 and 4.06 (xymtx405406B.pdf). Note that the xymtexps package provides the PostScript-compatible mode of the $\hat{X}MT_EX$ system.

```
\documentclass{article}
\usepackage{xymtexps}
\usepackage{chmst-ps}%%\usepackage{chemist,chmst-ps}
\begin{document}
(text)%default (normal)
\mathversion{bold}
(text)
\mathversion{chem}
(text)
\mathversion{chem}
(text)
\mathversion{boldchem}
(text)
\mathversion{normal}
(text)%return to the default
\end{document}
```

3. **PDF-compatible mode**: In the present extension, the chmst-pdf package is read to meet PDF requirements, where the chemist package is automatically loaded.

```
\documentclass{article}
\usepackage{xymtexpdf}
\usepackage{chmst-pdf}%%\usepackage{chemist,chmst-pdf}
\begin{document}
(text)%default (normal)
\mathversion{bold}
(text)
\mathversion{chem}
(text)
\mathversion{chem}
(text)
\mathversion{boldchem}
(text)
\mathversion{normal}
(text)%return to the default
\end{document}
```

The xymtexpdf package provides the PDF-compatible mode of the $\hat{X}MTEX$ system. Hence, the resulting dvi file should be converted into a PDF file by means of an appropriate converter (e.g., dvipdfm(x)). The resulting PDF file can be browsed by Adobe Reader. See Section 2.2. For an alternative browsing method, see Section 2.3.

9.2 New Mathverstions

There are two mathversions ("normal" and "bold") for mathematical usage in $\text{LAT}_{EX} 2_{\varepsilon}$. The chemist (chmst-pdf or chmst-ps) package provides additional two mathversions ("chem" and "boldchem") for

chemical usage. The latter two mathversions have already been introduced in Chapter 17 of the manual of $\hat{X}MT_{FX}$ version 1.01 (xymtex.pdf).

If a mathversion command is not explicitly declared, the mathversion "normal" is effective so as to provide usual (default) typesetting inherent in the math mode of $\operatorname{IATEX} 2_{\varepsilon}$, where letters etc. are typeset by using *italic* fonts. When \mathversion{bold} command is declared, the $\operatorname{IATEX} 2_{\varepsilon}$ typesetting is conducted under the mathversion "bold", where letters etc. are typeset by using *boldfaced italic* fonts. The chemist package (chmst-pdf or chmst-ps) provides us with two additional mathversions, i.e., "chem" and "boldchem", where letters etc. are typeset by using usual or boldfaced upright fonts, which mainly aim at the output of chemical elements or compounds (such as H_2 and H_2O).

This section is devoted to add further comments with examples, where the chemical environments discussed in Chapter 7 are tested under the respective mathversions.

9.2.1 Mathversion "normal"

The mathversion "normal" gives outputs of default mode, which are inherent in $\text{LATEX } 2_{\varepsilon}$ without any declaration or with declaring \mathversion{normal}.

Default Outputs

To show such standard outputs, the listing command \testmathversion (Output A) is defined as follows:

```
\def\testmathversion{%for Output A
\[abcdefghijklmnopqrstuvwxyz\imath \jmath ABCDEFGHIJKLMNOPQRSTUVWXYZ\]
\[\alpha\beta\gamma\delta\epsilon\zeta\theta\iota\kappa\lambda
\mu\nu\xi\pi\rho\sigma\tau\upsilon\phi\chi\psi\omega
\varepsilon\vartheta\varpi\varrho\varsigma\varphi
\Gamma\Delta\Theta\Lambda\Xi\Pi\Sigma\Upsilon\Phi\Psi\Omega\]
\[1234567890\]
\[\mathnormal{1234567890}\]
\[\mathcal{ABCDEFGHIJKLMNOPQRSTUVWXYZ}\]
\[\int \sum \prod \coprod \bigcup \bigcap
\bigodot \bigoplus \biguplus \bigotimes \]
((, ), [, ], ?, !, (, ), =, > (mathgreater), < (mathless),
\leftharpoonup, \leftharpoondown, \rightharpoonup, \rightharpoondown,
\ell, \wp, \partial, \flat, \natural, \sharp, \triangleleft, \triangleright,
\smile, \frown, \star\]
\left( \left( \left( x \right), \right), \left( x \right), 
\x \in x, \x \in x, \dot{x}, \bar{x},
\tilde{x}, \tilde{x}
```

Thereby, the following code using \testmathversion:

```
{\def\tboxtitle{\bf Output A due to ''normal''}
\begin{tboxscreen}
\testmathversion
\end{tboxscreen}}
```

is described in a tboxtitle environment (supported by the chemist package) so as to produce:

To test equation and equarray environments, the command \testequation (Output B) is defined as follows:

Thereby, the output produced by **\testequation** shows default outputs of equation and eqnarray environments in the present mathversion "normal" as follows:

Output B due to "normal" Euler's summation:

$$\sum_{a \le k < b} f(k) = \int_{a}^{b} f(x) dx + \sum_{k=1}^{m} \frac{B_{k}}{k!} f^{(k-1)}(x) \Big|_{a}^{b} + R_{m}.$$
(9.1)

The term R_m is represented as follows:

$$R_m = (-1)^{m+1} \int_a^b \frac{B_m(\{x\})}{m!} f^{(m)}(x) dx,$$

 $a \le b \text{ and } m \ge 1,$
(9.2)

where the symbols a, b, and m represent integers.

Convenient Environments for Chemical Equations

To test equation and eqnarray environments containing chemical formulas, the following test command \testequationforchemistry (Output C) is defined as follows:

```
\def\testequationforchemistry{%%for Output C
An equation enviroment:
\begin{equation}
2H_2 + 0_2 \rightarrow 2H_20
\end{equation}
An eqnarray environment:
\begin{eqnarray}
C + 0_2 & \rightarrow & CO_2 \\
Na^{+} + Cl^{-} & \rightarrow & NaCl\downarrow
\end{eqnarray}
}%
```

Under the mathversion "normal", alphabets in a math mode (such as an equation or equarray environment) are typeset by using italic fonts, which do not meet chemical requirements. Thus the test command \testequationforchemistry defined above gives the following output.

Output C due to "normal"		
An equation environment:		
	$2H_2 + O_2 \rightarrow 2H_2O$	(9.3)
An equarray environment:		
	$C + O_2 \rightarrow CO_2$	(9.4)
	$Na^+ + Cl^- \rightarrow NaCl \downarrow$	(9.5)
)

The chemist (chmst-ps or chmst-pdf) package defines chemeqn and chemeqnarray environments in order to support chemical requirements. (cf. Subsection 17.1 of the manual of $\hat{X}^{2}MTEX$ version 1.01 (xymtex.pdf)). The following \testchemequation command (Output D) is defined to test the functions of the chemeqn and chemeqnarray environments and related commands.

```
\def\testchemequation{%%for Output D
A chemegn enviroment:
\begin{chemeqn}
2H_2 + 0_2 \rightarrow 2H_20
\end{chemeqn}
\begin{chemeqn}
abcdefghijklmnopqrstuvwxyz\imath \jmath ABCDEFGHIJKLMNOPQRSTUVWXYZ
\end{chemeqn}
\begin{chemeqn}
\alpha\beta\gamma\delta\epsilon\zeta\eta\theta\iota\kappa\lambda
\mu\nu\xi\pi\rho\sigma\tau\upsilon\phi\chi\psi\omega
\varepsilon\vartheta\varpi\varrho\varsigma\varphi
\Gamma\Delta\Theta\Lambda\Xi\Pi\Sigma\Upsilon\Phi\Psi\Omega
\end{chemegn}
A chemegnarray environment:
\begin{chemeqnarray}
C + 0_2 & \rightarrow & CO_2 \\
Na<sup>{+}</sup> + Cl<sup>{-}</sup> & \rightarrow & NaCl\downarrow
```

```
\end{chemeqnarray}
A chemeqnarray$*$ environment:
\begin{chemeqnarray*}
C + 0_2 & \rightarrow & CO_2 \\
Na^{+} + Cl^{-} & \rightarrow & NaCl\downarrow
\end{chemeqnarray*}
In-text chemical formulas: \chemform{2H_2 + 0_2 \rightarrow 2H_20}
and \chemform{C + 0_2 \rightarrow CO_2}
}%
```

Under the mathversion "normal", alphabets in a chemeqn environment etc. are typeset by using upright fonts. Thus the test command **\testchemequation** defined above gives the following output.

Output D due to "normal"

A chemeqn environment:

$2\mathrm{H}_2 + \mathrm{O}_2 \rightarrow 2\mathrm{H}_2\mathrm{O}$	(9.6)
$abcdefghijklmnopqrstuvwxyz\imath\jmath ABCDEFGHIJKLMNOPQRSTUVWXYZ$	(9.7)
αβγδεζηθικλμνξπρστυφχψωεθϖ <u>ρ</u> ςφΓΔΘΛΞΠΣΥΦΨΩ	(9.8)
A chemequarray environment:	
$C + O_2 \rightarrow CO_2$	(9.9)

$$Na^{+} + Cl^{-} \rightarrow NaCl \downarrow$$
(9.10)

A chemequarray* environment:

$$\begin{array}{rcl} {\rm C} + {\rm O}_2 & \rightarrow \ {\rm CO}_2 \\ {\rm Na}^+ + {\rm Cl}^- & \rightarrow \ {\rm NaCl} \downarrow \end{array}$$

In-text chemical formulas: $\rm 2H_2 + O_2 \rightarrow 2H_2O$ and $\rm C + O_2 \rightarrow CO_2$

In addition to the chemeqn and chemeqnarray environments (cf. Subsection 17.1 of the manual of X²MT_EX version 1.01 (xymtex.pdf)), the latest version of the chemist (chmst-ps or chmst-pdf) package provides another set of commands for chemical requirements, i.e., ChemEquation, ChemEqnarray, and ChemEqnarray* environments as well as \ChemForm command. The following \testChemEquation command (Output E) is defined to test the functions of these newly-defined commands.

```
\def\testChemEquation{%%for Output E
A ChemEquation enviroment:
    \begin{ChemEquation}
    2H_2 + 0_2 \rightarrow 2H_20
    \end{ChemEquation}
    \begin{ChemEquation}
    abcdefghijklmnopqrstuvwxyz\imath \jmath ABCDEFGHIJKLMNOPQRSTUVWXYZ
    \end{ChemEquation}
    \begin{ChemEquation}
    \begin{ChemEquation}
    \alpha\beta\gamma\delta\epsilon\zeta\eta\theta\iota\kappa\lambda
    \mu\nu\xi\pi\rho\sigma\tau\upsilon\phi\chi\psi\omega
    \varepsilon\vartheta\varpi\varrho\varsigma\varphi
    \Gamma\Delta\Theta\Lambda\Xi\Pi\Sigma\Upsilon\Phi\Psi\Omega
    \end{ChemEquation}
A ChemEqnarray environment:
```

```
\begin{ChemEqnarray}
C + 0_2 & \rightarrow & CO_2 \\
Na^{+} + Cl^{-} & \rightarrow & NaCl\downarrow
\end{ChemEqnarray}
A ChemEqnarray$*$ environment:
\begin{ChemEqnarray*}
C + 0_2 & \rightarrow & CO_2 \\
Na<sup>{+}</sup> + Cl<sup>{</sup>-} & \rightarrow & NaCl\downarrow
\end{ChemEqnarray*}
In-text chemical formulas: \ChemForm{2H_2 + 0_2 \rightarrow 2H_20}
and \ChemForm{C + 0_2 \rightarrow CO_2}
}%
```

Under the mathversion "normal", alphabets in a ChemEquation etc. are typeset also by using upright fonts. Thus the test command \testChemEquation defined above gives the following output.

Output E due to "normal"	
A ChemEquation environment:	
$\rm 2H_2 + O_2 \rightarrow 2H_2O$	(9.11)
$abcdefghijklmnopqrstuvwxyz\imath\jmath ABCDEFGHIJKLMNOPQRSTUVWXYZ$	(9.12)
αβγδεζηθικλμνξπρστυφχψωεθϖϱςφΓΔΘΛΞΠΣΥΦΨΩ	(9.13)
A ChemEqnarray environment:	
$C + O_2 \rightarrow CO_2$	(9.14)
$\begin{array}{rcl} {\rm C} + {\rm O}_2 & \rightarrow & {\rm CO}_2 \\ {\rm Na}^+ + {\rm Cl}^- & \rightarrow & {\rm NaCl} \downarrow \end{array}$	(9.15)
A ChemEquarrav* environment:	

$$\begin{array}{rcl} {\rm C} + {\rm O}_2 & \rightarrow & {\rm CO}_2 \\ {\rm Na}^+ + {\rm Cl}^- & \rightarrow & {\rm NaCl} \downarrow \end{array}$$

In-text chemical formulas: $2H_2+O_2\rightarrow 2H_2O$ and $C+O_2\rightarrow CO_2$

Mathversion "bold" 9.2.2

The mathversion "bold" gives outputs of boldfaced fonts, when the switching command \mathversion is explicitly declared. An alternative (rather old) method to enter the mathversion "bold" is the declaration of \boldmath. For example, {\boldmath $x_{i} > 0$ produces x_i according to $AT_FX2.09$. This section is typeset after the declaration of

\mathversion{bold}

according to $IAT_FX 2_{\mathcal{E}}$.

Outputs under Mathversion "bold"

The mathversion "bold" gives outputs of "bold" mode, which are inherent in $IAT_EX 2_{\varepsilon}$. To show such outputs, the listing command \testmathversion defined above is used after the declaration of \mathversion{bold}. The result is shown as follows:

Output A due to "bold"

To test equation and eqnarray environments under the mathversion "bold", the above-defined command **\testequation** is again used here so as to give the following output:

Output B due to "bold"

Euler's summation:

$$\sum_{a \le k < b} f(k) = \int_{a}^{b} f(x) dx + \sum_{k=1}^{m} \frac{B_{k}}{k!} f^{(k-1)}(x) \Big|_{a}^{b} + R_{m}.$$
(9.16)

The term R_m is represented as follows:

$$R_{m} = (-1)^{m+1} \int_{a}^{b} \frac{B_{m}(\{x\})}{m!} f^{(m)}(x) dx,$$

$$a \le b \text{ and } m \ge 1,$$
(9.17)

where the symbols a, b, and m represent integers.

Environments and Commands for Chemistry

Under the mathversion "bold", alphabets in a math mode (such as an equation or equarray environment) are typeset by using boldfaced italic fonts. Thus the test command **\testequationforchemistry** defined above gives the following output, which does not meet chemical requirements.

Output C due to "bold"		
An equation environment:		
	$2H_2+O_2 ightarrow 2H_2O$	(9.18)
An equarray environment:		
	$C+O_2 \ ightarrow \ CO_2$	(9.19)
	$Na^+ + Cl^- \ o \ NaCl \downarrow$	(9.20)
)

Even under the mathversion "bold", alphabets in a chemeqn environment etc. are typeset by using upright fonts. Thus the test command **\testchemequation** defined above gives the following output, which is equivalent to the above output of the mathversion "normal".

Output D due to "bold"

A chemeqn environment: $2H_2 + O_2 \rightarrow 2H_2O \qquad (9.21)$ abcdefghijklmnopqrstuvwxyzıjABCDEFGHIJKLMNOPQRSTUVWXYZ (9.22) $\alpha\beta\gamma\delta\epsilon\zeta\eta\theta\iota\kappa\lambda\mu\nu\xi\pi\rho\sigma\tau\nu\phi\chi\psi\omega\varepsilon\vartheta\varpi\varrho\varsigma\varphi\Gamma\Delta\Theta\Lambda\Xi\Pi\Sigma\Upsilon\Phi\Psi\Omega \qquad (9.23)$ A chemeqnarray environment: $C + O_2 \rightarrow CO_2 \qquad (9.24)$ $Na^+ + Cl^- \rightarrow NaCl \downarrow \qquad (9.25)$

A chemeqnarray* environment:

 $\begin{array}{rcl} {\rm C} + {\rm O}_2 & \rightarrow \ {\rm CO}_2 \\ {\rm Na}^+ + {\rm Cl}^- & \rightarrow \ {\rm NaCl} \downarrow \end{array}$

In-text chemical formulas: $2H_2+O_2\rightarrow 2H_2O$ and $C+O_2\rightarrow CO_2$

In contrast, ChemEquation, ChemEqnarray, and ChemEqnarray* environments as well as a \ChemForm command typeset boldfaced alphabets of upright shape under the mathversion "bold". Thus the test command \testChemEquation defined above gives the following output, which is different from the corresponding output of the mathversion "normal".

Output E due to "bold"

A ChemEquation environment: $2H_2 + O_2 \rightarrow 2H_2O$ (9.26)abcdefghijklmnopqrstuvwxyzıjABCDEFGHIJKLMNOPQRSTUVWXYZ(9.27) $\alpha\beta\gamma\delta\epsilon\zeta\eta\theta\iota\kappa\lambda\mu\nu\xi\pi\rho\sigma\tau\nu\phi\chi\psi\omega\epsilon\vartheta\varpi\varrho\varsigma\varphi\Gamma\Delta\Theta\Lambda\Xi\Pi\Sigma\Upsilon\Phi\Psi\Omega$ (9.28)A ChemEqnarray environment: $C + O_2 \rightarrow CO_2$ (9.29) $Na^+ + Cl^- \rightarrow NaCl \downarrow$ (9.30)A ChemEqnarray* environment: $C + O_2 \rightarrow CO_2$
 $Na^+ + Cl^- \rightarrow NaCl \downarrow$

In-text chemical formulas: $\mathbf{2H_2} + \mathbf{O_2} \rightarrow \mathbf{2H_2O}$ and $\mathbf{C} + \mathbf{O_2} \rightarrow \mathbf{CO_2}$

9.2.3 Mathversion "chem"

Subsection 17.1 of the manual of $\hat{X}^{1}MT_{E}X$ version 1.01 (xymtex.pdf) has discussed the original version of the mathversion "chem", which involved some irregular outputs of letters. The latest version of the chemist (chmst-ps or chmst-pdf) package gives more sufficient results with respect to letter outputs.

This section is typeset after the declaration of

\mathversion{chem}

Outputs under Mathversion "chem"

The mathversion "chem" gives outputs of "chem" mode, which aim at upright letters for chemical formulas. To show such outputs, the listing command \testmathversion defined above is used after the declaration of \mathversion{chem}. The result is shown as follows:

Output A due to "chem"

$abcdefghijklmnopqrstuvwxyz\imath\jmath ABCDEFGHIJKLMNOPQRSTUVWXYZ$
$lphaeta\gamma\delta\epsilon\zeta\eta heta\iota\kappa\lambda\mu u\xi\pi ho\sigma au u\phi\chi\psi\omegaarepsilonarpiarpiarpiarpiarpiarpiarpiarpi$
1234567890
1234567890
ABCDEFGHIJKLMNOPQRSTUVWXYZ
$\int \Sigma \Pi \amalg \cup \cap \odot \oplus $
$(,),[,],?,!,\{,\},=, {:}(>), {:}(<), {\leftarrow}, {\rightarrow}, {\rightarrow}, \ell, \wp, \partial, \flat, \natural, \sharp, \triangleleft, \triangleright, \smile, \frown, \star$
$\check{\mathrm{x}},\check{\mathrm{x}},\dot{\mathrm{x}},\dot{\mathrm{x}},\dot{\mathrm{x}},\dot{\mathrm{x}},\ddot{\mathrm{x}},\bar{\mathrm{x}},\tilde{\mathrm{x}},\hat{\mathrm{x}},\hat{\mathrm{x}},\hat{\mathrm{x}}$

As found in the first line of Output A due to "chem", lowercase and uppercase alphabets are typeset upright except i and j.

Note that the symbols < and > are not properly typeset if they are input directly. The commands \$\mathless\$ and \$\mathgreater\$ should be used to give correct printing.

To test equation and equarray environments under the mathversion "chem", the above-defined command **\testequation** is again used here, although the resulting output is contrary to mathematical conventions:

Output B due to "chem"

Euler's summation:

$$\sum_{a \le k_i b} f(k) = \int_a^b f(x) dx + \sum_{k=1}^m \frac{B_k}{k!} f^{(k-1)}(x) \Big|_a^b + R_m.$$
(9.31)

The term R_m is represented as follows:

$$\begin{aligned} R_m &= (-1)^{m+1} \int_a^b \frac{B_m(\{x\})}{m!} f^{(m)}(x) dx, \\ &a \le b \text{ and } m \ge 1, \end{aligned} \tag{9.32}$$

where the symbols a, b, and m represent integers.

Note that the symbols < and > are not properly typeset if they are input directly. Thus, the symbol < in the lower limit of the above summation is erroneously replaced by the symbol ;. The commands \$\mathless\$ and \$\mathless\$ and \$\mathless\$ bould be used to give correct printing. For example, the code:

\[\sum_{a\leq k \mbox{\scriptsize \$\mathless\$} b}f(k)\]

gives the following output:

$$\sum_{a \leq k < b} f(k)$$

Because this output does not meet mathematical conventions, it should be written as follows:

{\mathversion{normal}

which gives the following output:

$$\sum_{a \le k < b} f(k)$$

Environments and Commands for Chemistry

Under the mathversion "chem", alphabets in a math mode (such as an equation or eqnarray environment) are typeset by using upright fonts. Thus the test command \testequationforchemistry defined above gives the following output, which meets chemical requirements.

Output C due to "chem"		
An equation environment:		
	$2H_2 + O_2 \rightarrow 2H_2O$	(9.33)
An equarray environment:		
	$C + O_2 \rightarrow CO_2$	(9.34)
	$Na^+ + Cl^- \rightarrow NaCl \downarrow$	(9.35)
<)

Under the mathversion "chem", alphabets in a **chemeqn** environment etc. are also typeset by using upright fonts. Thus the test command **\testchemequation** defined above gives the following output, which is equivalent to the above output of the mathversion "normal".

 $\begin{array}{c|c} \hline \textbf{Output D due to "chem"} \\ \hline \textbf{A chemeqn environment:} \\ & 2H_2 + O_2 \rightarrow 2H_2 \textbf{O} \\ & abcdefghijklmnopqrstuvwxyzij\textbf{ABCDEFGHIJKLMNOPQRSTUVWXYZ} \\ & \alpha\beta\gamma\delta\epsilon\zeta\eta\theta\iota\kappa\lambda\mu\nu\xi\pi\rho\sigma\tau\nu\phi\chi\psi\omega\varepsilon\vartheta\varpi\varrho\varsigma\varphi\Gamma\Delta\Theta\Lambda\Xi\Pi\Sigma\Upsilon\Phi\Psi\Omega \end{array}$ (9.38)

A chemequarray environment:

$$C + O_2 \rightarrow CO_2$$
 (9.39)

$$Na^+ + Cl^- \rightarrow NaCl \downarrow$$
 (9.40)

A chemeqnarray* environment:

$$\begin{array}{rcl} {\rm C} + {\rm O}_2 & \rightarrow \ {\rm CO}_2 \\ {\rm Na}^+ + {\rm Cl}^- & \rightarrow \ {\rm NaCl} \downarrow \end{array}$$

In-text chemical formulas: $\rm 2H_2 + O_2 \rightarrow 2H_2O$ and $\rm C + O_2 \rightarrow CO_2$

On the same line, ChemEquation, ChemEqnarray, and ChemEqnarray* environments as well as a \ChemForm command typeset alphabets of upright shape under the mathversion "chem". Thus the test command \testChemEquation defined above gives the following output, which is different from the corresponding output of the mathversion "bold" but equivalent to the corresponding output of the mathversion "normal".

Output E due to "chem"

A ChemEquation environment:

$$2H_2 + O_2 \rightarrow 2H_2O \tag{9.41}$$

abcdefghijklmnopqrstuvwxyzijABCDEFGHIJKLMNOPQRSTUVWXYZ (9.42)

 $\alpha\beta\gamma\delta\epsilon\zeta\eta\theta\iota\kappa\lambda\mu\nu\xi\pi\rho\sigma\tau\upsilon\phi\chi\psi\omega\varepsilon\vartheta\varpi\varrho\varsigma\varphi\Gamma\Delta\Theta\Lambda\Xi\Pi\Sigma\Upsilon\Phi\Psi\Omega\tag{9.43}$

A ChemEquarray environment:

$$C + O_2 \rightarrow CO_2 \tag{9.44}$$

$$Na^+ + Cl^- \rightarrow NaCl \downarrow$$
 (9.45)

A ChemEqnarray* environment:

 $\begin{array}{rcl} {\rm C} + {\rm O}_2 & \rightarrow & {\rm CO}_2 \\ {\rm Na}^+ + {\rm Cl}^- & \rightarrow & {\rm NaCl} \downarrow \end{array}$

In-text chemical formulas: $\rm 2H_2 + O_2 \rightarrow 2H_2O$ and $\rm C + O_2 \rightarrow CO_2$

9.2.4 Mathversion "boldchem"

In addition to the mathversion "chem" supported by the original version of the chemist (chmst-ps or chmst-pdf) package (cf. Subsection 17.1 of the manual of $\hat{X}^{2}MT_{E}X$ version 1.01 (xymtex.pdf)), the latest version of the chemist package packed in $\hat{X}^{2}MT_{E}X$ version 4.05 supports the mathversion "boldchem".

This section is typeset after the declaration of

\mathversion{boldchem}

Outputs under Mathversion "boldchem"

The mathversion "boldchem" gives outputs of "boldchem" mode, which aim at upright letters for chemical formulas. To show such outputs, the listing command \testmathversion defined above is used after the declaration of \mathversion{boldchem}. The result is shown as follows:

As found in the first line of Output A due to "boldchem", lowercase and uppercase alphabets are typeset in boldfaced upright fonts except i and j.

Note that the symbols < and > are not properly typeset if they are input directly. The commands \$\mathless\$ and \$\mathgreater\$ should be used to give correct printing. To test equation and eqnarray environments under the mathversion "boldchem", the above-defined command \testequation is again used here, although the resulting output is contrary to mathematical conventions:

Output B due to "boldchem"

Euler's summation:

$$\sum_{a \le k; b} f(k) = \int_{a}^{b} f(x) dx + \sum_{k=1}^{m} \frac{B_{k}}{k!} f^{(k-1)}(x) \Big|_{a}^{b} + R_{m}.$$
(9.46)

The term \boldsymbol{R}_m is represented as follows:

$$\begin{aligned} \mathbf{R}_{\mathbf{m}} &= (-1)^{\mathbf{m}+1} \int_{\mathbf{a}}^{\mathbf{b}} \frac{\mathbf{B}_{\mathbf{m}}(\{\mathbf{x}\})}{\mathbf{m}!} \mathbf{f}^{(\mathbf{m})}(\mathbf{x}) \mathbf{d}\mathbf{x}, \\ &\mathbf{a} \leq \mathbf{b} \text{ and } \mathbf{m} \geq \mathbf{1}, \end{aligned} \tag{9.47}$$

where the symbols **a**, **b**, and **m** represent integers.

Note that the symbols < and > are not properly typeset if they are input directly. Thus, the symbol < in the lower limit of the above summation is erroneously replaced by the symbol ;. The commands \$\mathless\$ and \$\mathless\$ and \$\mathless\$ bould be used to give correct printing. For example, the code:

\[\sum_{a\leq k \mbox{\scriptsize \$\mathless\$} b}f(k)\]

gives the following output:

$$\sum_{a \leq k < b} f(k)$$

Because this output does not meet mathematical conventions, it should be written as follows:

{\mathversion{bold}
 \[\sum_{a\leq k < b}f(k)\]
}</pre>

which gives the following output:

$$\sum_{a \leq k < b} f(k)$$

Environments and Commands for Chemistry

Under the mathversion "boldchem", alphabets in a math mode (such as an equation or eqnarray environment) are typeset by using upright fonts. Thus the test command \testequationforchemistry defined above gives the following output, which meets chemical requirements.

Output C due to "boldchem"		
An equation environment:		
	$2\mathrm{H}_2 + \mathrm{O}_2 \rightarrow 2\mathrm{H}_2\mathrm{O}$	(9.48)
An equarray environment:		
	$\mathrm{C}+\mathrm{O_2}~\rightarrow~\mathrm{CO_2}$	(9.49)
	$\mathrm{Na^{+}+Cl^{-}} \rightarrow \mathrm{NaCl}\downarrow$	(9.50)
	· · · · · · · · · · · · · · · · · · ·	

Under the mathversion "boldchem", alphabets in a **chemeqn** environment etc. are typeset by using upright fonts (not boldfaced). Thus the test command **\testchemequation** defined above gives the following output, which is equivalent to the above output of the mathversion "chem".

Output D due to "boldchem"

A chemeqn environment: $2H_2 + O_2 \rightarrow 2H_2O$ (9.51)abcdefghijklmnopqrstuvwxyzıjABCDEFGHIJKLMNOPQRSTUVWXYZ(9.52) $\alpha\beta\gamma\delta\epsilon\zeta\eta\theta\iota\kappa\lambda\mu\nu\xi\pi\rho\sigma\tau\nu\phi\chi\psi\omega\varepsilon\vartheta\varpi\rho\varsigma\varphi\Gamma\Delta\Theta\Lambda\Xi\Pi\Sigma\Upsilon\Phi\Psi\Omega$ (9.53)A chemeqnarray environment: $C+O_2 \rightarrow CO_2$ $Na^+ + Cl^- \rightarrow NaCl \downarrow$ (9.55)

A chemeqnarray* environment:

$$\begin{array}{rcl} {\rm C} + {\rm O}_2 & \rightarrow \ {\rm CO}_2 \\ {\rm Na}^+ + {\rm Cl}^- & \rightarrow \ {\rm NaCl} \downarrow \end{array}$$

In-text chemical formulas: $\rm 2H_2 + O_2 \rightarrow 2H_2O$ and $\rm C + O_2 \rightarrow CO_2$

In contrast, ChemEquation, ChemEqnarray, and ChemEqnarray* environments as well as a \ChemForm command typeset boldfaced alphabets of upright shape under the mathversion "boldchem". Thus the test command \testChemEquation defined above gives the following output, which is equivalent to the corresponding output of the mathversion "bold'.

Output E due to "boldcher	n"
---------------------------	----

A ChemEquation environment:	
$2\mathrm{H}_2 + \mathrm{O}_2 \rightarrow 2\mathrm{H}_2\mathrm{O}$	(9.56)
${ m abcdefghijklmnopqrstuvwxyz}\imath\jmath { m ABCDEFGHIJKLMNOPQRSTUVW}$	XYZ (9.57)
$lphaeta\gamma\delta\epsilon\zeta\eta heta\iota\kappa\lambda\mu u\xi\pi ho\sigma au v\phi\chi\psi\omegaarepsilonarpi\omegaarepsilonarpi$	(9.58)
A ChemEqnarray environment:	
$\mathbf{C} + \mathbf{O_2} \rightarrow \mathbf{CO_2}$	(9.59)
$\begin{array}{rcl} \mathrm{C} + \mathrm{O_2} & \rightarrow & \mathrm{CO_2} \\ \mathrm{Na^+} + \mathrm{Cl^-} & \rightarrow & \mathrm{NaCl} \downarrow \end{array}$	(9.60)
A ChemEqnarray* environment:	

$$\begin{array}{rcl} \mathbf{C} + \mathbf{O_2} & \rightarrow & \mathbf{CO_2} \\ \mathbf{Na^+} + \mathbf{Cl^-} & \rightarrow & \mathbf{NaCl} \downarrow \end{array}$$

In-text chemical formulas: $\mathbf{2H_2} + \mathbf{O_2} \rightarrow \mathbf{2H_2O}$ and $\mathbf{C} + \mathbf{O_2} \rightarrow \mathbf{CO_2}$

Chapter 10

Cooperative Use of the amsmath Package

On a similar line to the chmst-ps packages (on-line manual: xymtx405406B.pdf), the chmst-pdf package supports cooperative use of the amsmath package, where new chemical equation environments akin to the standard ChemEquation and ChemEqnarray environments are created to be in harmony with the methodology of the amsmath package. This section is a reproduced version of Section 2.3 of the on-line manual ("Creation of New Environments for Chemical Equations" in xymtx405406B.pdf), where the previous chmst-ps package is replaced by the present chmst-pdf package.

10.1 Creation of New Environments for Chemical Equations

When the **amsmath** package is loaded, several environments for printing multiline display equations are available. These mathematical environments can be converted into chemical versions by using the **\newchemenvironment** command supported by the present chmst-pdf package. ¹

```
format: \newchemenvironment{New Chem Environment}{Original Math Environment}
```

10.1.1 Creation of the chemmultline Environment

The multline environment of the amsmath package provides us with a mathematical tool for folding a long display equation into a multiline display equation in accord with the text width to be set up:

```
\begin{minipage}{0.6\textwidth}
\begin{multline}
Cu^{2+}(aq) + SO_{4}^{2-}(aq) + 2Li^{+}(aq) + 2OH^{-}(aq) \rightarrow \\
Cu(OH)_{2}(s) + 2Li^{+}(aq) + SO_{4}^{2-}(aq)
\end{multline}
{\mathversion{chem}
\begin{multline}
Cu^{2+}(aq) + SO_{4}^{2-}(aq) + 2Li^{+}(aq) + 2OH^{-}(aq) \rightarrow \\
Cu(OH)_{2}(s) + 2Li^{+}(aq) + SO_{4}^{2-}(aq)
\end{multline}
{\end{multline}
}
\end{multline}
```

¹The chemst or chmst-ps package is also effective for this purpose. See the on-line manual (xymtx405406B.pdf).

$$Cu^{2+}(aq) + SO_4^{2-}(aq) + 2Li^+(aq) + 2OH^-(aq) \rightarrow Cu(OH)_2(s) + 2Li^+(aq) + SO_4^{2-}(aq) \quad (10.1)$$

$$Cu^{2+}(aq) + SO_4^{2-}(aq) + 2Li^+(aq) + 2OH^-(aq) \rightarrow$$

 $Cu(OH)_2(s) + 2Li^+(aq) + SO_4^{2-}(aq)$ (10.2)

Note that the minipage environment reduces the text width to emphasize the function of the multiline environment. Each molecular formula in the multiline environment is printed in italic fonts on a similar line to equation and like environments of $IAT_EX 2_{\varepsilon}$. The latter example shows that the declaration of $Mathversion{chem}$ changes the appearance of the multiline environment.

We can create a chemical version of the multline environment of the amsmath package by declaring

\newchemenvironment{chemmultline}{multline}

Thereby, the newly-defined chemmultline environment is substituted for the multline environment shown above so as to give the following result:

```
\newchemenvironment{chemmultline}{multline}
\begin{minipage}{0.6\textwidth}
\begin{chemmultline}
Cu^{2+}(aq) + SO_{4}^{2-}(aq) + 2Li^{+}(aq) + 2OH^{-}(aq) \rightarrow \\
Cu(OH)_{2}(s) + 2Li^{+}(aq) + SO_{4}^{2-}(aq)
\end{chemmultline}
{\mathversion{boldchem}
\begin{chemmultline}
Cu^{2+}(aq) + SO_{4}^{2-}(aq) + 2Li^{+}(aq) + 2OH^{-}(aq) \rightarrow \\
Cu(OH)_{2}(s) + 2Li^{+}(aq) + SO_{4}^{2-}(aq)
\end{chemmultline}
\end{chemmultline}
```

$$Cu^{2+}(aq) + SO_4^{2-}(aq) + 2Li^+(aq) + 2OH^-(aq) \rightarrow$$

 $Cu(OH)_2(s) + 2Li^+(aq) + SO_4^{2-}(aq)$ (10.3)

$$\begin{aligned} \mathrm{Cu}^{2+}(\mathrm{aq}) + \mathrm{SO}_{4}^{2-}(\mathrm{aq}) + 2\mathrm{Li}^{+}(\mathrm{aq}) + 2\mathrm{OH}^{-}(\mathrm{aq}) \rightarrow \\ \mathrm{Cu}(\mathrm{OH})_{2}(\mathrm{s}) + 2\mathrm{Li}^{+}(\mathrm{aq}) + \mathrm{SO}_{4}^{2-}(\mathrm{aq}) \quad (10.4) \end{aligned}$$

The latter example shows the declaration of **\mathversion{boldchem}** changes the appearance of the **chemmultiline** environment.

On a similar line, the multiline* environment of the amsmath package can be converted into a chemical version named chemmultiline*. The newly-defined chemmultiline* environment is used in place of the chemmultiline environment so as to give the following result without printing equation numbers:

```
\newchemenvironment{chemmultline*}{multline*}
\begin{minipage}{0.6\textwidth}
\begin{chemmultline*}
Cu^{2+}(aq) + SO_{4}^{2-}(aq) + 2Li^{+}(aq) + 2OH^{-}(aq) \rightarrow \\
Cu(OH)_{2}(s) + 2Li^{+}(aq) + SO_{4}^{2-}(aq)
\end{chemmultline*}
\end{minipage}
```

$$Cu^{2+}(aq) + SO_4^{2-}(aq) + 2Li^+(aq) + 2OH^-(aq) \rightarrow$$
$$Cu(OH)_2(s) + 2Li^+(aq) + SO_4^{2-}(aq)$$

Environments created by \newchemenvironment have properties equivalent to ChemEquation and like environments, which are originally supported by the chmst-pdf (chmst-ps) package. Hence, fonts used in such newly-defined environments can be changed by declaring \let\ChemEqFont=\sf etc. After the declaration \let\ChemEqFont=\sf, the same code as shown above gives following result:

```
\let\ChemEqFont=\sf
\begin{minipage}{0.6\textwidth}
\begin{chemmultline*}
Cu^{2+}(aq) + SO_{4}^{2-}(aq) + 2Li^{+}(aq) + 2OH^{-}(aq) \rightarrow \\
Cu(OH)_{2}(s) + 2Li^{+}(aq) + SO_{4}^{2-}(aq)
\end{chemmultline*}
\end{minipage}
```

$$\begin{aligned} \mathsf{Cu}^{2+}(\mathsf{aq}) + \mathsf{SO}_4^{2-}(\mathsf{aq}) + 2\mathsf{Li}^+(\mathsf{aq}) + 2\mathsf{OH}^-(\mathsf{aq}) \rightarrow \\ \mathsf{Cu}(\mathsf{OH})_2(\mathsf{s}) + 2\mathsf{Li}^+(\mathsf{aq}) + \mathsf{SO}_4^{2-}(\mathsf{aq}) \end{aligned}$$

10.1.2 Creation of the chemgather Environment

The chemgather environment as a chemical version of the gather environment of the amsmath package can be created on a similar line by using \newchemenvironment. A typical example is shown as follows:

```
\newchemenvironment{chemgather}{gather}
\begin{chemgather}
Bi(NO_{3})_{3}\cdot 5H_{2}O \overset{50-60^{\circ}}{lllongrightarrow}
[Bi_{6}O_{6}]_{2}(NO_{3})_{11}(OH)\cdot 6H_{2}O
\overset{77-130^{\circ}}{lllongrightarrow} \notag \\
[Bi_{6}O_{6}](NO_{3})_{6}(OH)\cdot 3H_{2}O
\overset{400-450^{\circ}}{lllongrightarrow}
\alpha\mbox{-}Bi_{2}O_{3} \\
Bi(NO_{3})_{3}\cdot N_{2}O_{4}
\overset{200^{\circ}}{lllongrightarrow}
Bi_{2}O(NO_{3})_{4}
\overset{415^{\circ}}{lllongrightarrow}
Bi_{4}O_{5}(NO_{3})_{2}
\end{chemgather}
```

$$\operatorname{Bi}(\operatorname{NO}_3)_3 \cdot 5\operatorname{H}_2\operatorname{O} \xrightarrow{50-60^{\circ}} [\operatorname{Bi}_6\operatorname{O}_6]_2(\operatorname{NO}_3)_{11}(\operatorname{OH}) \cdot 6\operatorname{H}_2\operatorname{O} \xrightarrow{77-130^{\circ}} \\ [\operatorname{Bi}_6\operatorname{O}_6](\operatorname{NO}_3)_6(\operatorname{OH}) \cdot 3\operatorname{H}_2\operatorname{O} \xrightarrow{400-450^{\circ}} \alpha - \operatorname{Bi}_2\operatorname{O}_3$$
(10.5)

$$\operatorname{Bi}(\operatorname{NO}_3)_3 \cdot \operatorname{N}_2\operatorname{O}_4 \xrightarrow{200^{\circ}} \operatorname{Bi}_2\operatorname{O}(\operatorname{NO}_3)_4 \xrightarrow{415^{\circ}} \operatorname{Bi}_4\operatorname{O}_5(\operatorname{NO}_3)_2$$
(10.6)

where the equation number of the first line is suppressed by declaring \notag.

10.1.3 Creation of the chemalign Environment

The chemalign environment can be created as a chemical version of the align environment of the amsmath package. What you have to do is only to declare $\mexchemenvironment{chemalign}{align}$. Just as the align environment of the amsmath package is based on the alignment mechanism of T_EX , the present chemalign environment succeeds in functions due to the alignment mechanism. Hence, such commands as \mealign can be used in the chemalign environment so as to give the following output:

\newchemenvironment{chemalign}{align}

\begin{chemalign}
5Fe^{2+} & \llongrightarrow 5Fe^{3+} + 5e^{-} \notag \\
Mn0_{4}^{-} + 5e^{-} + 8H^{+} & \llongrightarrow Mn^{2+} + 4H_{2}O \notag \\
\noalign{\vskip-8pt}
\noalign{\hfil\hbox to9cm{\hrulefill\kern0.5cm}\hfil}
\noalign{\vskip-4pt}
Mn0_{4}^{-} + 5Fe^{2+} + 8H^{+} & \llongrightarrow Mn^{2+} + 5Fe^{3+} + 4H_{2}O \\
end{chemalign}

$$5Fe^{2+} \longrightarrow 5Fe^{3+} + 5e^{-}$$

$$MnO_4^- + 5e^- + 8H^+ \longrightarrow Mn^{2+} + 4H_2O$$

$$MnO_4^- + 5Fe^{2+} + 8H^+ \longrightarrow Mn^{2+} + 5Fe^{3+} + 4H_2O$$
(10.7)

On a similar line, the chemalign* environment corresponding to the align* environment of the amsmath package can be created by declaring \newchemenvironment{chemalign*}{align*}.

```
\newchemenvironment{chemalign*}{align*}
\begin{chemalign*}
Co(CN)_{6}^{4-} & \equilibarrow Co(CN)_{6}^{3-} + e^{-} \\
Fe(CN)_{6}^{4-} & \equilibarrow Co(CN)_{6}^{3-} + e^{-}
\end{chemalign*}
```

$$\begin{array}{c} \operatorname{Co}(\operatorname{CN})_{6}^{4-} & \longrightarrow & \operatorname{Co}(\operatorname{CN})_{6}^{3-} + e^{-} \\ \operatorname{Fe}(\operatorname{CN})_{6}^{4-} & \longrightarrow & \operatorname{Co}(\operatorname{CN})_{6}^{3-} + e^{-} \end{array}$$

10.1.4 Creation of the chemalignat Environment

The chemalignat environment can be created as a chemical version of the alignat environment of the amsmath package by declaring \newchemenvironment{chemalignat}{alignat}. The usage of the newly-defined chemalignat environment is exemplified as follows:

```
\newchemenvironment{chemalignat}{alignat}
\begin{chemalignat}{4}
\mbox{Solution of zinc oxide:} \qquad
& Zn0 + 2H^{+} & \llongrightarrow\quad & Zn^{2+} + H_{2}O \\
\mbox{Cathode reaction:} \qquad
& Zn^{2+} + 2e^{-} & \llongrightarrow\quad & Zn \\
\mbox{Anode reaction:} \qquad
& H_{2}O & \llongrightarrow\quad & {\textstyle\frac{1}{2}}O_{2} + 2H^{+} + 2e^{-} \\
\mbox{Over-all reaction:} \qquad
& Zn0 & \llongrightarrow\quad & Zn + {\textstyle\frac{1}{2}}O_{2}
\end{chemalignat}
```

Solution of zinc oxide:	$ZnO + 2H^+ \longrightarrow$	$\rm Zn^{2+} + H_2O$	(10.8)
Cathode reaction:	$Zn^{2+} + 2e^{-} \longrightarrow$	Zn	(10.9)
Anode reaction:	$H_2O \longrightarrow$	$\frac{1}{2}O_2 + 2H^+ + 2e^-$	(10.10)

 \longrightarrow Zn + $\frac{1}{2}O_2$ Over-all reaction: ZnO (10.11)

The Use of the split Environment 10.1.5

The split environment supported by the amsmath package is originally used in combination with equation (redefined in amsmath), gather, etc. Because the ChemEquation environment of the present chmst-pdf package has been tuned to the setting of the amsmath package, it can be used in combination with the **split** environment:

```
\begin{ChemEquation}
\begin{split}
 [(eta^{5}\mbox{-}C_{5}H_{5})(C0)_{2}MnSbPhI_{2}]
& + [(\eta^{5}\mbox{-}C_{5}H_{5})Mn(CO)_{2}]\cdot THF \\
& \reactrarrow{0pt}{3cm}{\scriptsize K/THF}{\scriptsize [18]crown-6}
[PhSb{Mn(CO)_{2}(\eta^{5}\mbox{-}C_{5}H_{5})}] + 2KI + \cdots \)
& \mbox{and further lines}
\end{split}
\end{ChemEquation}
```

$$[(\eta^{5}\text{-}C_{5}\text{H}_{5})(\text{CO})_{2}\text{MnSbPhI}_{2}] + [(\eta^{5}\text{-}C_{5}\text{H}_{5})\text{Mn}(\text{CO})_{2}] \cdot \text{THF}$$

$$\xrightarrow{\text{K/THF}} [\text{PhSb}\{\text{Mn}(\text{CO})_{2}(\eta^{5}\text{-}C_{5}\text{H}_{5})\}_{2}] + 2\text{KI} + \cdots \quad (10.12)$$
and further lines

where an ampersand is used to mark an alignment point.

Note that the equation number is centered vertically on the height of the split environment. To print the equation number at the end of the display equation, the switching command \ctagsplit@false is declared as follows:

```
{\makeatletter
\ctagsplit@false
\begin{ChemEquation}
\begin{split}
 [(eta^{5}\mbox{-}C_{5}H_{5})(C0)_{2}MnSbPhI_{2}]
& + [(\eta^{5}\mbox{-}C_{5}H_{5})Mn(CO)_{2}]\cdot THF \\
& \reactrarrow{0pt}{3cm}{\scriptsize K/THF}{\scriptsize [18]crown-6}
[PhSb{Mn(CO)_{2}(\eta^{5}\mbox{-}C_{5}H_{5})}_{2}] + 2KI + \cdots \)
& \mbox{and further lines}
\end{split}
\end{ChemEquation}
makeatother
}
  [(\eta^5-C_5H_5)(CO)_2MnSbPhI_2] + [(\eta^5-C_5H_5)Mn(CO)_2] \cdot THF
```

$$\xrightarrow{\text{K/THF}} [\text{PhSb}\{\text{Mn}(\text{CO})_2(\eta^5 - \text{C}_5\text{H}_5)\}_2] + 2\text{KI} + \cdots$$

and further lines (10.13)

The chemalign environment defined by \newchemenvironment{chemalign}{align} can be combined with the split environment.

\begin{chemalign}
\begin{split}
H_{2}O + H_{2}SO_{4} & \equilibarrow H_{3}O^{+} + HSO_{4}^{-}; \\
& \mathit{K}_{H_{2}O}(25^{\circ}) = \frac{[H_{3}O^{+}][HSO_{4}^{-}]}{[H_{2}O]} \sim 1
\end{split} \\
SO_{3} + H_{2}SO_{4} & \equilibarrow H_{2}S_{2}O_{7} \\
begin{split}
H_{2}S_{2}O_{7} + H_{2}SO_{4} & \equilibarrow H_{3}SO_{4}^{+} + HS_{2}O_{7}^{-}; \\
& \mathit{K}_{H_{2}S_{2}O_{7}} + H_{2}SO_{4} & \equilibarrow H_{3}SO_{4}^{+} + HS_{2}O_{7}^{-}; \\
& \mathit{K}_{H_{2}S_{2}O_{7}} + H_{2}SO_{7}^{-}; \\
& \mathit{K}_{H_{2}S_{2}O_{7}} + H_{2}SO_{7}^{-}]}{[H_{2}S_{2}O_{7}^{-}]} = 1.4 \times 10^{-2} \end{split}
\end{chemalign}

$$SO_3 + H_2SO_4 \rightleftharpoons H_2S_2O_7$$

$$H_2S_2O_7 + H_2SO_4 \rightleftharpoons H_3SO_4^+ + HS_2O_7^-;$$
(10.15)

$$K_{\rm H_2S_2O_7}(25^{\circ}) = \frac{[\rm H_3SO_4^+][\rm HS_2O_7^-]}{[\rm H_2S_2O_7]} = 1.4 \times 10^{-2} \tag{10.16}$$

The chemgather environment defined above is capable of containing the split environment (the amsmath package) and the chemalign* environment (defined above) at the same time. The following example exemplifies such a nested specification of chemical equations:

\begin{chemgather}
\begin{split}
[Cr(C0)_{6}] & \overset{A}{\llongrightarrow}
\underset{yellow}{[Cr(C0)_{5}(AsPhH_{2})]} \overset{B}{\llongrightarrow} \\
& \underset{orange}{[Cr(C0)_{5}(AsPhLi_{2})]} \overset{C}{\llongrightarrow}
\underset{\text{dark violet}~(mp~104^{\circ})}{[\{Cr(C0)_{5}\}_{2}AsPh]}
\end{split} \\
begin{chemalign*}
A:\quad & PhAsH_{2} \\
B:\quad & LiBu \\
C:\quad & \text{cyclohexyl-}NC1_{2}
\end{chemalign*}

$$[\operatorname{Cr}(\operatorname{CO})_{6}] \xrightarrow{A} [\operatorname{Cr}(\operatorname{CO})_{5}(\operatorname{AsPhH}_{2})] \xrightarrow{B}$$

$$[\operatorname{Cr}(\operatorname{CO})_{5}(\operatorname{AsPhLi}_{2})] \xrightarrow{C} [\{\operatorname{Cr}(\operatorname{CO})_{5}\}_{2}\operatorname{AsPh}]$$

$$\operatorname{dark \ violet \ (mp \ 104^{\circ})}$$

$$A: \quad \operatorname{PhAsH}_{2}$$

$$(10.17)$$

B: LiBu

 $\mathbf{C}:\qquad \mathbf{cyclohexyl-NCl}_2$
10.2 Objects Placed Over or Under Arrows

10.2.1 Combination of Commands

The command **\overset** of the **amsmath** package is applicable to place an object over an arrow:

```
\begin{ChemEquation}
CC1_{4} + HF \overset{SbFC1_{4}}{\lllongrightarrow}
CFC1_{3} + HC1
\end{ChemEquation}
```

$$\operatorname{CCl}_4 + \operatorname{HF} \xrightarrow{\operatorname{SbFCl}_4} \operatorname{CFCl}_3 + \operatorname{HCl}$$
(10.18)

```
\begin{ChemEquation}
CFC1_{3} + HF
\stackrel{SbFC1_{4}}{\lllongrightarrow}
CF_{2}C1_{2} + HC1
\end{ChemEquation}
```

$$\operatorname{CFCl}_3 + \operatorname{HF} \xrightarrow{\operatorname{SbFCl}_4} \operatorname{CF}_2 \operatorname{Cl}_2 + \operatorname{HCl}$$
 (10.19)

A nested usage of \underset and \verset comes out well in placing objects under and over an arrow.

```
\begin{ChemEquation}
Na + Al + 2H_{2}
\underset{350~atom}{\overset{THF/140^{\circ}/3\: h}{\lllongrightarrow}}
NaAlH_{4}\quad (99\%~yield)
\end{ChemEquation}
```

$$Na + Al + 2H_2 \xrightarrow{\text{THF}/140^{\circ}/3 \text{ h}}_{350 \text{ atom}} NaAlH_4 \quad (99\% \text{ yield}) \tag{10.20}$$

Chapter 11

Coloring Substituents and Substitution Bonds

One of main improvements of the $\hat{X}^{1}MT_{E}X$ system Version 5.00 (both the PostScript-compatible mode and the PDF-compatible mode) is a newly-developed set of enhanced functions for drawing colored structural formulas. The chapters after this chapter are devoted to explain such enhanced functions for coloring.

11.1 Commands for Colors

Because the PostScript-compatible mode of the X²MT_EX system (called by the xymtexps) is based on the pstricks package, such commands as \red, \green, and \blue are defined to generated colored objects. These commands are not defined in the PDF-compatible mode of the X²MT_EX system (called by the xymtexpdf), because the pgf package used as a graphical tool calls the xcolor package internally as a coloring tool. Instead of these commands, the \color command is used in the PDF mode. For example, the code {\color{red} Red} generates Red.

For the sake of compatibility, the X²MT_EX system supports a set of control sequences of declarationtype, i.e., \red, \green, blue, \cyan, \magenta, \yellow, white, and \black, which are differently defined according to a mode selected. Thereby, the codes {\red Red}, {\green Green}, and {\blue Blue} in the PDF-compatible mode generate colored objects such as Red, Green, and Blue in a parallel way to the PostScript-compatible mode.

Further, the X¹MT_EX system supports another set of control sequences of command-type, i.e., \redx, \greenx, bluex, \cyanx, \magentax, \yellowx, whitex, and \blackx, which are differently defined according to a mode selected. Thereby, the codes \redx{Red},\greenx{Green}, and \bluex{Blue} in the PDF-compatible mode as well as in the PostScript-compatible mode generate colored objects such as Red, Green, and Blue.

11.2 Coloring Substituents

To color a substituent, the substituent is designated in a SUBSTLIST (substitution list) by adding an appropriate command of coloring. For example, the codes:

```
\bzdrv{1=={\red OH}}
\bzdrh{4=={\red \bzdrh{1==(y1);4=={\blue CH$_{3}$}} \hskip60pt
{\green \bzdrh{4=={\red \bzdrh{1==(y1);4=={\blue CH$_{3}$}}}}
```

produce the following structures:



where the phenyl group designated by a (yl) function is regarded as a substituent to be colored. A hetero atom in a heterocycle can be colored by adding an appropriate command of coloring to the

ATOMLIST (atom list) of the command for drawing the heterocycle. For example, the codes: \sixheterovi[ace]{1=={\red N}}{2=={\blue F}}

produce the following structures:



Parts of a spiro ring can be colored by adding appropriate commands of coloring in the ATOMLIST. For example, the code:

 $sixheteroh{4s=={red sixheteroh}{1==(y1)}}; 1s=={blue sixheteroh}{4==(y1)}}{$

produces the following structure:



In most cases, the color of a substitution bond can be differentiated from the color of the corresponding substituent by using the \aftergroup command of T_EX primitive. The formulas drawn above are modified by this technique as follows:

```
\bzdrv{1=={\red\aftergroup\blue OH};4=={\blue\aftergroup\red OH}}
\bzdrh{4=={\red\aftergroup\blue \bzdrh{1==(y1);%
4=={\blue\aftergroup\green CH$_{3}$}} \hskip60pt
{\green \bzdrh{4=={\red\aftergroup\blue \bzdrh{1==(y1);%
4=={\blue\aftergroup\black CH$_{3}$}}}
```



On the same line, we obtain the following printing of structural formulas, where the color of a substitution bond is differentiated from that of the corresponding substituent. The codes:

```
\sixheterovi[ace]{1=={\red N}}{2=={\blue\aftergroup\red F};%
4=={\red\aftergroup\blue Cl}}
\sixheterovi{1==\downnobond{\red N}{\blue H};4=={\green 0}}%
{2=={\blue\aftergroup\red CH$_{3}}}
```

produce the following structures:



Note

The use of dvipdfmx in processing the resulting dvi files containing these coloring codes may result in the appearance of a warning:

** WARNING ** Color stack underflow. Just ignore.

In spite of this warning, we can obtain the correct printing of structural formulas. So we just ignore this warning in the preceding examples.

The warning can be avoided by using control sequences of command type such as \redx in place of control sequences of declaration type such as \red . Hereafter we will use the former type as far as possible.

11.3 Coloring Substitution Bonds

11.3.1 Systematic Method for Coloring Substitution Bonds

The bondcolor package of the $\hat{X}^{T}MT_{E}X$ version 5.00 supports the function of coloring substitution bonds. The command \addbscolor is defined to specify the color of a substitution bond. It has a format represented by

\addbscolor{COLOR}{SUBSTITUENT}

The argument COLOR is a command for specifying color $(\red)^1$ and the second argument SUBSTITUENT is a text or a set of commands for drawing a substituent. For example, the codes:

```
\label{eq:product} $$ \ 1==\dbscolor{\red}{F}; 4==Cl} \\ bzdrv{1==\dbscolor{\red}{\blackx{F}}; 4==Cl} \\ bzdrv{1==\dbscolor{\red}{\bluex{F}}; 4==Cl} \\ bzdrv{1==\bluex{F}}; 4==Cl}
```

generate the following structural formulas:

¹This command can be used in the PDF mode (as well as in the PostScript mode), because it has been defined for representing $color{red}$.



The **\addbscolor** command changes the color of a substitution bond as well as that of a substitution, as shown in the first example. If a substitution bond is desired to be solely changed, {**\black F**} should be declared as the second argument SUBSTITUENT, as shown in the second example. The third example shows the separate change of the colors of a substitution bond and a substituent.

11.3.2 Examples

Other commands defined by the carom package are also colored by the systematic methods. More lowlevel commands defined by the hetarom and hetaromh packages can be colored similarly. For example, a six-membered hydroxylamine is drawn in different ways by the codes:

which generate the following structures:



Substitution bonds in fused rings can be colored by the systematic method as follows:

```
\label{terov} = N^{{10}B==\addbscolor{\red}{bluex{0H}} \decaheterov{9==N\rlap{^{+}}}{9B==\addbscolor{\red}{\bluex{0}^{-}}}; {10}B==0H} \decaheterov[{{10}+}]{{10}}==N^{9B==\addbscolor{\red}{\bluex{0H}}}; {10}B==\addbscolor{\red}{\bluex{0H}}; {10}B==\addbscolor{\red}{\bluex{0}}} \decaheterov{0}{\red}{\bluex{0}}; {10}B==\addbscolor{\red}{\bluex{0}}; {10}B==\addbscolor{\red}{\Bluex{0}}; {10}B==\addbscolor{\red}{\Bluex{0}}; {10}B==\bluex{0}{\bluex{0}}; {10}B==\bluex{0}{\bluex{0}}; {10}B==\bluex{0}{\bluex{0}}; {10}B==\bluex{0}{\bluex{0}}; {10}B==\bluex{0}{\bluex{0}}; {10}B==\bluex{0}{\bluex{0}}; {10}B=\bluex{0}{\bluex{0}}; {10}B=\bluex{0}{\bl
```



Commands defined by the aliphat package are also colored by the systematic methods. The following examples illustrate colored structural formulas produced by using the \addbscolor command variously in the \tetrahedral command.

```
\tetrahedral{0==\greenx{C};
1==\addbscolor{\red}{\bluex{F}};%
2==Cl;3==\addbscolor{\red}{\bluex{Br}};4==I} \qquad
\tetrahedral{0==\greenx{C};
1A==\addbscolor{\red}{\bluex{F}};%
2B==Cl;3A==\addbscolor{\red}{\bluex{Br}};4B==I} \qquad
\tetrahedral{0==\greenx{C};1==F;%
2==\addbscolor{\blue}{\redx{Cl}};
3==Br;4==\addbscolor{\blue}{\redx{I}} \qquad
\tetrahedral{0==\greenx{C};1B==F;%
2A==\addbscolor{\blue}{\redx{Cl}};
3B==Br;4A==\addbscolor{\blue}{\redx{I}}
```



The following examples show colored structures produced by using the **\addbscolor** command in the **\ltrigonal** command.

```
\ltrigonal{0==C;1D==0;2==Cl;3==\addbscolor{\red}{CH$_{3}$}}quad \\ ltrigonal{0==\greenx{C};1D==\addbscolor{\red}{\bluex{0}};2==Cl;3==CH$_{3}$}\quad \\ ltrigonal{0==\greenx{C};1D==\addbscolor{\red}{\bluex{0}};2A==Cl;% \\ 3B==\addbscolor{\red}{\bluex{CH$_{3}}} \\
```



Wedged and dashed bonds for stereochemistry can be colored by the systematic way. For example, the codes:

```
\DtetrahedralS{0==C;1==F;
2==Cl;3A==Br;4B==I} \qquad
\DtetrahedralS{0==\green{C};
1==\addbscolor{\red}{\bluex{F}};
2==Cl;3A==Br;4B==I} \qquad
\DtetrahedralS{0==\greenx{C};1==F;
2==\addbscolor{\red}{\bluex{Cl}};3A==Br;4B==I} \qquad
\DtetrahedralS{0==\greenx{C};1==F;
2==Cl;3A==\addbscolor{\red}{\bluex{Br}};4B==I} \qquad
\DtetrahedralS{0==\greenx{C};1==F;
2==Cl;3A==Br;4B==\addbscolor{\red}{\bluex{I}}
```

produce the following colored structures:



```
\chair{1D==\addbscolor{\red}{\bluex{0}};%
4Se==\addbscolor{\red}{CH$_{3}$};4Sa==\addbscolor{\blue}{OH}}\qquad
\chairi{1D==\addbscolor{\red}{\bluex{0}};%
4Se==\addbscolor{\red}{CH$_{3}$};4Sa==\addbscolor{\blue}{OH}}
```

produce colored structural formulas:



The command **\bornane** defined in the ccycle package of the $\hat{X}^{2}MT_{E}X$ system has been also modified to support a systematic method of coloring substitution bonds.

```
\bornane{7D==\addbscolor{\red}{\bluex{0}};%
2Sa==\addbscolor{\red}{CH$_{3}$};2Sb==\addbscolor{\blue}{OH}}\qquad
\bornane{7Sa==\addbscolor{\red}{CH$_{3}$};7Sb==\addbscolor{\red}{CH$_{3}$};%
2D==\addbscolor{\blue}{0}}
```



The following examples show the coloring of substitution bonds by using the <code>\addbscolor</code> command in the arguments of the commands <code>\adamantane</code> and <code>\hadamantane</code>, the definitions of which have been modified in the ccycle package of the $\hat{X}^{2}MT_{E}X$ system for the purpose of supporting a systematic mechanism of coloring.

```
\adamantane{2D==\addbscolor{\blue}{0};%
6a==\addbscolor{\red}{\bluex{CH$_{3}$};6b==\addbscolor{\red}{\bluex{CH$_{3}$}} \qquad
\hadamantane{2D==\addbscolor{\blue}{0};%
6a==\addbscolor{\red}{\bluex{CH$_{3}$}};6b==\addbscolor{\red}{\bluex{CH$_{3}$}}
```



Substitution bonds in a furanose derivative drawn by the command furanose (defined in the hcycle package of the \hat{X}^2MT_EX system) can be colored by using the \addbscolor command. By applying the method to the example shown in Subsection 5.1.1, substitution bonds in the structural formula of ribavirin can be colored, as found in the following result:

```
\begin{XyMcompd}(1200,1000)(100,0){}{}
\furanose{1Sa==H;2Sb==H;2Sa==0H;3Sb==H;3Sa==0H;4Sa==H;%
4Sb==\addbscolor{\red}{\bluex{HOC\rlap{H$_{2}$}};
1Sb==\fiveheterov[bd]{1==N;2==N;4==N}{1==(y1);
3==\addbscolor{\blue}{\redx{CONH$_{2}$}}}
\end{XyMcompd}
```



The command \pyranose defined in the hcycle package of the X¹MT_EX system has been modified to support a systematic method of coloring substitution bonds. The following example shows the use of \black in the second argument of the \addbscolor command in order to avoid the coloring of a furan skeleton due to \fiveheterov and to avoid the coloring of a pyranose skeleton due to \pyranose. Note that the furan skeleton and the pyranose skeleton are regarded as substituents of the steroid skeleton in the following code:

```
\begin{XyMcompd}(2100,1800)(-500,-280){}{}
\steroid{{10}B==\addbscolor{\red}{\bluex{\lmoiety{OHC}}};%
{14}A==\addbscolor{\red}{\bluex{OH}};%
{13}B==\addbscolor{\red}{\bluex{\lmoiety{H$_{3}$C}}};%
{16}B==\addbscolor{\red}{\bluex{OH}};%
{17}B==\addbscolor{\red}{\bluex{0H}};%
{17}B==\addbscolor{\red}{\bluex{0H}};%
{3B==\addbscolor{\red}{\blackx{%
\fiveheterov[e]{3==0}{4D==\addbscolor{\red}{\bluex{0}};1==(y1)}};%
3B==\addbscolor{\red}{\blackx{\ly1(3==\bluex{0}){8==%
}pyranose{1Sb==(y1);1Sa==H;2Sa==0H;3Sb==H;3Sa==OH;4Sb==H0;%
4Sa==H;5Sb==H;5Sa==\addbscolor{\red}{\bluex{CH$_{3}$}}}}}\
end{XyMcompd}
```



Substitution bonds of polymethylene units defined in the methylene package are now capable of being colored by means of the \addbscolor command. The following structure is drawn by this technique applied to the command \heptamethylene.

 $\label{lim} $$ $ $ D = \ [a] {} {1W = \ (yl); $$ Sa = \ (yl); $$ Sa = \ (yl); $$ Sb = \ (yl)$



On a similar line, this technique applied to the command \nonamethylene draws the following structure with colored bonds and substituents.

```
\nonamethylene{%
1s==\bzdrv{1==\addbscolor{\blue}{\redx{C00H}};2==(y1);5==H0;6==H0};%
9s==\fiveheterov{1==0}{5==(y1);5SB==H;%
4GB==\addbscolor{\red}{\bluex{Me}};2GA==Et;%
2Su==\sixheterovi{1==0}{6==(y1);6FA==H;3SB==0H;3SA==Et;%
2A==\addbscolor{\red}{\bluex{Me}}}%
{4B==\addbscolor{\red}{\bluex{Me}};5B==0H;%
6A==\addbscolor{\red}{\bluex{Me}};7D==0;8A==\addbscolor{\red}{\bluex{Et}}}
```



Further, the following structure is drawn by applying this technique to the \decamethylenei command.

\decamethylenei[b]{}{1==\lmoiety{TBMSO};3==\null;%
4A==\addbscolor{\red}{\bluex{\lmoiety{MeO}}};6A==OTBDMS;%
9B==\null;{10}D==0;%
{10}W==\trimethylenei{1==0}{1==(y1);%
3D==\addbscolor{\red}{\bluex{0}};3W==OMe}}



The command **\polyethylene** defined in the **polymers** package has been also modified to support this technique of bond coloring, as shown in the following example:

```
\polyethylene{1==C;2==C}%
{1==CH$_{3}$;2==;3==H;4==\addbscolor{\red}{\bluex{CH$_{3}$};%
5==;6==\addbscolor{\red}{\bluex{CH$_{3}$};0=={\redx{n}}}
```



Substitution bonds in a spiro pyranose ring due to \pyranoseChairi (defined in the steroid package) as well as in a fused furan ring due to \fivefusev are now capable of being colored by means of the \addbscolor command, as found in the following example. Note that the bond coloring technique applied to these commands works well even when these are designated as a fusing unit or a more-nested spiro unit in the BONDLIST (bond list) argument of the command steroid.

```
\begin{XyMcompd}(2200,1350)(200,150){}{}
\steroid[%
{s{\fivefusev{2==0;%
3s==\mbox{\changeunitlength{0.07pt}%
\pyranoseChairi{1==(y1);%
4U==\addbscolor{\red}{\bluex{\normalsize CH$_{3}$}}}%
{4A==\addbscolor{\red}{\bluex{CH$_{3}$}}{e}%
}}]{5B==H;{10}B==\addbscolor{\red}{\bluex{CH$_{3}$}};
8B==H;9A==H;{13}B==\addbscolor{\red}{\bluex{CH$_{3}$}};
8B==H;9A==H;{13}B==\addbscolor{\red}{\bluex{CH$_{3}$}};
{14}A==H;{17}GA==H;{16}GA==H}
\end{XyMcompd}
```



It should be emphasized that wedged bonds, dashed bold bonds, and wavy bonds for stereochemistry are colored by the same technique of bond coloring due to the **\addbscolor** command.

Commands of high level have a fixed set of substituents, which are not changed by the systematic method described here. For the purpose of coloring the fixed set, you should use the corresponding commands of low level. For example, the high-level command \cholestanealpha is compared with the low-level command \steroidchain as follows:

```
\begin{tabular}{11}
high-level macro: & low-level macro: \\
\cholestanealpha{3B==\addbscolor{\blue}{HO};
{24}SA==\dbscolor{\red}{CH$_{3}$};{24}SB==\dbscolor{\red}{H}}
X.
\steroidchain{3B==\addbscolor{\blue}{HO};
{24}SA==\addbscolor{\blue}{CH$_{3}$};{24}SB==\addbscolor{\blue}{H};
5A==\addbscolor{\red}{H};
8B==\addbscolor{\red}{H};
9A==\addbscolor{\red}{H};
\{10\}B==\addbscolor\{\red}{CH$_{3}$};
\{13\}B==\addbscolor\{\red}{CH$_{3}$};
{14}A==\dbscolor{\Ted}{H};
{17}SA==\addbscolor{\red}{\raisebox{-3pt}{H}};
{20}SA=\dbscolor{\red}{CH}_{3}}; {20}SB=\addbscolor{\red}{H}
\backslash \backslash
\end{tabular}
```

high-level macro:

low-level macro



Note that the black bonds in the left structure (or the red bonds in the right structure) indicate the fixed set of substituents for the **\cholestanealpha** command of high level.

11.3.3 Switches for Coloring Substitution Bonds

The mechanism for coloring substitution bonds relies on a one-by-one procedure of testing whether or not the designation of each substituent contains the **\addbscolor** command. Because this mechanism works even if the **\addbscolor** command is not contained, there are accidental cases which require prolonged processing times. In particular, multiple nesting of substituents causes such prolongation, as exemplified in the structure shown in Page 29.² For the purpose of skipping such prolonged bond coloring, the switch **\bscolorswOFF** is defined. The switch **\bscolorswON** is ready to return to the default mode of coloring.

²In fact, the deletion of the \bscolorswOFF in the example of Page 29 results in such prolonged processing.

```
\begin{tabular}{ccc}
default (\verb/\bscolorswON/) & \verb/\bscolorswOFF/ & \verb/\bscolorswON/ \\
\bzdrv{1==\addbscolor{\red}{\bluex{F}};4==Cl} &
\bscolorswOFF \bzdrv{1==\addbscolor{\red}{\bluex{F}};4==Cl} &
\bscolorswON \bzdrv{1==\addbscolor{\red}{\bluex{F}};4==Cl} \\
\bzdrv{1==\addbscolor{\red}{F};4==Cl} &
\bscolorswOFF \bzdrv{1==\addbscolor{\red}{F};4==Cl} &
\bscolorswOFF \bzdrv{1==\addbscolor{\red}{F};4==Cl} &
\bscolorswON \bzdrv{1==\addbscolor{\red}{F};4==Cl} \\
\end{tabular}
```



It should be noted that the designation of each colored substituent ({\blue F}) shown in the first row of structures results in the same effect during such switching of coloring.

11.4 Dirty Techniques for Coloring Substitution Bonds

If no systematic routes to color substitution bonds have been defined with respect to your target command for drawing structures, rather dirty (but versatile) techniques should be tested case by case. For example, the **\PutBondLine** command is used to place a substitution bond (Page 23), which is colored by adding an appropriate command of coloring.

For the simplicity of explanation about such dirty techniques, let us use again the examples described above for the systematic technique. For example, the codes:

```
\sixheterov[ace]{1s==\redx{\PutBondLine(0,0)(0,140){\thinLineWidth}};%
1s==\bluex{\put(-30,150){OH}}}{}
\qquad \raisebox{45pt}{Compare with}
\sixheterov[ace]{}{1==\addbscolor{\red}{\bluex{OH}}}
```

generate the following structures:



where the ATOMLIST (atom list) is used in place of the SUBSLIST (substituent list) in order to place the substitution bond at position 1. Compare the left formula with the right formula drawn by the systematic method of bond coloring.

By replacing \thinLineWidth by \thickLineWidth in the last argument of \PutBondLine, the substitution bond appears as a bold-line bond. The same effect can be brought about locally by the first argument {\red\dashhasheddash\thicklines} of the \addbscolor command. Compare between the codes:

```
\sixheterov[ace]{1s==\redx{\PutBondLine(0,0)(0,140){\thickLineWidth}};%
1s==\bluex{\put(-30,150){0H}}{}
\qquad \raisebox{45pt}{Compare with}
\sixheterov[ace]{}{1==\addbscolor{\red
\dashhasheddash
\thickLines}{\bluex{0H}}}
```

which generate the following structures:



It should be pointed out again that the last argument of **\PutBondLine** is capable of accommodating a direct designation of a bond width, as found in the following example (left). Note that **\thinLineWidth** is equal to 0.4pt and **\thickLineWidth** is equal to 1.6pt.

The following example (left) shows a thicker substitution bond (2.5pt), which is given directly to the last argument of \PutBondLine. The same effect is accomplished by designating the first argument of \addbscolor as being {\red\dashhasheddash \def\thickLineWidth{2.5pt} \thicklines}, as shown in the right example.

```
\sixheterov[ace]{1s=={\red \PutBondLine(0,0)(0,140){2.5pt}};%
1s=={\blue \put(-30,150){0H}}}{
\qquad \raisebox{45pt}{Compare with}
\sixheterov[ace]{}{1==\addbscolor{\red
\dashhasheddash\def\thickLineWidth{2.5pt}%
\thicklines}{\bluex{0H}}}
OH
OH
OH
OH
OH
```

The \WedgeAsSubst command is used to draw a substitution bond of wedge type (Page 23), which is colored by adding an appropriate command of coloring. The systematic way of bond coloring supports this case as a standard usage. The codes:

```
\sixheterov{1s==\redx{\WedgeAsSubst(0,0)(0,1){140}};%
1s==\bluex{\put(-30,150){0H}}}{
\qquad \raisebox{45pt}{Compare with}
\sixheterov{}{1B==\addbscolor{\red}{\bluex{0H}}}
```

generate the following structures:



A nitrogen atom at position 1 can be typeset by placing the designation 1==N in the ATOMLIST (atom list) after setting up the substitution bond and the substituent. This position of the designation is important to restore a flag of truncation at the position 1, as shown in the following code:

```
\sixheterov{1s==\redx{\WedgeAsSubst(0,40)(0,1){100}};%
1s==\bluex{\put(-30,150){0H}};1==N}{}
\qquad \raisebox{45pt}{Compare with}
\sixheterov{1==N}{1B==\addbscolor{\red}{\bluex{0H}}}
```

The code and a comparison code of a systematic way generate the following structures:



The \PutDashedBond command (Page 26) is used to draw a dashed bond, as found in the following examples:

```
\sixheterov{1s==\redx{\PutDashedBond(0,0)(0,140){\thickLineWidth}};%
1s==\bluex{\put(-30,150){0H}}}{
\qquad \raisebox{45pt}{Compare with}
\sixheterov{}1a==\addbscolor{\red}{\bluex{0H}}} \par
\sixheterov{1s==\redx{\PutDashedBond(0,40)(0,140){\thickLineWidth}};%
1s=={\blue \put(-30,150){0H}};1==N}{}
\qquad \raisebox{45pt}{Compare with}
\sixheterov{1==N}{1A==\addbscolor{\red}{\bluex{0H}}}
```





11.5 Substitution Bonds of Unnatural Directions

Usual substitution bonds drawn by the $\hat{X}^{2}MT_{E}X$ system have natural directions intrinsic to the attached positions of a skeleton. In some cases, however, substitution bonds of unnatural directions are necessary to draw special structural formulas. To do this task along with bond coloring, the commands \BightAtomBond and \LeftAtomBond are defined in the bondcolor package.

11.5.1 Commands for Use of Atom Lists

The \RightAtomBond Command

The macro \RightAtomBond is defined to color substitution bonds (and substituents), when it is to be designated in an ATOMLIST (atom list) or in a BONDLIST (bond list) of a command for drawing a structural formula.

\RightAtomBond(x,y)[COLOR]<CLIPPING>{SUBSTLIST}

The first optional argument (x,y) is a set of increments of xy-coordinates, if some adjustment is necessary when a spiro function (1s==...) is not supported. The default value is (0,0). The second optional argument [COLOR] is a command of coloring (e.g., \red). The default value is \black. The third optional argument <CLIPPING> accommodates <s> for non-clipping; and a vacant or omitted argument represents a clipping (truncation) of a corner. The default value is a vacant value. The fourth argument is a SUBSLIST (substituent list), in which a substitution number 1–8 represents a bond slope to be drawn. Slopes to cover a right-handed half circle are numbered on a similar line to those of the \ryl macro, as follows:

SLOPE

0	:	(0,1)
1	:	(3,5)
2	:	(1,1)
3	:	(5,3)
4	:	(1,0)
5	:	(5,-3)
6	:	(1, -1)
7	:	(3,-5)
8	:	(0,-1)

As for the format of the SUBSLIST, see the specification of the ryl command (Chapter 4 of xymtx200.pdf [6]). When the number of a slope (n) and a bond specifier (M) are to be considered, designations in the SUBSLIST are given as being nM==... etc. For example, the designation OB==Cl for drawing a bond and a substituent corresponds to a slope 0 ((0,1)), a bond specifier B (β bond) and a substituent Cl (a chlorine atom). Figure 11.1 shows directions of substitution bonds drawn by \RightAtomBond, where each number corresponds to the direction listed above.



Figure 11.1: Directions of substitution bonds drawn by \RightAtomBond

Natural and unnatural directions of substitution bonds depend on respective substitution positions of a skeleton. For example, the top position (position 1) of a six-membered ring drawn by the \sixheterov command is substituted by a bond (a dashed bond) due to a \RightAtomBond command, whose ATOM-LIST (OA==Cl etc.) indicates a direction selected as being 1 to 8.

```
\textbf{Bonds of natural directions}: \\[10pt]
\sixheterov{1s==\RightAtomBond[\red]{0A==Cl}}{}
\sixheterov{1s==\RightAtomBond[\red]{1A==Cl}}{}
\sixheterov{1s==\RightAtomBond[\red]{2A==Cl}}{}
\sixheterov{1s==\RightAtomBond[\red]{3A==Cl}}{} \par
\textbf{Bonds of unnatural directions}: \\
\sixheterov{1s==\RightAtomBond[\red]{4A==Cl}}{
\sixheterov{1s==\RightAtomBond[\red]{5A==Cl}}{}
\sixheterov{1s==\RightAtomBond[\red]{5A==Cl}}{
\sixheterov{1s==\RightAtomBond[\red]{6A==Cl}}{
\sixheterov{1s==\RightAtomBond[\red]{7A==Cl}}{}
\sixheterov{1s==\RightAtomBond[\red]{8A==Cl}}{}
```

Bonds of natural directions:



The four structures in the first row show natural directions, while the five structures in the second row show unnatural directions, because the latter five contain a bond situated within the domain of the six-membered ring.

The following examples shows the placement of a nitrogen atom on the position 1 and the substitution due to \RightAtomBond, where the designation 1s== is replaced by 1==N;1== in the ATOMLIST of each \sixheterov command. Thereby, a clipping mechanism at the position 1 is realized.

```
\textbf{Bonds of natural directions}: \\[10pt]
\sixheterov{1==N;1==\RightAtomBond[\red]{0A==C1}}{}
\sixheterov{1==N;1==\RightAtomBond[\red]{2A==C1}}{}
\sixheterov{1==N;1==\RightAtomBond[\red]{3A==C1}}{}
\sixheterov{1==N;1==\RightAtomBond[\red]{3A==C1}}{}
\sixheterov{1==N;1==\RightAtomBond[\red]{4A==C1}}{}
\sixheterov{1==N;1==\RightAtomBond[\red]{5A==C1}}{}
\sixheterov{1==N;1==\RightAtomBond[\red]{5A==C1}}{}
\sixheterov{1==N;1==\RightAtomBond[\red]{6A==C1}}{}
\sixheterov{1==N;1==\RightAtomBond[\red]{7A==C1}}{}
\sixheterov{1==N;1==\RightAtomBond[\red]{8A==C1}}{}
```

Bonds of natural directions:



Bonds of unnatural directions:



The SUBSTLIST of the **\RightAtomBond** command has a similar format to the SUBSTLIST of commands for drawing structural formulas, where bond specifiers such as A (α -bonds), B (β -bonds), U (wavy bonds), D (exocyclic double bonds), and T (exocyclic triple bonds) can be declared. Random use of these bond specifiers is illustrated in the following examples:

```
\sixheterov{%
1s==\RightAtomBond[\red]{0==Cl};%
2s==\RightAtomBond[\red]{3D==0};%
3s==\RightAtomBond[\red]{5U==Cl};%
4s==\RightAtomBond[\red]{8B==OH};%
}{}
\sixheterov{%
1s==\RightAtomBond[\red]{0A==C1;2B==C1};%
2s==\RightAtomBond[\red]{3U==C1;4U==C1};%
3s==\RightAtomBond[\red]{6T==X};%
4s==\RightAtomBond[\red]{7B==C1;8A==C1};%
}{}
\sixheteroh{%
5s==\RightAtomBond[\red]{0B==H;8A==Cl};%
6s==\RightAtomBond[\red]{0B==Cl;8A==H}%
}{}
\sixheteroh{%
```

5s==\RightAtomBond[\red]{0B==C1;8A==H};%
6s==\RightAtomBond[\red]{0B==C1;8A==H}%
}{



The \LeftAtomBond Command

On a similar way, the macro \LeftAtomBond is defined to color substitution bonds (and substituents) in a left-handed half circle. The macro is to be designated in a ATOMLIST or in a BONDLIST of a command for drawing a structural formula. The format of specifying arguments for \LeftAtomBond is parallel to that of the \RightAtomBond command, as shown in the following scheme:

\LeftAtomBond(x,y)[COLOR]<CLIPPING>{SUBSTLIST}

The first optional argument (x,y), the first optional argument [COLOR], and the third optional argument <CLIPPING> are parallel to those of RightAtomBond. The fourth argument is a SUBSLIST, in which a substitution number 1–8 represents a bond slope to be drawn. Slopes to cover a left-handed half circle are numbered on a similar line to those of the 1y1 macro, as follows:

SLOPE

0	:	(0,1)
1		(-3,5)
2	:	(-1,1)
3	:	(-5,3)
4	:	(-1,0)
5	:	(-5,-3)
6	:	(-1,-1)
7	:	(-3,-5)
8	:	(-0,-1)

As for the format of the SUBSLIST, see the specification of the \lyl command (Chapter 4 of xymtx200.pdf [6]). Figure 11.2 shows directions of substitution bonds drawn by \LightAtomBond, where each number corresponds to the direction listed above.



Figure 11.2: Directions of substitution bonds drawn by \LeftAtomBond

The following examples show natural and unnatural directions of substitution bonds, where modes of depending on respective substitution positions of a skeleton are illustrated by using the \LeftAtomBond command.



Bonds of natural directions:



Bonds of unnatural directions:



The \LeftAtomBond command supports a function for permitting the placement of a hetero atom (e.g., a nitrogen atom) at the substitution position (e.g., the position 1), where the designation 1s== is replaced by 1==N; 1== in the ATOMLIST of each \sixheterov command.

```
\textbf{Bonds of natural directions}: \\[10pt]
\sixheterov{1==N;1==\LeftAtomBond[\red]{0A==C1}}{}
\sixheterov{1==N;1==\LeftAtomBond[\red]{2A==C1}}{}
\sixheterov{1==N;1==\LeftAtomBond[\red]{3A==C1}}{} \par
\textbf{Bonds of unnatural directions}: \\
\sixheterov{1==N;1==\LeftAtomBond[\red]{4A==C1}}{
\sixheterov{1==N;1==\LeftAtomBond[\red]{5A==C1}}{}
\sixheterov{1==N;1==\LeftAtomBond[\red]{5A==C1}}{}
\sixheterov{1==N;1==\LeftAtomBond[\red]{6A==C1}}{
\sixheterov{1==N;1==\LeftAtomBond[\red]{6A==C1}}{}
\sixheterov{1==N;1==\LeftAtomBond[\red]{7A==C1}}{}
\sixheterov{1==N;1==\LeftAtomBond[\red]{8A==C1}}{}
```

Bonds of natural directions:



Unnatural Directions of Substitution Bonds

13-Bromo-(2,6) metacyclophane (cf. Section 13.4 of the on-line document of $\hat{X}^{2}MTEX$ version 1.01 (xym-tex.pdf)) is printed by the following code, where the RightAtomBond command is used to draw a bromine substituent.

```
\sixheterov[ace%
{c{\sixfusev{}{F}[ef]}}%
{d{\sixfusev{}{A}[ab]}}%
]{4s==\RightAtomBond[\red\def\thinLineWidth{1.6pt}]%
{7==\kern0.1em\raise1ex\hbox{\blue Br}}{}
\sixheterov[ace%
{c{\sixfusev{}{F}[ef]}}%
{d{\sixfusev{}{A}[ab]}}%
]{}{4Sa==\addbscolor{\red\def\thinLineWidth{1.6pt}}%
{\kern.5em\raise1ex\hbox{\blue Br}}}
```



This case can be drawn as a natural direction as shown in the right structure, where the bromine substituent is regarded as a substituent of the benzene ring.

Unnatural directions of substitution bonds are illustrated by *trans*-cyclododecene, where structures of different positions of double bonds are drawn as follows by using \RightAtomBond and \LeftAtomBond.

```
\decaheterov[j]{%
{10}s==\LeftAtomBond(-32,30)[\red]{7==\raise1ex\hbox{H}\kern0.1em}}%
{1==\addbscolor{\red}{H}}[k]
\decaheterov[e]{%
9s==\RightAtomBond(-32,35)[\blue]{1==\kern0.1em\lower1ex\hbox{H}}}%
{5==\addbscolor{\blue}{H}}[k]
```



These formulas are drawn by using \decaheterov, where each central bond is erased by using the optional argument [k] placed at the end of the command. Because the spiro mechanism of the $\hat{X}^{4}MTEX$ system is not supported for the bridgehead positions of \decaheterov,³ the optional argument (x,y) should be used to adjust the joint positions of substitution bonds. The values of (x,y) are decided in a trial-and-error fashion.

Further examples for illustrating unnatural directions of substitution bonds are shown as follows:

```
\decaheterov{%
{10}s==\RightAtomBond(-32,30) [\red] {6A==Cl};%
{10}s==\LeftAtomBond(-32,30) [\red] {6B==Cl}}{[k]
\decaheterov{%
9s==\RightAtomBond(-32,35) [\red] {2A==Cl};%
9s==\LeftAtomBond(-32,35) [\red] {2B==Cl}}{[k]
```



Because these formulas are drawn by using $\cite{caheterov}$, similar situations to *trans*-cyclododecene hold true for these cases, so that the optional argument (x,y) is inserted to adjust the joint positions of substitution bonds.

Further additional examples for illustrating natural and unnatural directions of substitution bonds are shown by using \decaheterovt:

```
\begin{center}
\begin{tabular}{cc}
Natural directions & Unnatural directions \\[10pt]
\decaheterovt{}{9B==\addbscolor{\red}{\bluex{H}};%
{10}A==\addbscolor{\red}{\bluex{H}}}
&
\decaheterovt{9s==\RightAtomBond(35,29)[\red]{0B==\bluex{H}};%
{10}s==\RightAtomBond(-31,48)[\red]{8A==\bluex{H}}}{} \\
end{tabular}
```

³In spite of no support of the spiro mechanism, the specifier s of the designation $9s=or \{10\}s=is$ necessary to assure the correct behavior of the \RightAtomBond or \LeftAtomBond command.



The \decaheterovt command takes the same format as \decaheterov . Hence, similar situations described above hold true for these cases, so that the optional argument (x,y) is inserted to adjust the joint positions of substitution bonds.

Hetero atoms and substitution bonds at the bridgehead positions of \decaheterovt are illustrated in the following examples:

```
\begin{center}\\begin{tabular}{cc}\\Natural directions & Unnatural directions \\[10pt]\\\decaheterovt{9==\llap{$_{+}$}N;{{10}}=N\rlap{$^{+}$}}{9B==\addbscolor{\red}{bluex{0$^{-}$}};%{10}A==\addbscolor{\red}{bluex{0$^{-}$}};%{10}A==\addbscolor{\red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A==\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A=\normalfont{red}{bluex{0$^{-}$}};%{10}A
```



11.5.2 Commands for Use of Bond Lists

The \RightAtomBond and \LeftAtomBond Commands in Bond Lists

The commands \RightAtomBond and \LeftAtomBond described above are also capable of being declared in the BONDLIST. Hence, they are effective to commands which do not support the ATOMLIST.

Examples shown in Page 89 (in ATOMLIST) can be obtained by the present method using BONDLIST, where the optional argument <s> is used to assure correct behaviors of the command \RightAtomBond.

```
\sixheterov[{a{\RightAtomBond[\blue]<s>{0A=={\red OH}}}]{}{}
\sixheterov[{a{\RightAtomBond[\blue]<s>{1A=={\red OH}}}]{}{}
\sixheterov[{a{\RightAtomBond[\blue]<s>{2A=={\red OH}}}]{}{}
\sixheterov[{a{\RightAtomBond[\blue]<s>{3A=={\red OH}}}]{}}
```

Examples shown in Page 90 (in ATOMLIST) can be also obtained by the present method using BONDLIST.

```
\sixheterov[{a{\RightAtomBond[\blue]{0A=={\red OH}}}]{1==N}{}
\sixheterov[{a{\RightAtomBond[\blue]{1A=={\red OH}}}]{1==N}{}
\sixheterov[{a{\RightAtomBond[\blue]{2A=={\red OH}}}]{1==N}{}
\sixheterov[{a{\RightAtomBond[\blue]{3A=={\red OH}}}]{1==N}{}
\sixheterov[{a{\RightAtomBond[\blue]{4A=={\red OH}}}]{1==N}{}
\sixheterov[{a{\RightAtomBond[\blue]{5A=={\red OH}}}]{1==N}{}
\sixheterov[{a{\RightAtomBond[\blue]{5A=={\red OH}}}]{1==N}{}
\sixheterov[{a{\RightAtomBond[\blue]{6A=={\red OH}}}]{1==N}{}
\sixheterov[{a{\RightAtomBond[\blue]{6A=={\red OH}}}]{1==N}{}
\sixheterov[{a{\RightAtomBond[\blue]{7A=={\red OH}}}]{1==N}{}
\sixheterov[{a{\RightAtomBond[\blue]{8A=={\red OH}}}]{1==N}{}}
```



Unnatural Directions of Substitution Bonds

By using steroid of the \hat{X}^2MT_EX system, a substitution bond at the position 8 or 9 (a bridgehead position) is drawn as an endocyclic bond of the resulting steroid ring. Thus, the structure drawn by the

code:

$\label{eq:steroid} addscolor{\red}{\bluex{Br}}; 9B == \addscolor{\red}{\bluex{Br}}$

has endocyclic substitution bonds, as shown in the resulting formula:



Hence, the direction appearing in the above structure is regarded as being natural in the X²MT_EX system.

Exocyclic substitution bonds may be required in some situation. Because the **\steroid** command lacks an ATOMLIST but has a BONDLIST, the present technique is effective to draw such exocyclic substitution bonds. For example, the code:

\steroid[{g{\RightAtomBond[\red]<s>{5A==\bluex{Br}}}%
{h{\LeftAtomBond[\red]<s>{3B==\bluex{Br}}}]{}

generates the following structural formula:



where unnatural exocyclic substitution bonds are available.

In bicyclic systems such as decalines, being natural and being unnatural are inverted as compared with steroids. When we use the \decalinevt command in place of \decaheterovt (cf. Page 94), we should use the \RightAtomBond or \LeftAtomBond command in its BONDLIST. The following examples compares natural and unnatural directions:

```
\decalinevt{9B==\addbscolor{\red}{\bluex{H}};%
{10}B==\addbscolor{\red}{\bluex{H}}}
\decalinevt[%
{e{\RightAtomBond[\red]<s>{0B==\bluex{H}}}
{j{\RightAtomBond[\red]<s>{8B==\bluex{H}}}]}
```



11.5.3 Other Unnatural Directions of Substitution Bonds

The **\Ethylene** command draws a planar ethylene derivative, as shown in the left-hand structure of the following examples. The substitution bonds in this structure are regarded as being in natural directions.



On the other hand, an electrophilic addition to an olefinic double bond requires a stereochemical representation, as shown usually in the middle structure or more extremely in the right-hand structure. These three structural formulas are drawn by the following codes:

```
\Ethylene{}{1==H;2==\ChemForm{H_3C};3==\ChemForm{CH_3};4==H} \hskip50pt
\Ethylene{%
1s==\LeftAtomBond[\blue]{5B==\redx{\ChemForm{H_3C}}};%
1s==\LeftAtomBond(0,80)[\blue]{3A=={\red H};1==C;%
2s==\RightAtomBond(80,0)[\blue]{5B==\redx{\ChemForm{CH_3}}};
2s==\RightAtomBond(80,80)[\blue]{3A==\redx{H}};2==C;
1==\put(100,-200){\bluex{H$^{+}$}};
1=={\red \electronrshiftarrow(140,0)(100,-150)}}{ \hskip50pt
\Ethylene{%
1=={\red \WedgeAsSubst(0,20)(-5,-1){171}};%
1 = \left(-200, -50\right) \left(\left(-200, -50\right)\right) \right);
1=={\red \PutDashedBond(0,50)(-171,90){\thickLineWidth}};%
1==\put(-200,60){\llap{\bluex{H}}};%
1==C;%
2=={\red \WedgeAsSubst(80,20)(5,-1){171}};%
2==\put(280,-50){\rlap{\bluex{\ChemForm{CH_3}}};%
2=={\red \PutDashedBond(80,50)(251,90){\thickLineWidth}};%
2==\put(280,60){\rlap{\bluex{H}}};%
2==C;%
1==\put(100,-200){\bluex{H$^{+}$};%
1 = \operatorname{k}(-150)
```

The middle structure is drawn by using \RightAtomBond and \LeftAtomBond defined in this section. On the other hand, the right-hand structure is drawn by using \WedgeAsSubst and \PutDashedBond according to Section 11.4.

The position number 1 of the designation $1=\left(-200,-50\right)$... in the SUBSLIST of the Ethylene command (for drawing the right-hand formula) indicates that the origin is located at the lowerleft corner of C₁ (the left carbon) of ethylene and that the command $\mu u places$ an object (e.g., { $llap{blue \ChemForm{H_3C}})$ just as the origin of the object is superposed on the origin decided by the C₁ of ethylene. Thereby, the object appears at the position separated from the origin (0,0) by the xy-coordinate (-200,-50).

Chapter 12

Coloring Skeletal Bonds and Double Bonds

12.1 Coloring Skeletal Bonds

12.1.1 Systematic Method for Coloring Skeletal Bonds

Low-level commands (\genCOM) such as \sixheterov have an optional argument (SKBONDLIST) (cf. Section 10.1 of the on-line manual of $\hat{X}^{1}MT_{E}X$ version 2.00 (xymtx200.pdf)) as follows:

\genCOM(SKBONDLIST)[BONDLIST]{ATOMLIST}{SUBSLIST}[OMIT]

The optional argument (SKBONDLIST) is originally designed to change a skeletal bond into a bold bond (β bond) or a dashed bond (α bond), e.g.,

```
\sixheterov({eB}){6==0;%
5s==\pentamethylenei{}{1W==H0;1D==0;2B==\null;3B==OH;4B==\null;5==(yl)}%
}(1D==0;2A==\null;4A==\null}
```



where the designation ({eB}) indicates that the skeletal bond e of a six-membered ring is changed into a bold bond (β bond).

This mechanism of changing a skeletal bond is extended to be suitable to coloring the skeletal bond. According to this guideline, the **bondcolor** package of the $\hat{X}^{1}MT_{E}X$ version 5.00 has been developed to support the function of coloring skeletal bonds.

The command **\addskbcolor** is defined to specify the color of a skeletal bond, where it has a format represented by

\addskbcolor{COMMAND-TYPE}{COLOR}

where the first argument COMMAND-TYPE is used to specify the vertical or horizontal type of a command. The vacant argument is the same thing as v for corresponding to commands of vertical types (suffix v) or of inverse vertical type (suffix vi). The second argument COLOR is a command for specifying color (e.g.,

\red or \color{red}). The \addskbcolor command is designated in the (SKBONDLIST) (skeletal-bond list) of each command for drawing a skeleton.

```
For example, the codes:
```

```
\sixheterov({a{\addskbcolor{}{\red}}){}}\}\hskip-20pt
\sixheterov({b{\addskbcolor{}{\red}}){}}\}\hskip-20pt
\sixheterov({c{\addskbcolor{}{\blue}}){}{\hskip-20pt
\sixheterov({d{\addskbcolor{}{\blue}}){}{\hskip-20pt
\sixheterov({e{\addskbcolor{}{\green}}){}{}\hskip-20pt
\sixheterov({f{\addskbcolor{}{\green}}){}{} \par
\sixheterovi({a{\addskbcolor{}{\green}}){}{} \par
\sixheterovi({a{\addskbcolor{}{\red}}){}{} \hskip-20pt
\sixheterovi({b{\addskbcolor{}{\red}}){}}{} \hskip-20pt
\sixheterovi({c{\addskbcolor{}{\red}}){}}{} \hskip-20pt
\sixheterovi({c{\addskbcolor{}{\blue}}){}}{} \hskip-20pt
\sixheterovi({c{\addskbcolor{}{\blue}}){}}{} \hskip-20pt
\sixheterovi({c{\addskbcolor{}{\blue}}){}}{} \hskip-20pt
\sixheterovi({e{\addskbcolor{}{\blue}}){}}{} \hskip-20pt
\sixheterovi({e{\addskbcolor{}{\blue}}){}}{} \hskip-20pt
\sixheterovi({e{\addskbcolor{}{\blue}}){}}{} \hskip-20pt
\sixheterovi({e{\addskbcolor{}{\blue}}){}}{} \hskip-20pt
\sixheterovi({e{\addskbcolor{}{\blue}}}){}}{} \hskip-20pt
\sixheterovi({e{\addskbcolor{}{\blue}}}){}}{} \hskip-20pt
\sixheterovi({e{\addskbcolor{}{\blue}}}){} \}{} \}
```

show the use of the **\addskbcolor** command in **\sixheterov** (vertical type) and **\sixheterovi** (inverse vertical type), which generate the following structures:



The use of the **\addskbcolor** command in **\sixheteroh** (horizontal type) and **\sixheterohi** (inverse horizontal type) is illustrated by the following codes:

```
\sixheteroh({a{\addskbcolor{h}{\red}}){}{\hskip-20pt
\sixheteroh({b{\addskbcolor{h}{\red}}){}{\hskip-20pt
\sixheteroh({c{\addskbcolor{h}{\blue}}){}{\hskip-20pt
\sixheteroh({d{\addskbcolor{h}{\blue}}){}{\hskip-20pt
\sixheteroh({e{\addskbcolor{h}{\green}}){}{} \par
\sixheteroh({f{\addskbcolor{h}{\green}}){}{} \par
\sixheterohi({a{\addskbcolor{h}{\red}}){}{} \par
\sixheterohi({b{\addskbcolor{h}{\red}}){}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\red}}){}{} \par
\sixheterohi({c{\addskbcolor{h}{\red}}}){}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}){}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){}}}{} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}}){} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}})}){} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}})}){} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}})}){} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}})}){} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}}})}){} \hskip-20pt
\sixheterohi({c{\addskbcolor{h}{\blue}})}){} \hskip-20pt
```

where the first argument {h} of \addskbcolor corresponds to horizontal type or horizontal inverse type. These codes generate structural formulas with colored skeletal bonds:



The presence of a ring atom results in bond shortening, where such a shortened bond can be also colored by the use of the **\addskbcolor** command. Thus the codes:

generate structural formulas with colored shortened bonds:



The use of the **\addskbcolor** command in **\fiveheterov** (vertical type) and **\fiveheterovi** (inverse vertical type) as shown in the codes:

```
\fiveheterov({a{\addskbcolor{}{\red}}){}{}
\fiveheterov({b{\addskbcolor{}{\red}}){}{}
\fiveheterov({c{\addskbcolor{}{\blue}}){}{}
\fiveheterov({d{\addskbcolor{}{\blue}}){}{}
\fiveheterov({e{\addskbcolor{}{\red}}){}{}
\fiveheterovi({a{\addskbcolor{}{\red}}){}{}
\fiveheterovi({b{\addskbcolor{}{\red}}){}{}
\fiveheterovi({c{\addskbcolor{}{\red}}){}{}
\fiveheterovi({c{\addskbcolor{}{\red}}){}{}
\fiveheterovi({c{\addskbcolor{}{\blue}}){}{}
\fiveheterovi({c{\addskbcolor{}{\blue}}){}{}}
\fiveheterovi({e{\addskbcolor{}{\blue}}){}{}}}
```

generates the following structures:





Other commands which support a skeletal bond list can use the **\addskbcolor** command. The following structure is drawn by using **\addskbcolor** commands in **\fourhetero** and **\fivefusevi**.

```
\begin{XyMcompd}(2100,600)(-800,100){}{}
\fourhetero({b{\addskbcolor{}{\red}})%
[{b\fivefusevi({a{\addskbcolor{}{\red}})%
{1==S;4==\null}{2Sa==CH$_{3}$;2Sb==CH$_{3}$;%
3A==\addbscolor{\red}{\bluex{COOH}}{d}]%
{2==N}{1D==0;3FA==H;4GA==H;%
4Su==\addbscolor{\red}{\blackx{%
\lyl(4=={\blue OCH$_{2}$CONH}){4==\bzdrh{4==(y1)}}}}
\end{XyMcompd}
```



The following structure is drawn by using \addskbcolor commands in \decaheterov.

```
\decaheterov({b{\addskbcolor{v}{\red}}}%
{g{\addskbcolor{v}{\red}}){2==NH}%
{9A==\addbscolor{\red}{\bluex{Me}};{10}B==\addbscolor{\red}{\bluex{Me}}}
```



The line width of a skeletal bond drawn by an \addskbcolor command can be changed by the first argument {\red\thicklines}. Such a line width can be reset by redefining the line width stored by \thickLineWidth, e.g., \def\thickLineWidth{2.5pt}.

```
\decaheterov({b{\addskbcolor{v}{\red\def\thickLineWidth{2.5pt}\thicklines}}}%
{g{\addskbcolor{v}{\red\thicklines}}){2==NH}%
{9A==\addbscolor{\red}{\bluex{Me}};{10}B==\addbscolor{\red}{\bluex{Me}}}
```



12.1.2 Dirty Techniques for Coloring Skeletal Bonds

The technique base on \addskbcolor cannot be applied to commands which have no skeletal bond list (SKBONDLIST), e.g., \steroid. Because an original object can be erased by overwriting a white object in the PDF mode (also in the PostScript mode), the macro \addPDFLine (or \addPSLine) is defined tentatively to draw a colored skeletal bond.

The \addPSLine command for the PostScript mode is defined as follows:

```
\makeatletter
\def\white{\color{white}}
\def\addPSLine{%
\@ifnextchar[{\@ddPSLine}{\@ddPSLine[\thinLineWidth]}}
\def\@ddPSLine[#1](#2,#3)(#4,#5)#6{%
\psline[unit=\unitlength,linewidth=#1,linecolor=white](#2,#3)(#4,#5)%
\put(0,0){#6\PutBondLine(#2,#3)(#4,#5){#1}}%
\makeatother
```

The optional argument can be applied to change a bond width and the last argument is used to designate a color.

On the other hand, the **\addPDFLine** command for the PDF mode is defined as follows:

```
\makeatletter
\def\white{\color{white}}
\def\addPDFLine{%
\@ifnextchar[{\@ddPDFLine}{\@ddPDFLine[\thinLineWidth]}}
\def\@ddPDFLine[#1](#2,#3)(#4,#5)#6{%
\put(0,0){\white\PutBondLine(#2,#3)(#4,#5){1.6pt}}%
\put(0,0){#6\PutBondLine(#2,#3)(#4,#5){#1}}%
\makeatother
```

The optional argument can be applied to change a bond width and the last argument is used to designate a color.

Because this document is typeset in the PDF mode, the macro \addPDFLine is written in a bond list, as shown in the following example:

```
\steroid[{b{\addPDFLine(0,0)(0,200){\red}}}%
{i{\addPDFLine[1.6pt](0,0)(171,103){\red}}}%
{o{\addPDFLine[1.6pt](0,0)(0,-200){\blue}}]}}
```



It should be noted that the macro \addPDFLine is so local as defined in the PDF mode, because the inner macro \PutBondLine depends upon the selected mode. The macro \addPSLine for the PostScript mode can be used by replacing \addPDFLine by \addPSLine.

The command **\replaceSKbond** is defined in the **bondcolor** package in order to color skeletal bonds in the PDF mode as well as in the PostScript mode, where a line to be colored is designated by specifying its starting point, slope, and length (as the x-axis projection):

\replaceSKbond[BONDWIDTH](Starting Point)(SLOPE){LENGTH}{COLOR}

The last argument COLOR of the \replaceSKbond command is set for a bond color. The first optional argument [BONDWIDTH] specifies the bond width to be drawn, as found in the following example:

\steroid[{b{\replaceSKbond(0,0)(0,1){200}{\red}}}%
{i{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{o{\replaceSKbond[1.6pt](0,0)(0,-1){200}{\blue}}]}}



The command **\replaceSKbond** is effective to color skeletal bonds of polymethylenes, as found in the following list of examples:

```
\tetramethylene[{a{\replaceSKbond(0,0)(5,3){171}{\red}}]{}{
\tetramethylenei[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}]{}{
\pentamethylenei[{c{\replaceSKbond[0,0)(5,3){171}{\red}}]{}{
\pentamethylenei[{d{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}]{}{
\pentamethylene[{c{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}]{}
{e{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}]{}
{e{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\red}}]{}
{hexamethylenei[{c{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\red}}]{}
{e{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\red}}]{}
{e{\replaceSKbond[1.6pt](0,0)(5,3){171}{\blue}}]{}{}
{e{\replaceSKbond[1.6pt](0,0)(5,3){171}{\blue}}]{}}
{heptamethylenei[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\blue}}]{}}
{heptamethylenei[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\blue}}]{}}}
```

```
{e{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\blue}}]{} \par
\cctamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\red}}}%
{f{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\blue}}]{}
{cctamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\red}}}%
{f{\replaceSKbond[1.6pt](0,0)(5,3){171}{\blue}}]{}
{g{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\blue}}}}%
{g{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\blue}}}{}
{cctamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\red}}}%
{f{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\blue}}}}]{}
{cctamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\red}}}%
{h{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\blue}}}]{}
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,-3){171}{\red}}}%
{h{\replaceSKbond[1.6pt](0,0)(5,3){171}{\blue}}}]{}
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{cccamethylene[{b{\replaceSKbond[1.6pt](0,0){171}{\red}}}%
{cccameth
```





The colored structure shown in Page 102 can be alternatively drawn by using the replaceSKbond command. Note that the pair of bond specifiers 'b/B' or 'g/G' designates the alternative terminals of a bond to be selected as starting points. With respect to the pair 'b/B' or 'g/G', check the slope (0,1) or (0,-1) specified as the argument of replaceSKbond.

```
\decaheterov[{b{\replaceSKbond[1.6pt](0,-50)(0,-1){150}{\red}}}%
{g{\replaceSKbond[1.6pt](0,0)(0,1){200}{\red}}]%
{2==NH}%
{9A==\addbscolor{\red}{\bluex{Me}};{10}B==\addbscolor{\red}{\bluex{Me}}}
\decaheterov[{B{\replaceSKbond[1.6pt](0,0)(0,1){150}{\red}}%
{G{\replaceSKbond[1.6pt](0,0)(0,-1){200}{\red}}]%
{2==NH}%
{9A==\addbscolor{\red}{\bluex{Me}};{10}B==\addbscolor{\red}{\bluex{Me}}}
```



12.2 Coloring Double Bonds

12.2.1 A Systematic Way

The bondcolor package of the $X^{1}MT_{E}X$ version 5.00 supports the function of coloring double bonds.¹ The command **\adddbcolor** is defined to specify the color of a double bond, where it has a format represented by

\addbcolor{COMMAND-TYPE}{COLOR}

where the first argument COMMAND-TYPE is used to specify the vertical or horizontal type of a command, such as v for commands of vertical types (suffix v), vi for commands of inverse vertical types (suffix vi), and so on. The second argument COLOR is a command for specifying color (e.g., \red or \color{red}). The \addbcolor command is designated in the BONDLIST (bond list) of each command for drawing a double bond. For example, the codes:

```
\sixheterov[{a{\addbcolor{v}{\red}}]}}}}}}}}}Sixheterov[{b{\addbcolor{v}{\red}}]}}}}}bskip-20pt
\sixheterov[{c{\addbcolor{v}{\blue}}]}{}{\hskip-20pt
\sixheterov[{d{\addbcolor{v}{\blue}}]}}}}}bskip-20pt
\sixheterov[{e{\addbcolor{v}{\green}}]}}}}bran
\sixheterov[{a{\addbcolor{v}{\green}}]}}}
\sixheterov[{a{\addbcolor{v}{\green}}]}}
\sixheterov[{b{\addbcolor{v}}{\red}}]}
\sixheterov[{b{\addbcolor{v}}{\sixheterov}}]
\sixheterov[{b{\addbcolor{v}}{\sixheterov}}]
\sixheterov[{b{\addbcolor{v}}{\sixheterov}}]
\sixheterov[{b{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}}]
\sixheterovi[{c{\addbcolor{v}}{\sixheterov}]
\sixheterovi[{c{\addbcolor{v}
```

generate the following structures with colored double bonds:



¹According to the terminology of the $\hat{X}^{1}MTEX$ system, one bond of a double bond is regarded as a skeletal bond, while the other bond is regarded as a double bond to be added by an optional designation in BONDLIST.

Double bonds in polymethylene chains drawn by the macros of the methylen package can be colored by using the \addbcolor command. For commands of normal type (e.g., \decamethylene), the first argument of \addbcolor is set to be vacant. Lowercase bond specifiers a to i are used in the following examples:

```
\decamethylene[%
{a{\addbcolor{}{\red}}}%
{c{\addbcolor{}{\red}}}%
{e{\addbcolor{}{\red}}}%
{g{\addbcolor{}{\red}}}%
{i{\addbcolor{}{\red}}}%
{i{\addbcolor{}{\red}}}%
{b{\addbcolor{}{\red}}}%
{d{\addbcolor{}{\red}}}%
{f{\addbcolor{}{\red}}}%
{h{\addbcolor{}{\red}}}%
]{}{
```





Uppercase bond specifiers A to I are used to draw double bonds at the opposite sides in comparison with bond specifiers a to i.

```
\decamethylene[%
{A{\addbcolor{}{\red}}%
{C{\addbcolor{}{\red}}%
{E{\addbcolor{}{\red}}%
{G{\addbcolor{}{\red}}%
{I{\addbcolor{}{\red}}%
]{}{}
\decamethylene[%
{B{\addbcolor{}{\red}}%
{D{\addbcolor{}{\red}}%
{F{\addbcolor{}{\red}}%
{H{\addbcolor{}{\red}}%
]{}{}
```



For commands of inverse type (e.g., \decamethylenei), on the other hand, the first argument of \addbcolor is set to be i. Lowercase (a to i) and uppercase bond specifiers (A to I) are tested as follows by using the \decamethylenei command.

```
\decamethylenei[%
  a{\addbcolor{i}}{\c}{\addbcolor{i}}\
   \{e^{\dbcolor{i}}\red} \\ g^{\dbcolor{i}}\red} \\ \
  {i{\addbcolor{i}{\red}}}%
 ]{}{}
  \decamethylenei[%
  {b{\addbcolor{i}{\red}}}{d{\addbcolor{i}{\red}}}%
   {f{\addbcolor{i}}{h}}{h}{\addbcolor{i}}{\begin{tighted}{linewidth}} \label{eq:linewidth} \label{eq:linewidth} \label{eq:linewidth} \end{tighted} \label{eq:linewidth} \end{tighted} \
 ]{}{} \par
  \decamethylenei[%
   A{\addbcolor{i}}\C{\addbcolor{i}}\
  {E{\addbcolor{i}{\red}}}{G{\addbcolor{i}{\red}}}%
  {I{\addbcolor{i}{\red}}}%
]{}{}
  \decamethylenei[%
   B^{\Delta ddbcolor{i}}^{D}^{Addbcolor{i}}^{M} \
   \{F\{\adddbcolor{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\\ensuremath{i}\
]{}{}
```



The following examples show the comparison between a non-color version and a colored version of the same structure.

```
\steroid[bio]{}
\steroid[{b{\addbcolor{v}{\red}}}%
{i{\addbcolor{v}{\red
\dashhasheddash\def\thickLineWidth{2.5pt}%
\thicklines}}}%
{o{\addbcolor{v}{\blue}}]{}
```


12.3 Coloring Both Skeletal and Double Bonds

The techniques for coloring skeletal bonds (Section 12.1) can be used simultaneously with those for coloring double bonds (Section 12.2). Double bonds in bombykol (an insect pheromone of a silkworm moth) are colored in three different versions as follows. The last structure is a non-color version.

```
\begin{XyMcompd}(2600,600)(-100,-100){}{}
\decamethylene[%
a{\blue \label{red}}\
c{\addbcolor}{\red}}
{a{\replaceSKbond(0,0)(5,3){171}{\red}}}%
{c{\replaceSKbond(0,0)(5,3){171}{\red}}}%
]{%
{10}s==\tetramethylenei{}{1==(yl);4W==OH};%
1s==\sixheterov{}{2==(y1)}[aef]%
}{1W==H;2==H;3==H;4==H}
\end{XyMcompd}\par
\begin{XyMcompd}(2600,600)(-100,-100){}{}
\decamethylene[ac%
{a{\replaceSKbond(0,0)(5,3){171}{\red}}}%
c{\replaceSKbond(0,0)(5,3){171}{\red}}\
]{%
{10}s==\tetramethylenei{}{1==(yl);4W==OH};%
1s==\sixheterov{}{2==(y1)}[aef]%
}{1W==H;2==H;3==H;4==H}
\end{XyMcompd}\par
\begin{XyMcompd}(2600,600)(-100,-100){}{}
\decamethylene[%
a{\blue} {\close} 
c{\addbcolor}{\red}}
]{%
{10}s=\tetramethylenei{}{1==(y1);4W==OH};%
1s==\sixheterov{}{2==(y1)}[aef]%
}{1W==H;2==H;3==H;4==H}
\end{XyMcompd}\par
\begin{XyMcompd}(2600,600)(-100,-100){}{}
\decamethylene[ac]{%
{10}s==\tetramethylenei{}{1==(yl);4W==OH};%
1s==\sixheterov{}{2==(y1)}[aef]%
}{1W==H;2==H;3==H;4==H}
\end{XyMcompd}
```





Colored and non-colored structures of β -Carotene are drawn by the following codes:

```
%color version %
{\def\thinLineWidth{0.8pt}
\begin{XyMcompd}(3850,800)(-100,-150){}{}
\decamethylene[%bdfh
{b{\addbcolor{}{\red}}}%
{d{\addbcolor{}{\red}}}%
{f{\addbcolor{}{\red}}}%
h{\addbcolor}{\red}}
{b{\replaceSKbond[0.8pt](0,0)(5,-3){171}{\red}}}%
{d{\replaceSKbond[0.8pt](0,0)(5,-3){171}{\red}}}%
{f{\replaceSKbond[0.8pt](0,0)(5,-3){171}{\red}}}%
{h{\replaceSKbond[0.8pt](0,0)(5,-3){171}{\red}}}%
٦
{1s==\sixheterov(%
a{\blue v}{\vert}
)[%a
a{\dbcolor{v}}\
]{}{2==(y1);1==\null;3Sa==\null;3Sb==\null};
{10}s==\decamethylenei[%acegi
a{\dbcolor{i}}\
{c{\addbcolor{i}{\red}}}%
{e{\addbcolor{i}{\red}}}%
{g{\addbcolor{i}{\red}}}%
{i{\addbcolor{i}{\red}}}%
{a{\replaceSKbond[0.8pt](0,0)(5,-3){171}{\red}}}%
{c{\replaceSKbond[0.8pt](0,0)(5,-3){171}{\red}}}%
\{e\{\replaceSKbond[0.8pt](0,0)(5,-3)\{171\}\{\red\}\}\}
{g{\replaceSKbond[0.8pt](0,0)(5,-3){171}{\red}}}%
{i{\replaceSKbond[0.8pt](0,0)(5,-3){171}{\red}}}%
]{{10}s==\dimethylene{%
2s==\sixheterov(%
d{\lambda dskbcolor{v}{\lambda ed}}
) [%d
d{\addbcolor{v}}\
]{}{5==(y1);4==\null;6Sa==\null;6Sb==\null}
}{1==(y1)}}{1==(y1);4==\null;8==\null}}{4==\null;8==\null}
\end{XyMcompd}\par
```



The top declaration \def\thinLineWidth{0.8pt} results in printing bold bonds. In accord with this declaration, the command \replaceSKbond should take an optional argument [0.8pt].

12.4 Background Colors

Commands for coloring skeletal and double bonds are summarized as follows:

Skeletal bonds	\replaceSKbond	in a bond list $[\ldots]$
	\addskbcolor	in a skeletal bond list $()$
Double bonds	\adddbcolor	in a bond list []

The three techniques can be combined freely, as well as in combination with \addbscolor for coloring substitution bonds and substituents. The following examples illustrate a combined use of \addskbcolor, \addbscolor and \addbscolor in the arguments of the \decaheterov command.

```
\begin{tabular}{ccc}
Background: blue & Background: blue & Background: green \\
\bluex{%
\decaheterov[bg]{}{9B==H;{10}B==H;1D==0;4D==0}}
&
\bluex{%
\bluex{%
\decaheterov(%SKBONDLIST
{f{\addskbcolor{v}}\red}}%
{g{\addskbcolor{v}{\red}}%
{h{\addskbcolor{v}{\red}}%
)[b%BONDLIST
```

```
\{g\{\dbcolor\{v\}\{\red\}\}\}\
]{}{%SUBSTLIST
9B==\bluex{H};{10}B==\bluex{H};%\blue is necessary
1D = \left( 2 \right); 4D = \left( 2 \right)
&
greenx{%
\decaheterov(%SKBONDLIST
{e{\addskbcolor{v}{\blue}}}%
f{\addskbcolor{v}{\red}}%
\{g\{\ dskbcolor\{v\}\{\ ed\}\}\}
h{\addskbcolor{v}{\red}}
{i{\addskbcolor{v}{\blue}}}%
k{\addskbcolor{v}{\blue}}%
)[b%BONDLIST
\{g\{\dbcolor\{v\}\{\red\}\}\}\
]{}{%SUBSTLIST
\label{eq:loss} $$ 1D=\dbscolor{blue}{\redx{0}}; $$ 4D=\addbscolor{blue}{\redx{0}}; $$ addbscolor{blue}. $$ addb
9B==\addbscolor{\blue}{H};{10}B==\addbscolor{\blue}{H}} \\
\end{tabular}
```

The background color of a structural formula can be changed from a default black into any color, where explicit designations of bond colors due to the present techniques are maintained during the change of the background color.



Chapter 13

Coloring Chemical Schemes

The techniques for coloring bonds and substituents (Chapters 11 and 12) can be combined freely. This chapter is devoted to show that combined techniques are effective to draw chemical schemes which contain colored structural formulas.

13.1 Formation of Cyanohydrins

Formation of cyanohydrin is represented by using the ChemEqnarray environment (defined in the chemist package), as found in the following code:

```
\begin{ChemEqnarray}
X.X.
\tetrahedral{0==C;2==\ChemForm{CH_3};3D==\addbscolor{\red}{0};%
4 = \operatorname{ChemForm}{CH_3}
\qquad \raisebox{25pt}{ + \qquad Na{\red CN}} \qquad
\reactrarrow{25pt}{2cm}{\ChemForm{H_2S0_4}}{\strut} \quad
\tetrahedral{0==C;2==\ChemForm{CH_3};3==\addbscolor{\red}{OH};%
1==\ChemForm{CH_3};4==\addbscolor{\red}{CN}} \nonumber \\
&&
\reactrarrow{25pt}{2.5cm}{\ChemForm{H_20, H_2S0_4}}{\strut} \qquad
\tetrahedral{0==C;2==\ChemForm{CH_3};3==\addbscolor{\red}{OH};%
1==\ChemForm{CH_3};4==COOH} \qquad
\reactrarrow{25pt}{1cm}{\strut} \qquad
\tetrahedral{0==C;2D==\addbscolor{\red}{\ChemForm{H_3C}};%
1==\ChemForm{CH_3};4==COOH} \label{eq:cyanohydrin}
\end{ChemEqnarray}
```

By means of the ChemEqnarry environment (not by means of the eqnarray environment of $IAT_EX 2_{\mathcal{E}}$), the compound NaCN is printed out in upright format (cf. Section 7.5 and Subsection 9.2.1). The participating compounds are drawn by using the \tetrahedral command, where substituents and substitution bonds are colored by using the \addbscolor command. The reaction arrows are drawn by means of the \reactrarrow command, which is defined in the chemist (chmst-pdf or chmst-ps) package.



It should be noted that subscripts due to the \ChemForm command are printed to show the same depth, even if they coexist with superscripts. Compare the subscripts appearing in $H_2SO_4^{--}$ (due to \ChemForm{H_2 SO_4^{--}}) with those appearing in $H_2SO_4^{--}$ (due to H_2SO_4^{--}$)$). Thus, the subscripts 2 and 4 printed by \ChemForm appear at the positions of the same depth, while the counterparts printed by \$...\$ appear at the positions of different depths.

13.2 Grignard Reactions

A Grignard reagent derived from benzyl chloride reacts with actone so as to give a *tert*-alcohol, as shown in the following scheme due to the ChemEquation environment defined by the chemist package. The code:

```
\begin{ChemEquation}
\def\thinLineWidth{0.8pt}
\begin{XyMcompd}(600,300)(0,100){}{}
\tetrahedral{0==C;2==\ChemForm{CH_3};3D==\addbscolor{\blue}{0};%
4 = \operatorname{ChemForm} \{CH_3\}
\end{XyMcompd}
\quad \raisebox{0pt}{+} \quad
\begin{XyMcompd}(800,300)(350,250){}{}
\redx{\bzdrh{4==\ChemForm{CH_2{\blackx{MgCl}}}}
\end{XyMcompd}
\quad \reactrarrow{Opt}{1cm}{\strut} \quad
\begin{XyMcompd}(1100,500)(-500,50){}{}
\tetrahedral{0==C;1==\ChemForm{CH_3};3==\addbscolor{\blue}{OH};%
2==\redx{\lyl(4==\ChemForm{CH_2}){4==\bzdrh{4==(y1)}};%
4 = \ChemForm{CH_3}
\end{XyMcompd}
\end{ChemEquation}
```

generates the following equation containing structures with colored bonds and moieties:



The declaration $\det \tilde{0.8pt}$ just after the beginning of the ChemEquation environment indicates that the line width of each bond is changed from the standard width (0.4pt) to a thicker one (0.8pt).

13.3 Resonance Structures

A versatile convention in organic chemistry is based on resonance structures, which represent electron shift in an intuitive way. For example, a benzenediazonium ion is represented by a set of resonance structures shown below:



Equation 13.3 is drawn by the following code. Curved arrows for representing electron shifts in resonance structures are drawn by using such commands as electronrshiftarrow, which have been defined in the chemist (chmst-pdf or chmst-ps) package (cf. Subsection 7.6.1). Lone pairs on a nitrogen atom are drawn by using lonepairA, which has been defined in the lewisstruc package of the \hat{X}^2MTEX system (cf. Chapter 2.1 of the on-line manual for \hat{X}^2MTEX version 4.05, i.e., xymtx405A.pdf).

```
\begin{ChemEquation}
\left[%
```

```
\begin{tabular}{cccccc}
\begin{XyMcompd}(300,900)(230,200){}{}
{f{\redx{\electronrshiftarrow(100,30)(160,170)}}]%
{}{1==\tetrahedral{3==(y1);0==N\rlap{$^{+}$};%
0==\redx{\electronrshiftarrow[1](-30,130)(-30,240)};%
1T = \lonepairA[1]{N}
\end{XyMcompd}
&\raisebox{-20pt}{$\llongleftrightarrow$}&
\begin{XyMcompd}(300,900)(230,200){}{}
\sixheterov[b%
{d{\addbcolor{v}{\red}}}%
\{d\{\red\electronlshiftarrow(-160,220)(-70,90)\}\}
]%
{6s=={\blue$^{+}\!$}}{1D=\tetrahedral{3==(y1);0==N\rlap{$^{+}$};%
1D = \left( \frac{10 + 10}{10 + 10} \right)
\end{XyMcompd}
&\raisebox{-20pt}{$\llongleftrightarrow$}&
\begin{XyMcompd}(300,900)(230,200){}{}
\sixheterov[e%
{b{\addbcolor{v}{\red}}}%
b{\operatorname{electronrshiftarrow}(-40,-90)(-90,-240)}
]%
{4==\lower.4em\hbox{\blue\scriptsize$+$};4s==\null}%
{1D==\tetrahedral{3==(y1);0==N\rlap{$^{+}$};%
1D = \frac{124}{N} 
\end{XyMcompd}
&\raisebox{-20pt}{$\llongleftrightarrow$}&
\begin{XyMcompd}(300,900)(230,200){}{}
\sixheterov[ce]%
{2s==\bluex{$\!^{+}$}}{1D==\tetrahedral{3==(y1);0==N\rlap{$^{+}$};%
```

```
1D==\redx{\lonepairA[24]{N\rlap{$\,^{-}$}}}}
\end{XyMcompd}
\\
\end{tabular}
\right]
\label{eq:diazonium}
\end{ChemEquation}
```

On a similar line, pyridine is represented by a set of resonance structures shown below.



Equation 13.4 is drawn by the following code.

```
\begin{ChemEquation}
\left[ \right]
scalebox{0.8}{%}
\begin{tabular}{cccccccc}
\begin{XyMcompd}(300,500)(230,200){}{}
\sixheterov[%
b{\addbcolor{v}}\
b{\operatorname{electronrshiftarrow}(-40,-90)(-90,-240)}
d{\addbcolor{v}{\red}}
\{d\{\red\electronlshiftarrow(-160,200)(-70,100)\}\}
\{f\{\ ddb color\{v\}\}\}\
{f{\red\electronrshiftarrow[1](100,20)(260,20)}}%
]%
{4==\lonepairA[3]{N}}
\end{XyMcompd}
&$\llongleftrightarrow$&
\begin{XyMcompd}(300,500)(230,200){}{}
\sixheterov[ae%
c{\dbcolor{v}}\
{c{\red\electronlshiftarrow[2](-160,-50)(-80,0)}}
]%
{4==\lonepairA[3]{N}}
\end{XyMcompd}
&$\llongleftrightarrow$&
\begin{XyMcompd}(300,500)(230,200){}{}
\sixheterov[e%
a{\dbcolor{v}}\
{a{\red\electronrshiftarrow[1](70,-90)(160,-220)}}
]%
{4=\lonepairA[13]{N}\rlap{\rd $_{-}$};3s=={\blue},_{+}$}
}{}
\end{XyMcompd}
&$\llongleftrightarrow$&
\begin{XyMcompd}(300,500)(230,200){}{}
\sixheterov[b%
\{e\{\dbcolor\{v\}\{\red\}\}\}\
{e{\red\electronrshiftarrow[1](50,90)(100,240)}}
```

```
]%
{4==\lonepairA[13]{N}\rlap{\red $_{-}$};%
1==\raise1em\hbox{\blue\scriptsize$+$};1s==\null
}{}
\end{XyMcompd}
&$\llongleftrightarrow$&
\begin{XyMcompd}(300,500)(230,200){}{}
\sixheterov[bf%
1%
\{4=\lonepairA[13]{N}\rlap{\rd $_{-}$};5s=={\blue$_{+}}!
}{}
\end{XyMcompd}
\backslash \backslash
\end{tabular}
\right]
\label{eq:pyridine}
\end{ChemEquation}
```

13.4 Beckmann Rearrangement

The following scheme concerning a Beckmann rearrangement is a color version of the reaction scheme drawn in Section 4.5 of the on-line manual of $\hat{X}^{2}MT_{E}X$ versions 4.05 and 4.06 (xymtx405406B.pdf).



This scheme is drawn by the following code:

```
\begingroup
\def\tboxtitle{\bf Beckmann Rearrangement}
\begin{tboxscreen}
\changeunitlength{0.07pt}
\begin{ChemEqnarray*}
&&
\begin{XyMcompd}(1000,850)(-150,-150){}{
\Ethylenev{1==C;2==N}{3==0H;2==\bzdrv{6==(y1)};1==\bzdrv{2==(y1)}}
\end{XyMcompd}
\mskip6mu \reacteqarrow{0pt}{1cm}{\small H$_{3}$0$^{+}$}{\strut} \mskip6mu
\begin{XyMcompd}(1000,850)(-150,-150){}{
\Ethylenev{1==C;2==N;%
1=={\blue \putRoundArrow{(-85,-20)(-100,150)(-20,250)}};%
```

```
2=={\blue \putRoundArrow{(130,140)(150,350)(250,260)}}%
%1==\pscurve[unit=\unitlength,linewidth=0.4pt]{->}%
%(-85,-20)(-100,150)(-20,250);%
                                             %
%2==\pscurve[unit=\unitlength,linewidth=0.4pt]{->}%
%(130,140)(150,350)(250,280)%
                                             %
}{3==\addbscolor{\red}{\llap{$^{+}$}OH$_{2}$};%
2==\bzdrv{6==(y1)};1==\addbscolor{\red}{\bzdrv{2==(y1)}}}
\end{XyMcompd}
\mskip6mu \reacteqarrow{0pt}{0.8cm}{} \mskip6mu
\left\lgroup
\begin{tabular}{c}
\small Ph\sbond C$^{+}$\dbond N\sbond Ph \\[-8pt]
\reactduarrow{0pt}{20pt}{} \\
\small Ph\sbond C\tbond N$^{+}$\sbond Ph \\
\end{tabular}
\right\rgroup
+ H_{2}0
\mskip6mu \reacteqarrow{0pt}{0.8cm}{}}
\\ \noalign{\vskip20pt}
& &
\begin{XyMcompd}(1100,500)(-400,0){}{}
\dimethylenei[a]{1==C;2==N}{2W==\bzdrh{1==(y1)};1W==\bzdrh{4==(y1)};%
1==\Utrigonal{1==(y1);0==\upnobond{0}{+};2==H;3==\addbscolor{\red}{H};%
0=={\blue \putRoundArrow{(-80,140)(-40,300)(40,170)}
}}}
\end{XyMcompd}
\mskip6mu \reacteqarrow{0pt}{1.5cm}{proton}{shift} \mskip6mu
\begin{XyMcompd}(1100,500)(-400,0){}{}
\dimethylenei[%
{a{\replaceSKbond(25,-15)(5,-3){120}{\red}}}%
a{\dbcolor{i}}\
]{1==C;2==\upnobond{N}{+};%
1=={\blue \putRoundArrow[<-]{(-40,100)(-120,180)(-120,280)(-40,330)(60,320)}};%
1=={\blue \putRoundArrow{(60,50)(100,200)(150,80)}}%
%1==\pscurve[unit=\unitlength,linewidth=0.4pt]{<-}%
%(-40,100)(-120,180)(-120,280)(-40,330)(60,320);% %
%1==\pscurve[unit=\unitlength,linewidth=0.4pt]{->}%
%(60,50)(100,200)(150,80)%
                                             %
}%
{2==H;2W==\bzdrh{1==(y1)};1W==\bzdrh{4==(y1)};%
1==\addbscolor{\red}{\Utrigonal{0=={\black 0};2==H;1==(y1)}}
\end{XyMcompd}
\mskip6mu \reactrarrow{0pt}{1cm}{} \mskip6mu
\begin{XyMcompd}(1100,500)(-400,0){}{}
\label{eq:limit} $$ dimethylenei{2=}downnobond{N}{H}{2W=}bzdrh{1==(y1)};1W=bzdrh{4==(y1)};1D==0 $
\end{XyMcompd}
\end{ChemEqnarray*}
\end{tboxscreen}
```

\endgroup

Skeletal and double bonds participating in the Beckman reaction are colored by using \replaceSKbond (skeletal bonds) and \adddbcolor (double bonds). Although round arrows have been drawn by using \psline (defined in the pstricks package) in Section 4.5 of xymtx405406.pdf, they are now drawn by using the newly-defined command \putRoundArrow, which is compatible to the PDF mode as well as to the PS mode of the $\hat{X}^{1}MT_{E}X$ system. Because the old designations due to \psline are commented out, the comparison between \putRoundArrow and \psline shows their correspondence.¹ Compare these commands with another set of commands for drawing round arrows described Section 13.3.

13.5 Nucleophilic Substitutions

The chlorine atom of 1-cloro-2,4-dinitrobenzene is reactive to a nucleophilic reagent (Nu:), where its intermediate (a Meisenheimer complex) is delocalized by the two nitro groups, as shown in the following scheme:



This scheme (eq. 13.5) is drawn by the following code:

```
\begin{ChemEquation}\def\thinLineWidth{0.6pt}
\scalebox{0.7}{%
\begin{XyMcompd}(800,900)(50,0){}{}
\sixheterov({f{\addskbcolor{v}{\red}})[bd{f{\addbcolor{v}{\red}}]{%
1s=={\red \electronrshiftarrow(-200,100)(-20,20)};%
1s==\put(-200,100){\makebox(0,0)[rc]{\red Nu:~}};%
6s=={\red \electronlshiftarrow[1](20,-20)(120,20)}%
}{1==Cl;2==N0$_{2}$;4==N0$_{2}$}
\end{XyMcompd}}
\mathrel{\scalebox{0.7}{\reactrarrow{0pt}{1cm}}}}
\left[%
\scalebox{0.7}{%
\begin{tabular}
```

¹The mechanism of drawing curves by the pgf package (used in the PDF-compatible mode of the $\hat{X}^{1}MT_{E}X$ system is different from the corresponding mechanism by the pstricks package (used in the PostScript-compatible mode). Hence, the same argument of \putRoundArrow of the two modes results in different outputs.

```
\begin{XyMcompd}(700,900)(150,0){}{}
6s=={\red \electronrshiftarrow[1](-30,30)(-20,-100)};%
4s=={\red \electronrshiftarrow(-80,100)(5,25)}%
}{1SA==Cl;1SB==\addbscolor{\red}{Nu};2==N0$_{2}$;4==N0$_{2}$}
\end{XyMcompd}
& $\llongleftrightarrow$ &
\begin{XyMcompd}(700,1100)(150,-200){}{}
\sixheterov[be]{%
4 = \log \left( \frac{1}{\frac{1}{1}}\right) 
4s=={\red \electronrshiftarrow[1](35,50)(10,-80)}%
}{1SA==Cl;1SB==\addbscolor{\red}{Nu};2==N0$_{2}$;
4==\Dtrigonal{1==(y1);0=={\blue N\rlap{$^{+}$}};%
0=={\red \electronlshiftarrow(-200,-70)(-90,-20)};%
2==\lonepairA[123]{0\rlap{\red $\,\,^{-}$}};3D==\addbscolor{\red}{\lonepairA[34]{0}}}
\end{XyMcompd}
& $\llongleftrightarrow$ &
\begin{XyMcompd}(800,1000)(250,0){}{}
\sixheterov[ce]{%
2==\lower0.1em\hbox{\red \lonepairB[3]{\phantom{C}\rlap{_{-}}};%
2s=={\red \electronrshiftarrow(-35,0)(60,50)}%
}{1SA==Cl;1SB==\addbscolor{\red}{Nu};
4==NO$_{2}$;%
2==\Dtrigonal{3==(y1);0=={\blue N\rlap{$^{+}$}};%
0=={\red \electronrshiftarrow[1](70,120)(90,250)};%
2==\lonepairA[123]{0\rlap{\red $\,\,^{-}$}};%
1D=\addbscolor{\red}{\lonepairA[14]{0}}}
\end{XyMcompd}
\backslash \backslash
& & \reactduarrow{0pt}{1.5cm}{}} & & \reactduarrow{0pt}{1.5cm}{}} \\
X.X.
\begin{XyMcompd}(800,1100)(50,-200){}{}
\sixheterov[be]{%
}{1SA==Cl;1SB==\addbscolor{\red}{Nu};2==N0$_{2}$;
4D==\Dtrigonal{1==(y1);0=={\blue N\rlap{$^{+}$}};%
%0=={\red \electronlshiftarrow(-200,-70)(-90,-20)};%
2==\lonepairA[123]{0\rlap{\red $\,\,^{-}$}};%
3=\addbscolor{\red}{\lonepairA[134]{\lap{$^{-},,,$}0}}}
\end{XyMcompd}
$ $
\begin{XyMcompd}(800,1000)(250,0){}{}
\sixheterov[ce]{%
}{1SA==Cl;1SB==\addbscolor{\red}{Nu};
4==NO$_{2}$;%
2D==\Dtrigonal{3==(y1);0=={\blue N\rlap{$^{+}$}};%
2==\lonepairA[123]{0\rlap{\red $\,\,^{-}$}};%
1==\addbscolor{\red}{\lonepairA[124]{0\rlap{\,,,,{-}\}}}}
\end{XyMcompd}
\backslash \backslash
\left\{ tabular \right\}
```

\end{ChemEquation}

In this code, a set of round arrows **\electronrshiftarrow** and **\electronlshiftarrow** (Subsection 7.6.1) is used to indicate electron shifts on a similar line to Section 13.3. Compare these commands with another set of commands for drawing round arrows described Section 13.4. The scheme of delocalization surrounded by a pair of brackets is typeset by using the **tabular** environment, the size of which is reduced by using the **\scalebox** command. The total scheme (eq. 13.5) is typeset by using the **ChemEquation** environment, which is supported by the chemist (or chmst-pdf or chmst-ps) package.

13.6 A New Diels-Alder Bulding Block

The following scheme (eq. 13.6) is cited from a short article "Diels-Alder Bulding Block Debuts" published in *Chem. & Eng. News*, August 9 (2010) pages 30–31.



This scheme is drawn by writing the following code:

```
\def\markashR(#1,#2){%
\ifPSmode
{\put(#1,#2){\tetrahedral{4==(y1);%
2==\addbscolor{\psset{linecolor=white}\def\thinLineWidth{5pt}}{\null};%
2==R\kern-5pt}}}%
\else
{\put(#1,#2){\tetrahedral{4==(y1);%
2==\addbscolor{\white\def\thinLineWidth{5pt}}{\null};%
2==R\kern-5pt}\fi%
}
{\blue \def\thinLineWidth{0.8pt}
\begin{ChemEqnarray}
\begin{XyMcompd}(300,600)(100,150){}{}
{ blue \sixheterov[df{e}\red\markashR(100,100)}]{}{ abc]}
\end{XyMcompd}
\qquad + \qquad
\begin{XyMcompd}(300,400)(400,200){}{}\black
fourhetero[d]{}{3D==0}
\end{XyMcompd}
& \reactrarrow{Opt}{2cm}{\black Diels-Alder}{\strut}
```

&

```
\begin{XyMcompd}(900,600)(100,150){}{}
\sixheterov({b{\addskbcolor{v}{\black}})
[e{e{\red\markashR(100,100)}}%
{b{\black \fourfuse{}{3D==0;1FB==H;4GB==H}{d}}}}{}
\end{XyMcompd}
\nonumber \\
\noalign{\vskip10pt}
& \reactrarrow{Opt}{2cm}{\black Ring-expansion}{\strut}
k
\begin{XyMcompd}(900,600)(100,150){}{}
\sixheterov({b{\addskbcolor{v}{\black}})
[e{e{\red\markashR(100,100)}}%
{b{\black \fivefusevi{1=={\red X}}{2D==0;4FB==H;5GB==H}{D}}}%
]{}{}
\end{XyMcompd}
\label{eq:Diels-Alder} \\
\noalign{\black\centering {\red R} = various group;
{\red X} = \ChemForm{CH_2}, 0, or NH} \nonumber
\end{ChemEqnarray}
}
```

The newly-defined command \markashR is to put a Markash-type substituent on a specific bond of a structural formula. The white line due to the code,

2==\addbscolor{\white\def\thinLineWidth{5pt}}{\null},

erases a part of the single bond to be substituted in a Markash way. By declaring the code, \blue \def\thinLineWidth{0.8pt}, just before \begin{ChemEqnarray}, the whole domain derived by the ChemEqnarray environment is colored in blue and typeset with using bold-lined bonds.

13.7 Thiols and Cyanine Dyes

The following scheme (eq. 13.7) is cited from a short article "How Thiols Photoswitch Cyanine Dyes" published in *Chem. & Eng. News*, December 14 (2009) page 34. Photoswitching of cyanine dyes is a key to super-resolution fluorescence microscopy techniques used for biological imaging. Red laser light switches a dye from a fluorescent to a dark state, where a thiol attacks the polymethine unit to give an adduct. Inversely, ultraviolet illumination stimulates the release of the thiol unit from the adduct so as to turn the fluorescence back on.



Fluorescent



This scheme is drawn by writing the following code:

```
{\blue
\begin{ChemEqnarray*}
&
%\fbox{
\begin{XyMcompd}(3000,700)(-800,-50){}{}
\heptamethylenei[bdf]{%
1s==\nonaheterov[aegj{1+}]{1==N}{2==(y1);1==\dimethylene{}{2==(y1)};%
3Sa==\null;3Sb==\null;5==\ChemForm{HO_{3}S};%
7s = fiveheterov[{b}(sixfusev[ace]{}2=ChemForm{SO_{3}H}{e}]%
{1==N}{5==(y1);1==\dimethylenei{}{1==(y1)};4Sa==\null;4Sb==\null}}{
\end{XyMcompd}
%}
& \nonumber \\
\noalign{\centering\black Fluorecent}\\
\noalign{\vskip10pt}
&
\reactVEqarrow{0pt}{2cm}%
{\shortstack{Red light \\ \ChemForm{HOCH_2CH_2S^-}}}
{\shortstack{UV light \\ \ChemForm{- HOCH_2CH_2S^-}}}
& \nonumber\\
&
%\fbox{
\begin{XyMcompd}(3000,1000)(-800,-50){}{}
\heptamethylenei[adf]{%
1s==\nonaheterov[egj]{1==N}{2==(y1);1==\dimethylene{}{2==(y1)};%
3Sa==\null;3Sb==\null;5==\ChemForm{HO_{3}S};%
7s = fiveheterov[{b}(sixfusev[ace]{}2=ChemForm{SO_{3}H}{e}]%
{1==N}{5==(y1);1==\dimethylenei}{3==\null}{4Sb==\null}%
\{3=\ \{1=(y1); 3=0H\} \
\end{XyMcompd}
```

%}

```
& \label{eq:cyaninedye} \\
\noalign{\centering\black Dark}
\end{ChemEqnarray*}
}
```

The left indole (benzopyrrole) moiety is drawn by using a ready-made command \nonaheterov , while the right indole (benzopyrrole) moiety is drawn by a ring-fusion technique of the \fiveheterov and \sixfusev commands. It should be noted that the SCH₂CH₂OH unit of the adduct is drawn by using the \fiveheterov command, where an optional deletion list [cde] is used to designate skeletal bonds to be erased. Compare the following two structures drawn by using the \fiveheterov command.

 $\tiveheterov{1==S}{3==0H} \\fiveheterov{1==S}{3==0H}[cde]$



13.8 RNA Derived by a Counterintuitive Start

The following scheme (eq. 13.7) is cited from "Chemical Year in Review 2009" published in *Chem. & Eng. News*, December219 (2009) page 37 (entitled "DNA May Have Had a Counterintuitive Start").



This scheme is drawn by writing the following code:

```
{\blue \def\thinLineWidth{0.8pt}
\let\substfont=\sffamily
\begin{ChemEquation}
\begin{tabular}{c}
\begin{XyMcompd}(500,500)(300,200){}{}
\fiveheterov({e{\addskbcolor{v}{\red}})%
[c{e{\addbcolor{i}{\red}}]{2==0;4==N}{3==NH$_{2}$}
\end{XyMcompd}\\
\black 2-Aminooxazole
```

```
\end{tabular}
\qquad
\shortstack{$\llongrightarrow$ \\
\kern10pt$\llongrightarrow$ \\
\kern20pt$\llongrightarrow$}
\qquad
\begin{XyMcompd}(1000,1550)(100,-350){}{}
\fivesugarh{5==0;1s=={\red \WedgeAsSubst(0,0)(-3,-5){125}};%
4s==\WedgeAsSubst(0,0)(3,-5){125};%
3s==\PutBondLine(-15,0)(292,0){3.6pt}%
}{2Sa==0\put(-171,-190){\squareplanar{1==(y1);0==P;2==0$^{-}$;3D==0;1==;4==}};
%OH;
3Sa==0;4Sb==HOC\rlap{H$_{2}$};
1Sb==\sixheterov[ae]{2==N;4==N}{4==(y1);3D==0;1==NH$_{2}}%
}[abc]
\end{XyMcompd}
\label{eq:DNA}
\end{ChemEquation}
}
```

For drawing furanose derivatives having bold skeletal bonds, see Page 24. The cyclic phosphate group is drawn by using the \squareplanar command. The declaration \let\substfont=\sffamily changes the font for printing substituents.

13.9 Hydroxylations

The following scheme (eq. 13.8) is cited from a short article "Diiron Enzymes Spark Hydroxylations" published in *Chem. & Eng. News*, August 23 (2010) page 33, which is concerned with a step in the biosynthesis of the antibiotic chloramphenicol.



This scheme is drawn by writing the following code:

```
\begin{ChemEquation}\blue
\begin{XyMcompd}(500,1500)(300,-50){}{}
\sixheterov{4==S}{1==\bzdrv{4==(y1);1==NH$_{2}$};%
3D==0;2A==NH$_{2}$;4==\makebox(0,0){\kern6pt\fbox{CmIP}}[def]
\end{XyMcompd}
\reactrarrow{0pt}{2cm}{\black CmIA}{\strut}
```

```
\begin{XyMcompd}(600,1500)(200,-50){}{}
\sixheterov{4==S;1s==\LeftAtomBond[\red]{5A=={H0}}
}{1==\bzdrv{4==(y1);1==NH$_{2}$};%
3D==0;2A==NH$_{2}$;4==\makebox(0,0){\kern6pt\fbox{CmIP}}}[def]
\end{XyMcompd}
\qquad
\shortstack{$\llongrightarrow$ \\
\kern10pt$\llongrightarrow$}
\qquad
\begin{tabular}{c}
\begin{XyMcompd}(1000,1200)(200,250){}{}
\sixheterov
{1s==\LeftAtomBond[\red]{5A=={H0}};4==\lmoiety{H0}}
{1==\bzdrv{4==(y1);1==N0$_{2}$};3D==0;%
2A = \bluex{\ryl(4 = {\red \mbox{H}}){%}
5==\dimethylene{}{1==(y1);1D==0;2==C1;2W==C1}}}[def]
\end{XyMcompd}\\
\noalign{\vskip5pt}
{\black Chloramphenicol}
\end{tabular}
\label{eq:CmIA}
\end{ChemEquation}
```

For the command \LeftAtomBond, see Subsection 11.5.1.

13.10 Friedel-Crafts Acylation

As found in several textbooks, substitution bonds of aromatic rings are omitted so that substituents link to ring positions directly. Although the $\hat{X}^{2}MT_{E}X$ system does not support structural formulas of this type, a rather dirty technique is available by combining utilities supported by $\hat{X}^{2}MT_{E}X$. For example, the codes:

```
\sixheterov[ace]{%
1==\upnobond{}{Cl};1s==\null;%
2==\phantom{C}Cl;2s==\null;%
4==\downnobond{}{Cl};4s==\null%
}{}
\sixheterov[ace]{}{1==Cl;2==Cl;4==Cl}
```

generate the following structural formulas:



where the latter shows a structural formula due to a standard drawing of $\hat{X}^{1}MT_{E}X$. Note that the substituents of the former structure are designated in the ATOMLIST (the atom list) of \sixheterov, while those of the latter structure are designated in the the SUBSLIST (the substituent list) of \sixheterov.



Figure 13.1: Friedel-Crafts acylation of naphthalene

The designation 1s==\null in the former code redraws erased lines at a corner position, which have been brought about as a result of a designation placed just before (e.g., \upnobond{}{Cl}).

This technique is applicable to draw a reaction scheme of the Friedel-Crafts acylation of naphthalene, as shown in Figure 13.1. Coloring portions of formulas is realized by a usual technique of coloring.

Figure 13.1 has been drawn by the following code, where $\hat{X}^{1}MT_{E}X$ codes are written in a picture environment supported by $I^{2}T_{E}X 2_{\varepsilon}$. It should be noted that (=0.1pt) of $\hat{X}^{1}MT_{E}X$ codes is common to the the picture environment.

Chemical formulas such as $C_6H_5NO_2$ are typeset by such designations as $ChemForm{C_6_H_5_NO_2}$ in this code, where a space (_) is inserted to differentiate every units for the same of readability. The same chemical formulas can be obtain by writing $ChemForm{C_6H_5NO_2}, ChemForm{C_{6}H_{5}NO_{2}}, and C_{6}H_{5}NO_{2}$.

The two steps of Figure 13.1 can be numbered by using the ChemEqnarray environment, which is supported by the chemist package. To do this task, the command \FrCrArrows is first defined to draw an arrow unit which appears in Figure 13.1, as follows:

```
\def\FrCrArrows{%
%\fbox{
%\begin{picture}(1600,2600)(1000,0)%totally surrounding box
\begin{picture}(1600,0)(1000,1450)%surrounding box of height zero
%Lines and arrows%
\put(1000,1450){\line(1,0){800}}
\put(1800,450){\line(0,1){2000}}
\mu(1800, 2450) \{ \nu(1, 0) \{ 800 \} \}
\mu(1800, 450) \{\nu(1, 0), 800\}
%Reaction conditions%
\put(1000,1480){\makebox(800,0)[bc]{\red \ChemForm{CH_3 COC1}}}
\put(1000,1420){\makebox(800,0)[tc]{\blue \ChemForm{AlCl_3}}}
\put(1800,2480){\makebox(800,0)[bc]{\blue solvent:}}
\put(1800,2420){\makebox(800,0)[tc]{\blue \ChemForm{C_2 H_2 C1_4}}}
\put(1800,480){\makebox(800,0)[bc]{\blue solvent:}}
\put(1800,420){\makebox(800,0)[tc]{\blue \ChemForm{C_6 H_5 NO_2}}}
\end{picture}%}
}
```

where the designation **\begin{picture}(1600,0)(1000,1450)** is a key to decide a surrounding box; i.e., its height is equal to zero and its width is equal to the domain of the resulting arrow unit.

Then the following code based on the ${\tt ChemEqnarray}$ environment is written:

\begin{ChemEqnarray}
&&
\begin{tabular}{c}

```
\begin{XyMcompd}(600,450)(300,250){}{}
\decaheterov[acfhk]{1==\upnobond{}{\red\ChemForm{COCH_3}};1s==\null}{}
\end{XyMcompd} \\
1-Acetonaphthalene \\
\end{tabular}
\label{eq:FrCr1} \\
\noalign{\vskip35pt}%vertical space adjustment
\begin{tabular}{c}
\begin{XyMcompd}(600,450)(300,250){}{}
\decaheterov[acfhk]{}{
\end{XyMcompd} \\
Naphthalene \\
\end{tabular}
& \raisebox{8pt}{\FrCrArrows} & \nonumber \\
\noalign{\vskip35pt}%vertical space adjustment
&&
\begin{tabular}{c}
\begin{XyMcompd}(600,450)(300,250){}{}
\decaheterov[acfhk]{2=={\red \phantom{C}\ChemForm{COCH_3}};2s==\null}{}
\end{XyMcompd} \\
2-Acetonaphthalene \setminus
\end{tabular}
\label{eq:FrCr2}
\end{ChemEqnarray}
```

Thereby, we are able to obtain the following scheme, each step of which is numbered sequentially. The reaction-step numbers (eqs. 13.9 and 13.10) are cited by the \label and \ref mechanism of $\text{LAT}_{\text{EX}} 2_{\varepsilon}$.



13.11 Vitamin D_2

This is a color version of the reaction scheme drawn in Section 2.5 of the on-line manual of X^2MT_EX version 4.04 (xymtx404.pdf), where skeletal and double bonds participating photochemical and thermochemical reactions are colored by using \replaceSKbond (skeletal bonds) and \adddbcolor (double bonds).

Irradiation of ergosterol (and lumisterol) causes the opening of the B ring to produce previtamin D_2 having a conjugated triene, which is a precursor of vitamin D_2 (ergocalciferol), as shown in Figure 13.2. The double bonds of each intermediate are colored by the present technique of coloring double bonds.



Figure 13.2: Photochemistry of ergosterol and lumisterol

To draw the intermediates contained in Figure 13.2, macros for drawing them are defined by applying the systematic way of coloring double bonds, i.e., \lumisterol for drawing limisterol, \ergosterol for drawing ergosterol, \previtaminD for drawing previtamin D, and \vitaminDii for drawing vitamin D₂.

```
%lumisterol
\def\lumisterol{%
\begin{XyMcompd}(2050,1150)(0,250){}{}
\steroidChain[{Zc}%
{i{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{e{\addbcolor{v}{\red\def\thinLineWidth{1.6pt}}}%
{g{\addbcolor{v}{\red\def\thinLineWidth{1.6pt}}}%
]%
{3B==H0;9B==\addbscolor{\blue}{\greenx{H}};%
{10}A==\addbscolor{\blue}{\greenx{l}moiety{H$_{3}$C}};%
{13}B==\null;{14}A==H;{17}GA==H;%
{20}A==\null;{24}A==\null}
\end{XyMcompd}
```

}

```
%ergosterol
\def\ergosterol{%
\begin{XyMcompd}(2050,1150)(0,250){}{}
\steroidChain[{Zc}%
{i{\replaceSKbond[1.6pt](0,0)(5,3){171}{\red}}}%
{e{\addbcolor{v}{\red\def\thinLineWidth{1.6pt}}}%
{g{\addbcolor{v}{\red\def\thinLineWidth{1.6pt}}}%
]%
{3B==H0;9A==\addbscolor{\blue}{\greenx{H}};%
{10}B==\addbscolor{\blue}{\greenx{\lmoiety{H$_{3}$C}}};%
{13}B==\null;{14}A==H;{17}GA==H;%
{20}A==\null;{24}A==\null}
\end{XyMcompd}
}
%previtamin D
\def\previtaminD{%
\begin{XyMcompd}(2050,1150)(0,250){}{}
\sixheterov[%
{b{\sixfusev[%
{a{\addbcolor{v}{\red\def\thinLineWidth{1.6pt}}}}%
{c{\addbcolor{v}{\red\def\thinLineWidth{1.6pt}}}%
{e{\addbcolor{v}{\red\def\thinLineWidth{1.6pt}}}%
{a{\sixfusev[%
{b{\fivefusevi[%
{a{\sixfusev[a]{%
2s=\trimethylene{}{1==(yl);2A==\null;3==\null;3W==\null}
}{6A==\null}{D}[bc]}}
]{}{1GA==H}{D}}}
]{}{2FB==\null;3GA==H}{D}}}
]{}{}{E}[f]}}]
{}{2Sb==\addbscolor{\blue}{\null};5B==HO}
\end{XyMcompd}
}
%vitamin D$_{2}$ (ergocalciferol)
\def\vitaminDii{%
\begin{XyMcompd}(1650,1750)(0,250){}{}
\sixheterov[%
{a{\sixfusev[%
{c{\addbcolor{v}{\red\def\thinLineWidth{1.6pt}}}%
{e{\addbcolor{v}{\red\def\thinLineWidth{1.6pt}}}%
{f{\sixfusev[%b%
{b{\addbcolor{v}{\red\def\thinLineWidth{1.6pt}}}%
{a{\sixfusev[%
{b{\fivefusevi[
{a{\sixfusev[a]{%
2s=\trimethylene{}{1==(yl);2A==\null;3==\null;3W==\null}
}{6A==\null}{D}[bc]}}
]{}{1GA==H}{D}}
]{}{2FB==\null;3GA==H}{D}%
```

```
}}]{}{{C}[def]}}]{}{D}[ab]}}]{}{5A==HO}
\end{XyMcompd}
}
```

By using these macros, Figure 13.2 is drawn by the following code:

```
\begin{figure}[h]
\begin{center}
\begin{tabular}{ccc}
\scalebox{0.7}{\ergosterol} &
\reactrarrow{Opt}{1cm}{$h\nu$}{\strut} & \scalebox{0.7}{\previtaminD} \\
ergosterol & & previtamin D$_{2}$ \\[10pt]
& \reactnearrow{Opt}{1cm}{\raisebox{10pt}{\rlap{$h\nu$}}}{\strut} &
\reactdarrow{Opt}{1cm}{$\Delta$}{\strut} \\
\scalebox{0.7}{\lumisterol} && \scalebox{0.7}{\vitaminDii} \\
lumisterol & & vitamin D$_{2}$
\end{tabular}
\end{center}
\caption{Photochemistry of ergosterol and lumisterol}
\label{ff:PEL}
\end{figure}
```

13.12 Vitamin A_1

Vitamin A_1 is synthesized industrially by using a Wittig reaction, as shown in the following scheme (eq. 13.11), where a phosphorus yilde is an important intermediate.



This scheme (eq. 13.11) is obtained by the following code due to the ChemEqnarray environment of the chemist package (or the chemst-pdf or chemst-ps package):

```
{\def\thinLineWidth{0.6pt}
\begin{ChemEqnarray}
&&
\scalebox{0.7}{\VitaminAiIntA}\qquad
\reactrarrow{-5pt}{2cm}{\small \ChemForm{CH_30^{-}}}{\small \ChemForm{CH_30H}}\qquad
\scalebox{0.7}{\VitaminAiIntB} \nonumber \\
&&
```

```
\reactrarrow{0pt}{4cm}{\def\thinLineWidth{1pt}
\scalebox{0.5}{\VitaminAiIntC}\\[5pt]}{\small \ChemForm{-(C_6H_5)_3P0}}\qquad
\scalebox{0.7}{\VitaminAiEster} \nonumber \\
&&
\reactrarrow{25pt}{2cm}{\small \ChemForm{H0^{-}}}{\small \ChemForm{H_20}}\qquad
\shortstack{\scalebox{0.7}{\VitaminAi} \\ Vitamin A$_{1}$}
\end{ChemEqnarray}}
```

where respective intermediates (A–C and a Vitamin A_1 ester) and the final product (Vitamin A_1) are drawn after we define such commands as:

\VitaminAiIntA, \VitaminAiIntB, \VitaminAiIntC, \VitaminAiEster, and \VitaminAi.

The top declaration \def\thinLineWidth{0.6pt} is placed to make bonds thicker, so that the narrowing effects of size reduction (\scalebox) are cancelled out.

The intermediate A (the right intermediate in the first row of eq. 13.11 is drawn by the command \VitaminAiIntA, which is defined as follows:

```
\def\VitaminAiIntA{%
\begin{XyMcompd}(1800,650)(300,250){}{}blue
\sixheterov[b]{%
2s==\hexamethylene[bd]{}{1==(y1);4==\null;%
6W==\addbscolor{\black}{\ChemForm{P^{+}(C_6H_5)_3X^{-}}};%
6Sa==\addbscolor{\red}{H};6Sb==\addbscolor{\red}{H}}
{1Sa==\null;1Sb==\null;3==\null}
\end{XyMcompd}
}
```

where \addbscolor commands are used to color substituents and substitution bonds. Although the whole structure of \VitaminAiIntA is colored blue by declaring \blue globally, the local declarations of \red and \black overwrite respective colored objects over the blue backbone of the structure.

The intermediate B (the left intermediate in the first row of eq. 13.11) is a Wittig reagent (a phosphorus ylide), which is drawn by the macro \VitaminAiIntB. The macro is defined as follows:

```
\def\VitaminAiIntB{%
  \begin{XyMcompd}(1800,600)(300,250){}{}\blue
  \sixheterov[b]{%
  2s==\heptamethylene[bd
  {f{\adddbcolor{}{red}}}%
  {f{\replaceSKbond(0,0)(5,-3){130}{\red}}%
  ]{7==\redx{\ChemForm{P(C_6H_5)_3}}
  }{1==(y1);4==\null}
  {1Sa==\null;1Sb==\null;3==\null}
  \end{XyMcompd}
  }
```

where the **\adddbcolor** command is used to color a double bond and the **\replaceSKbond** command is used to color a skeletal bond.

The intermediate C above the reaction arrow in the second row of eq. eq. 13.11 is an aldehyde intermediate, which is drawn by the command \VitaminAiIntC. The macro for drawing the intermediate C is defined as follows:

```
\def\VitaminAiIntC{%
  \begin{XyMcompd}(1300,500)(250,0){}}\green
  \pentamethylenei[c%
```

```
{a{\addbcolor{i}{\red}}}%
{a{\replaceSKbond(171,-103)(-5,3){135}{\red}}}%
]{1==\redx{0}}%
{2==H;3==\ChemForm{CH_3};%
5W==\ryl(2==0){4==\tetrahedral{2==(y1);0==C;1D==0;4==\ChemForm{CH_3}}}
\end{XyMcompd}
}
```

where the **\addbcolor** command is used to color a double bond and the **\replaceSKbond** command is used to color a skeletal bond. Note that the local declaration of **\red** partially cancels the global coloring of the intermediate C by **\green**.

The right intermediate in the second row of eq. 13.11 is a Vitamin A_1 ester, which is drawn by the macro $\forall itaminAiEster$ defined as follows:

```
\def\VitaminAiEster{%
  \begin{XyMcompd}(2500,600)(300,250){}{\blue
  \sixheterov[b]{%
  2s==\heptamethylene[bd
  {f{\addbcolor{}{red}}%
  {f{\replaceSKbond(0,0)(5,-3){171}{\red}}%
  {f{\replaceSKbond(0,0)(5,-3){171}{\red}}}%
  {f{\red}{\text{replaceSKbond(0,0)(5,-3){171}{\red}}}%
  {f{\red}{\text{replaceSKbond(0,0)(5,-3){171}{\red}}}%
  {f{\red}{\text{replaceSKbond(0,0)(5,-3){171}{\red}}}%
  {f{\red}{\text{replaceSKbond(0,0)(5,-3){171}{\text{replaceSKbond(0,0)(5,-3){171}{\text{replaceSKbond(0,0)(5,-3){171}{\text{replaceSKbond(0,0)(5,-3){171}{\text{replaceSKbond(0,0)(5,-3){171}{\text{replaceSKbond(0,0)(5,-3){171}{\text{replaceSKbond(0,0)(5,-3){171}{\text{replaceSKbond(0,0)(5,-3){171}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{replaceSKbond(0,0){17}{\text{
```

where the left moiety (due to the Wittig reagent) and the right moiety (due to the aldehyde intermediate) are differentiated by colors (blue and green). Moreover, the resulting olefinic function is emphasized by drawing in red. The terminal acetyl group is differentiated from other portions by drawing in black, because it participates in a subsequent hydrolysis.

The final product, Vitamin A_1 , in the third row of eq. 13.11 is drawn by the following code:

```
\def\VitaminAi{%
    \begin{XyMcompd}(2000,600)(300,250){}{}\blue
    \sixheterov[b]{%
    2s==\heptamethylene[bd%
    {f{\addbcolor{}{\red}}}%
    {f{\addbcolor{}\red}}%
    {f{\replaceSKbond(0,0)(5,-3){171}{\red}}}]
    {7s==\greenx{\tetramethylene[b]{}%
    {1==(y1);2==\null;4W==0H}}%
    {1Sa==\null;1Sb==\null;3==\null}
    \end{XyMcompd}
}
```

where a newly introduced double bond is differentiated by red color.

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